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Appendix A

Details of Programs of Physico-Chemical Mathematical Models

- Program of Physico-Chemical Mathematical Model for Brimblecombe and Spedding (1974)'s Reaction Rate, Freiberg (1974)'s Reaction Rate and Ibusuki, Ohsawa and Takeuchi (1990)'s Reaction Rate in Ammonia-Rich Environment

```

#include <stdio.h>
#include <math.h>
#include <time.h>
#include <memory.h>
#define XMAX 107
#define YMAX 80
#define ZMAX 35
#define RAND_MAX 32767 /* (2^15)-1 */
main (ac,av)
int ac;
char **av;
{
int    c1[XMAX][YMAX][ZMAX],c2[XMAX][YMAX][ZMAX],
        c3[XMAX][YMAX][ZMAX],c4[XMAX][YMAX][ZMAX],
        c5[XMAX][YMAX][ZMAX],c6[XMAX][YMAX][ZMAX],
        i,j,k,i0,j0,k0,ii,jj,kk,n,N,T,no_of_row,no_of_col,no_of_height,
        opt,no_of_quanta,no_of_time,no_of_printing,max_of_printing,sum;
float  x,x0,y,y0,z,z0,t,w,l,h,dummx,dummy,dummz,y_coeff,z_coeff,rd,
        W,L,H,K,K_time_step,u,Hs,Vs,d,delta_h,x1,Ts,Ta,dummh,
        delta_h0,delta_h1,delta_h2,delta_h3,delta_h4,delta_h5;
FILE   *fp,*parafp;
char   fname[50],name[50];

    if ( ac != 2)
    {
        printf("No parameter file name\n");
        exit(1);
    }
    parafp = fopen(av[1],"r");
    fscanf (parafp,"%s",fname);
    fp = fopen(fname,"w");
    fprintf (fp, "The output file name is %s\n",fname);
    fscanf (parafp,"%s",name);
    fprintf (fp, "The condition of the reaction is %s\n",name);
    fscanf (parafp,"%f",&W);
    fprintf (fp,"The width of interested area (m) = %.2f\n",W);

```

```

fscanf (paraftp, "%f",&L);
fprintf (fp, "The length of interested area (m) = %.2f\n",L);
fscanf (paraftp, "%f",&H);
fprintf (fp, "The height of interested area (m) = %.2f\n",H);
fscanf (paraftp, "%f",&w);
fprintf (fp, "The width of each cell (m) = %.2f\n",w);
fscanf (paraftp, "%f",&l);
fprintf (fp, "The length of each cell (m) = %.2f\n",l);
fscanf (paraftp, "%f",&h);
fprintf (fp, "The height of each cell (m) = %.2f\n",h);
fscanf (paraftp, "%f",&x0);
fprintf (fp, "The location of each point source in the x-axis (m) = %.2f\n",x0);
fscanf (paraftp, "%f",&y0);
fprintf (fp, "The location of each point source in the y-axis (m) = %.2f\n",y0);
fscanf (paraftp, "%d",&N);
fprintf (fp, "The number of SO2 quanta in each point source = %d\n",N);
fscanf (paraftp, "%f",&u);
fprintf (fp, "The velocity of wind at the stack height (m/sec) = %.2f\n",u);
fscanf (paraftp, "%f",&Hs);
fprintf (fp, "The height of stack (m) = %.2f\n",Hs);
fscanf (paraftp, "%f",&Vs);
fprintf (fp, "The velocity of gas (m/s) = %.3f\n",Vs);
fscanf (paraftp, "%f",&d);
fprintf (fp, "The diameter of stack (m) = %.2f\n",d);
fscanf (paraftp, "%f",&x1);
fprintf (fp, "The downwind distance from source (m) = %.2f\n",x1);
fscanf (paraftp, "%f",&Ts);
fprintf (fp, "The temperature of stack (K) = %.2f\n",Ts);
fscanf (paraftp, "%f",&Ta);
fprintf (fp, "The atmosphere temperature (K) = %.3f\n",Ta);
fscanf (paraftp, "%d",&T);
fprintf (fp, "The number of time step = %d\n",T);
fscanf (paraftp, "%f",&t);
fprintf (fp, "The time step (sec) = %.2f\n",t);
fscanf (paraftp, "%f",&y_coeff);
fprintf (fp, "The coeff. of dispersion in the y-axis (m) = %.2f\n",y_coeff);
fscanf (paraftp, "%f",&z_coeff);
fprintf (fp, "The coeff. of dispersion in the z-axis (m) = %.2f\n",z_coeff);
fscanf (paraftp, "%f",&K);
fprintf (fp, "The rate constant (sec-1 or cell/q-sec) = %.20f\n",K);
fprintf (fp, "%32s\n", "Menu selection");
fprintf (fp, "\n");
fprintf (fp, " 1. Brimblecombe and Spedding (1974)'s reaction rate\n");
fprintf (fp, " 2. Freiberg (1974)'s reaction rate in ammonia-rich environment\n");
fprintf (fp, " 3. Ibusuki et al. (1990)'s reaction rate in ammonia-rich
environment\n");
fprintf (fp, "\n");
fscanf (paraftp, "%d",&opt);
fprintf (fp, "          Selection =====> %d\n",opt);

```

```

fscanf (parafp, "%2d",&max_of_printing);
fprintf (fp, "The number of printed outputs =%d\n",max_of_printing);
printf("parameters ok\n");

no_of_col    = W/w;
no_of_row    = L/l;
no_of_height = H/h;

if ( no_of_col > XMAX || no_of_row > YMAX || no_of_height > ZMAX)
{
    fprintf(fp, "\n Range of cell error\n");
    exit(1);
}

memset(c2, '\0', sizeof(c2));
memset(c3, '\0', sizeof(c3));
memset(c5, '\0', sizeof(c5));
memset(c6, '\0', sizeof(c6));

    delta_h0    = 1.6/u;
    delta_h1    = 9.81*Vs*d*d/4;
    delta_h2    = (Ts-Ta)/Ts;
    delta_h3    = delta_h1*delta_h2;
    delta_h4    = pow(delta_h3,0.3333);
    delta_h5    = pow(x1,0.6667);
    delta_h     = delta_h0*delta_h4*delta_h5;
    fprintf(fp, "The plume rise (m) = %.3f\n",delta_h);
    z0         = Hs+delta_h;
    fprintf(fp, "The effective height (m) = %.3f\n",z0);

    i0         = (int)floor(x0/w);
    j0         = (int)floor(y0/l);
    z0         = z0/h;
    dummh     = fmod(z0,1.0);
    rnd       = 1.0*rand()/RAND_MAX;
    if (rnd >= dummh)
        k0 = (int)floor(z0);
    else
        k0 = (int)ceil(z0);

    c2[i0][j0][k0] = N;
    fprintf(fp, "c2[%d][%d][%d] = %d\n",i0,j0,k0,c2[i0][j0][k0]);

for(no_of_time=1;no_of_time<=T;no_of_time++)
{
    for(i=0;i<=no_of_col;i++)
        for (j=0;j<=no_of_row;j++)
            for (k=0;k<=no_of_height;k++)
                {

```



```

        if(c2[i][j][k] != 0)
        c3[i][j][k] += c2[i][j][k];
        if (c5[i][j][k] !=0)
        c6[i][j][k] += c5[i][j][k];
    }

memcpy(c1,c2,sizeof(c2));
memset(c2,'\0',sizeof(c2));
memcpy(c4,c5,sizeof(c5));
memset(c5,'\0',sizeof(c5));

/*SO2 advection and dispersions*/

for(i=0;i<=no_of_col;i++)
    for(j=0;j<=no_of_row;j++)
        for (k=0;k<=no_of_height;k++)
            {
                for(n=1;n<=c1[i][j][k];n++)
                    {
                        /* SO2 point source location */
                        x = i;
                        y = j;
                        z = k;

                        /*SO2 advection*/
                        x = x + (u*t)/w;
                        dummx = fmod(x,1.0);
                        rnd = 1.0 * rand()/RAND_MAX;
                        if (rnd >= dummx)
                            x = floor(x);
                        else
                            x = ceil(x);

                        /*SO2 dispersion in the y-axis*/
loop1: rnd = 1.0 * rand()/RAND_MAX;
                        if (rnd < 0.5)
                            y = y - (y_coeff/1);
                        else
                            if (rnd > 0.5)
                                y = y + (y_coeff/1);
                            else
                                goto loop1;
                        dummy = fmod(y,1.0);
                        rnd = 1.0 * rand()/RAND_MAX;
                        if (rnd >= dummy)
                            y = floor(y);
                        else
                            y = ceil(y);
                    }
            }

```

```

loop2:      /*SO2 dispersion in the z-axis*/
            rnd = 1.0 * rand()/RAND_MAX;
            if (rnd < 0.5)
                z = z - (z_coeff/h);
            else
                if (rnd > 0.5)
                    z = z + (z_coeff/h);
                else
                    goto loop2;
            if (z < 0.0)
                z = -z;
            dummz = fmod(z,1.0);
            rnd = 1.0 * rand()/RAND_MAX;
            if (rnd >= dummz)
                z = floor(z);
            else
                z = ceil(z);

            ii    = (int)floor(x);
            jj    = (int)floor(y);
            kk    = (int)floor(z);

            /*SO2 to SO4 transformation*/
            switch (opt)
            {
            case 1 : /*Brimblecombe and Spedding (1974)*/
                    K_time_step = K*t;
                    rnd = 1.0 * rand()/RAND_MAX;
                    if (rnd <= K_time_step)
                        c5[ii][jj][kk] += 1 ;
                    else
                        c2[ii][jj][kk] += 1 ;
                    break;

            case 2 : /*Freiberg (1974) in ammonia-rich environment*/
                    no_of_quanta = (n*n)-((n-1)*(n-1));
                    K_time_step = K*t*no_of_quanta;
                    rnd = 1.0 * rand()/RAND_MAX;
                    if (rnd <= K_time_step)
                        c5[ii][jj][kk] += 1 ;
                    else
                        c2[ii][jj][kk] += 1 ;
                    break;

            case 3 : /*Ibusuki et al. (1990) in ammonia-rich environment*/
                    no_of_quanta = (n*n)-((n-1)*(n-1));
                    K_time_step = K*t*no_of_quanta;
                    rnd = 1.0 * rand()/RAND_MAX;
                    if (rnd <= K_time_step)

```

```

        c5[i][j][kk] -= 1 ;
        else
        c2[i][j][kk] += 1 ;
        break;

    default :
        fprintf (fp, "Out of menu\n");
    }
}

for(i=0;i<=no_of_col;i++)
    for(j=0;j<=no_of_row;j++)
        for (k=0;k<=no_of_height;k++)
            {
                for(n=1;n<=c4[i][j][k];n++)
                    {

                        /*SO4 Point source location */
                        x = i;
                        y = j;
                        z = k;

                        /*SO4 advection*/
                        x = x + (u*t)/w;
                        dummx = fmod(x,1.0);
                        rnd = 1.0 * rand()/RAND_MAX;
                        if (rnd >= dummx)
                            x = floor(x);
                        else
                            x = ceil(x);

                        /*SO4 dispersion in the y-axis*/
loop3:      rnd = 1.0 * rand()/RAND_MAX;
                        if (rnd < 0.5)
                            y = y - (y_coeff/1);
                        else
                            if (rnd > 0.5)
                                y = y + (y_coeff/1);
                            else
                                goto loop3;
                        dummy = fmod(y,1.0);
                        rnd = 1.0 * rand()/RAND_MAX;
                        if (rnd >= dummy)
                            y = floor(y);
                        else
                            y = ceil(y);
                    }
            }
}

```

```

/*SO4 dispersion in the z-axis*/
loop4:  rnd = 1.0 * rand()/RAND_MAX;
        if (rnd < 0.5)
            z = z - (z_coeff/h);
        else
            if (rnd > 0.5)
                z = z + (z_coeff/h);
            else
                goto loop4;
            if (z < 0.0)
                z = -z;
        dummz = fmod(z,1.0);
        rnd = 1.0 * rand()/RAND_MAX;
        if (rnd >= dummz)
            z = floor(z);
        else
            z = ceil(z);

        ii    = (int)floor(x);
        jj    = (int)floor(y);
        kk    = (int)floor(z);
        if (ii >= 0 && ii < no_of_col && jj < no_of_row && kk < no_of_height)
            c5[ii][jj][kk] += 1;
    }
}

for(i=0;i<=no_of_col;i++)
    for (j=0;j<=no_of_row;j++)
        for (k=0;k<=no_of_height;k++)
            {
                if(c2[i][j][k] != 0)
                    c3[i][j][k] += c2[i][j][k];
                if (c5[i][j][k] !=0)
                    c6[i][j][k] += c5[i][j][k];
            }

/*print output data*/
for(no_of_printing=1;no_of_printing<=max_of_printing;no_of_printing++)
{
    fscanf (parafp, "%f",&x0);
    fprintf (fp, "\n\nThe location in the x-axis (m) = %.2f\n",x0);
    i0 = (int)floor(x0/w);
    fprintf(fp, "\n\nThe number of SO2 quanta in each cell is : ");
    sum = 0;
    for(kk=no_of_height;kk>=0;kk--)
    {
        fprintf(fp, "\n");
        for(ij=0;ij<=no_of_row;ij++)

```

```
    {
        fprintf(fp, "%3d ", c3[i0][jj][kk]);
        sum += c3[i0][jj][kk];
    }
}
fprintf(fp, "\nThe total of SO2 quanta are =%d\n", sum);

sum = 0;
fprintf(fp, "\nThe number of SO4 quanta in each cell is : ");
for(kk=no_of_height;kk>=0;kk--)
{
    fprintf(fp, "\n");
    for(jj=0;jj<=no_of_row;jj++)
    {
        fprintf(fp, "%3d ", c6[i0][jj][kk]);
        sum += c6[i0][jj][kk];
    }
}
fprintf(fp, "\nThe total of SO4 quanta is =%d\n", sum);
}

fprintf(fp, "\n");
fclose(fp);
fclose(parafp);
}
```

- Program of Physico-Chemical Mathematical Model for Freiberg (1974)'s Reaction Rate in Ammonia-Deficient Environment

```

#include <stdio.h>
#include <math.h>
#include <time.h>
#include <memory.h>
#define XMAX 107
#define YMAX 85
#define ZMAX 35
#define RAND_MAX 32767 /* (2^15)-1 */
main (ac,av)
int ac;
char **av;
{
int    c1[XMAX][YMAX][ZMAX],c2[XMAX][YMAX][ZMAX],
       c3[XMAX][YMAX][ZMAX],c4[XMAX][YMAX][ZMAX],
       c5[XMAX][YMAX][ZMAX],c6[XMAX][YMAX][ZMAX],
       i,j,k,i0,j0,k0,ii,jj,kk,n,N,T,no_of_row,no_of_col,no_of_height,
       opt,no_of_quanta,no_of_time,no_of_printing,max_of_printing,sum;
float  x,x0,y,y0,z,z0,t,w,l,h,dummx,dummy,dummys,y_coeff,z_coeff,md,
       W,L,H,K,K_time_step,u,Hs,Vs,d,delta_h,x1,Ts,Ta,dummh,
       delta_h0,delta_h1,delta_h2,delta_h3,delta_h4,delta_h5,Q,NH3,sum1,
       c7[XMAX][YMAX][ZMAX],c8[XMAX][YMAX][ZMAX];
FILE   *fp,*parafp;
char   fname[50],name[50];
if ( ac != 2)
{
    printf("No parameter file name\n");
    exit(1);
}
parafp = fopen(av[1],"r");
fscanf (parafp, "%s",fname);
fp = fopen(fname,"w");
fprintf (fp,"The output file name is %s\n",fname);
fscanf (parafp, "%s",name);
fprintf (fp,"The condition of the reaction is %s\n",name);
fscanf (parafp, "%f",&W);
fprintf (fp,"The width of interested area (m) = %.2f\n",W);
fscanf (parafp, "%f",&L);
fprintf (fp,"The length of interested area (m) = %.2f\n",L);
fscanf (parafp, "%f",&H);
fprintf (fp,"The height of interested area (m) = %.2f\n",H);
fscanf (parafp, "%f",&w);
fprintf (fp,"The width of each cell (m) = %.2f\n",w);
fscanf (parafp, "%f",&l);
fprintf (fp,"The length of each cell (m) = %.2f\n",l);

```

```

fscanf (parafp, "%f", &h);
fprintf (fp, "The height of each cell (m) = %.2f\n", h);
fscanf (parafp, "%f", &x0);
fprintf (fp, "The location of each point source in the x-axis (m) = %.2f\n", x0);
fscanf (parafp, "%f", &y0);
fprintf (fp, "The location of each point source in the y-axis (m) = %.2f\n", y0);
fscanf (parafp, "%d", &N);
fprintf (fp, "The number of SO2 quanta in each point source = %d\n", N);
fscanf (parafp, "%f", &Q);
fprintf (fp, "The SO2 emission rate (g/sec) = %.2f\n", Q);
fscanf (parafp, "%f", &u);
fprintf (fp, "The velocity of wind at the stack height (m/sec) = %.2f\n", u);
fscanf (parafp, "%f", &Hs);
fprintf (fp, "The height of stack (m) = %.2f\n", Hs);
fscanf (parafp, "%f", &Vs);
fprintf (fp, "The velocity of gas (m/s) = %.3f\n", Vs);
fscanf (parafp, "%f", &d);
fprintf (fp, "The diameter of stack (m) = %.2f\n", d);
fscanf (parafp, "%f", &x1);
fprintf (fp, "The downwind distance from source (m) = %.2f\n", x1);
fscanf (parafp, "%f", &Ts);
fprintf (fp, "The temperature of stack (K) = %.2f\n", Ts);
fscanf (parafp, "%f", &Ta);
fprintf (fp, "The atmosphere temperature (K) = %.3f\n", Ta);
fscanf (parafp, "%d", &T);
fprintf (fp, "The number of time step = %d\n", T);
fscanf (parafp, "%f", &t);
fprintf (fp, "The time step (sec) = %.2f\n", t);
fscanf (parafp, "%f", &y_coeff);
fprintf (fp, "The coeff. of dispersion in the y-axis (m) = %.2f\n", y_coeff);
fscanf (parafp, "%f", &z_coeff);
fprintf (fp, "The coeff. of dispersion in the z-axis (m) = %.2f\n", z_coeff);
fscanf (parafp, "%f", &NH3);
fprintf (fp, "The initial concentration of NH3 in each cell (ppb) = %.2f\n", NH3);
fscanf (parafp, "%f", &K);
fprintf (fp, "The rate constant (m12/mol4-sec) = %.2f\n", K);
fprintf (fp, "%32s\n", "Menu selection");
fprintf (fp, "\n");
fprintf (fp, " 1. Freiberg(1974)'s reaction rate in ammonia-deficient
environment\n");
fprintf (fp, "\n");
fscanf (parafp, "%d", &opt);
fprintf (fp, "      Selection =====> %d\n", opt);
fscanf (parafp, "%2d", &max_of_printing);
fprintf (fp, "The number of printed outputs = %d\n", max_of_printing);
printf("parameters ok\n");

no_of_col   = W/w;
no_of_row   = L/l;

```

```

no_of_height = H/h;

if ( no_of_col > XMAX || no_of_row > YMAX || no_of_height > ZMAX)
{
    fprintf(fp, "\n Range of cell error\n");
    exit(1);
}

memset(c2, '\0', sizeof(c2));
memset(c3, '\0', sizeof(c3));
memset(c5, '\0', sizeof(c5));
memset(c6, '\0', sizeof(c6));
memset(c7, '\0', sizeof(c7));
memset(c8, '\0', sizeof(c8));

    delta_h0      = 1.6/u;
    delta_h1      = 9.81*Vs*d*d/4;
    delta_h2      = (Ts-Ta)/Ts;
    delta_h3      = delta_h1*delta_h2;
    delta_h4      = pow(delta_h3,0.3333);
    delta_h5      = pow(x1,0.6667);
    delta_h       = delta_h0*delta_h4*delta_h5;
    fprintf(fp, "The plume rise (m) = %.3f\n", delta_h);
    z0 = Hs+delta_h;
    fprintf(fp, "The effective height (m) = %.3f\n", z0);

    i0      = (int)floor(x0/w);
    j0      = (int)floor(y0/l);
    z0      = z0/h;
    dummh   = fmod(z0,1.0);
    rnd     = 1.0*rand()/RAND_MAX;
    if (rnd >= dummh)
        k0 = (int)floor(z0);
    else
        k0 = (int)ceil(z0);

    c2[i0][j0][k0] = N;
    fprintf(fp, "c2[%d][%d][%d] = %d\n", i0, j0, k0, c2[i0][j0][k0]);

for(no_of_time=1;no_of_time<=T;no_of_time++)
{
    for(i=0;i<=no_of_col;i++)
        for(j=0;j<=no_of_row;j++)
            for(k=0;k<=no_of_height;k++)
            {
                if(c2[i][j][k] != 0)
                    c3[i][j][k] += c2[i][j][k];
                if (c5[i][j][k] !=0)

```



```

        c6[i][j][k] += c5[i][j][k];
    }

    memcpy(c1,c2,sizeof(c2));
    memset(c2,'\0',sizeof(c2));
    memcpy(c4,c5,sizeof(c5));
    memset(c5,'\0',sizeof(c5));

    /*SO4 advection and dispersions*/
    for(i=0;i<=no_of_col;i++)
        for(j=0;j<=no_of_row;j++)
            for (k=0;k<=no_of_height;k++)
    {
        for(n=1;n<=c4[i][j][k];n++)
        {
            /*SO4 Point source location */
            x = i;
            y = j;
            z = k;

            /*SO4 advection*/
            x = x + (u*t)/w;
            dummx = fmod(x,1.0);
            rnd = 1.0 * rand()/RAND_MAX;
            if (rnd >= dummx)
                x = floor(x);
            else
                x = ceil(x);

            /*SO4 dispersion in the y-axis*/
loop3:            rnd = 1.0 * rand()/RAND_MAX;
            if (rnd < 0.5)
                y = y - (y_coeff/1);
            else
                if (rnd > 0.5)
                    y = y + (y_coeff/1);
                else
                    goto loop3;

            dummy = fmod(y,1.0);
            rnd = 1.0 * rand()/RAND_MAX;
            if (rnd >= dummy)
                y = floor(y);
            else
                y = ceil(y);

loop4:            /*SO4 dispersion in the z-axis*/
            rnd = 1.0 * rand()/RAND_MAX;
            if (rnd < 0.5)

```

```

        z = z - (z_coeff h);
    else
        if (rnd > 0.5)
            z = z + (z_coeff h);
        else
            goto loop4;
        if (z < 0.0)
            z = -z;
    dummz = fmod(z,1.0);
    rnd = 1.0 * rand()/RAND_MAX;
    if (rnd >= dummz)
        z = floor(z);
    else
        z = ceil(z);

    ii    = (int)floor(x);
    jj    = (int)floor(y);
    kk    = (int)floor(z);

if (ii >= 0 && ii < no_of_col && jj < no_of_row && kk < no_of_height)
    {
    c5[ii][jj][kk] += 1;
    if ((2*c5[ii][jj][kk]*Q*t*0.000001)/(N*64) > (NH3*0.000001)/24.5)
        {
        c5[ii][jj][kk] -= 1;
        goto loop3;
        }
    }
    c8[ii][jj][kk] = NH3-(2*c5[ii][jj][kk]*Q*t*24.5)/(N*64);
}

/*SO2 advection and dispersions*/
for(i=0;i<=no_of_col;i++)
    for(j=0;j<=no_of_row;j++)
        for (k=0;k<=no_of_height;k++)
    {
        for(n=1;n<=c1[i][j][k];n++)
    {
        /*SO2 Point source location */
        x = i;
        y = j;
        z = k;

        /*SO2 advection*/
        x = x + (u*t)/w;
        dummx = fmod(x,1.0);
        rnd = 1.0 * rand()/RAND_MAX;

```

```

if (rnd >= dummx)
    x = floor(x);
else
    x = ceil(x);

/*SO2 dispersion in the y-axis*/
loop1: rnd = 1.0 * rand()/RAND_MAX;
if (rnd < 0.5)
    y = y - (y_coeff/l);
else
    if (rnd > 0.5)
        y = y + (y_coeff/l);
    else
        goto loop1;
dummy = fmod(y,1.0);
rnd = 1.0 * rand()/RAND_MAX;
if (rnd >= dummy)
    y = floor(y);
else
    y = ceil(y);

/*SO2 dispersion in the z-axis*/
loop2: rnd = 1.0 * rand()/RAND_MAX;
if (rnd < 0.5)
    z = z - (z_coeff/h);
else
    if (rnd > 0.5)
        z = z + (z_coeff/h);
    else
        goto loop2;

if (z < 0.0)
    z = -z;
dummz = fmod(z,1.0);
rnd = 1.0 * rand()/RAND_MAX;
if (rnd >= dummz)
    z = floor(z);
else
    z = ceil(z);

ii    = (int)floor(x);
jj    = (int)floor(y);
kk    = (int)floor(z);

/*SO2 to SO4 transformation*/

switch (opt)
{
case 1 : /*Freiberg (1974) in ammonia-deficient environment */

```

```

/*[NH3]t = [NH3]o - 2[SO4]*/
c7[i][j][kk] = ((NH3*0.000001)/24.5)-((2*c5[i][j][kk]*Q*t*0.000001)/(N*64));
if (c7[i][j][kk] >= ((2*c5[i][j][kk]*Q*t*0.000001)/(N*64)))
{
no_of_quanta = (n*n)-((n-1)*(n-1));
K_time_step =
(K*t*no_of_quanta*c7[i][j][kk]*c7[i][j][kk]*c7[i][j][kk]*Q*t*0.000001)/(64*N);
rnd = 1.0 * rand()/RAND_MAX;
if (rnd <= K_time_step)
{
c5[i][j][kk] += 1 ;
}
else
c2[i][j][kk] += 1;
}
else
{
/*fprintf(fp, "\nNH3 insufficiency in c7[%d][%d][%d]\n", i, j, kk);*/
c2[i][j][kk] += 1 ;
}
c8[i][j][kk] = NH3-(2*c5[i][j][kk]*Q*t*24.5)/(N*64);
break;

default :
fprintf (fp, "Out of menu\n");
}
}
}
}

for(i=0;i<=no_of_col;i++)
for (j=0;j<=no_of_row;j++)
for (k=0;k<=no_of_height;k++)
{
if(c2[i][j][k] != 0)
c3[i][j][k] += c2[i][j][k];
if (c5[i][j][k] !=0)
c6[i][j][k] += c5[i][j][k];
}

/*print output data*/
for(no_of_printing=1;no_of_printing<=max_of_printing;no_of_printing++)
{
fscanf (parafp, "%f", &x0);
fprintf (fp, "\nThe location in the x-axis (m) = %.2f\n", x0);
i0 = (int)floor(x0/w);
fprintf(fp, "\nThe number of SO2 quanta in each cell is : ");

```

```

sum = 0;
for(kk=no_of_height;kk>=0;kk--)
{
    fprintf(fp, "\n");
    for(ij=0;ij<=no_of_row;ij++)
    {
        fprintf(fp, "%3d ", c3[i0][ij][kk]);
        sum += c3[i0][ij][kk];
    }
}
fprintf(fp, "\nThe total number of SO2 quanta is =%d\n", sum);

sum1 = 0;
fprintf(fp, "\nThe concentration of remaining NH3 (ppb) in each cell is : ");
for(kk=no_of_height;kk>=0;kk--)
{
    fprintf(fp, "\n");
    for(ij=0;ij<=no_of_row;ij++)
    {
        if(c8[i0][ij][kk] != 0)
        {
            fprintf(fp, "%7.2f ", c8[i0][ij][kk]);
            sum1 += c8[i0][ij][kk];
        }
        else
            fprintf(fp, "%7.2f ", NH3);
    }
}
fprintf(fp, "\nThe total concentration of remaining NH3 that reacts with SO4
(ppb) is =%.2f\n", sum1);

sum = 0;
fprintf(fp, "\nThe number of SO4 quanta in each cell is : ");
for(kk=no_of_height;kk>=0;kk--)
{
    fprintf(fp, "\n");
    for(ij=0;ij<=no_of_row;ij++)
    {
        fprintf(fp, "%3d ", c6[i0][ij][kk]);
        sum += c6[i0][ij][kk];
    }
}
fprintf(fp, "\nThe total number of SO4 quanta is =%d\n", sum);
}

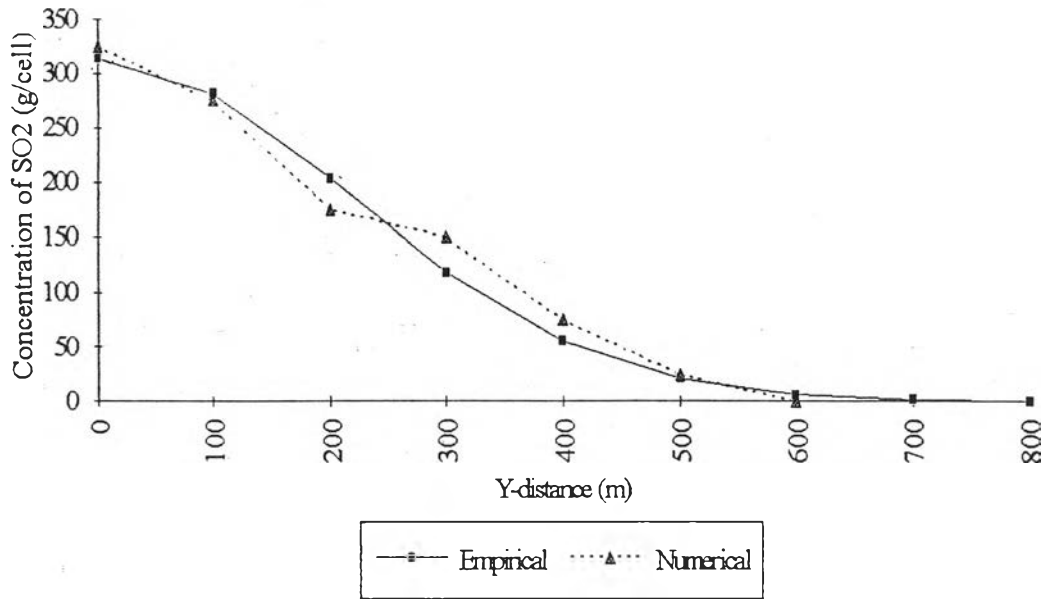
fprintf(fp, "\n");
fclose(fp);
fclose(parafp);
}

```

Appendix B

Results of Simulations

a)



b)

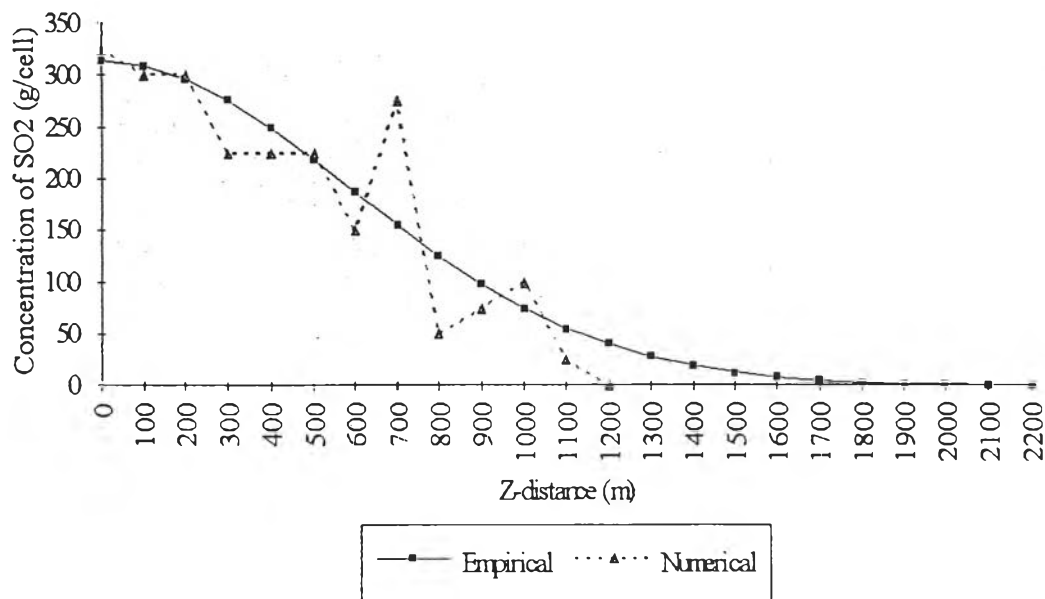


Figure 4.1 Comparison of the Empirical and Numerical Concentrations of SO₂ for Atmospheric Stability Class A at 1 km Downwind from the Source

a) Varying Y-Distance (Fixed z=0)

b) Varying Z-Distance (Fixed y=0)

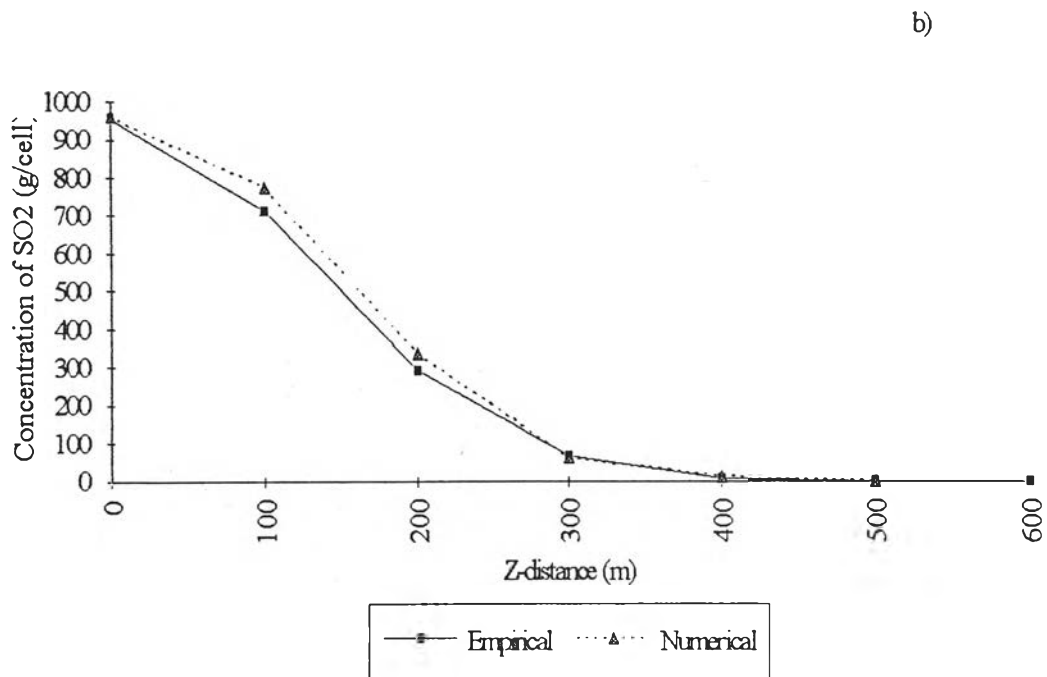
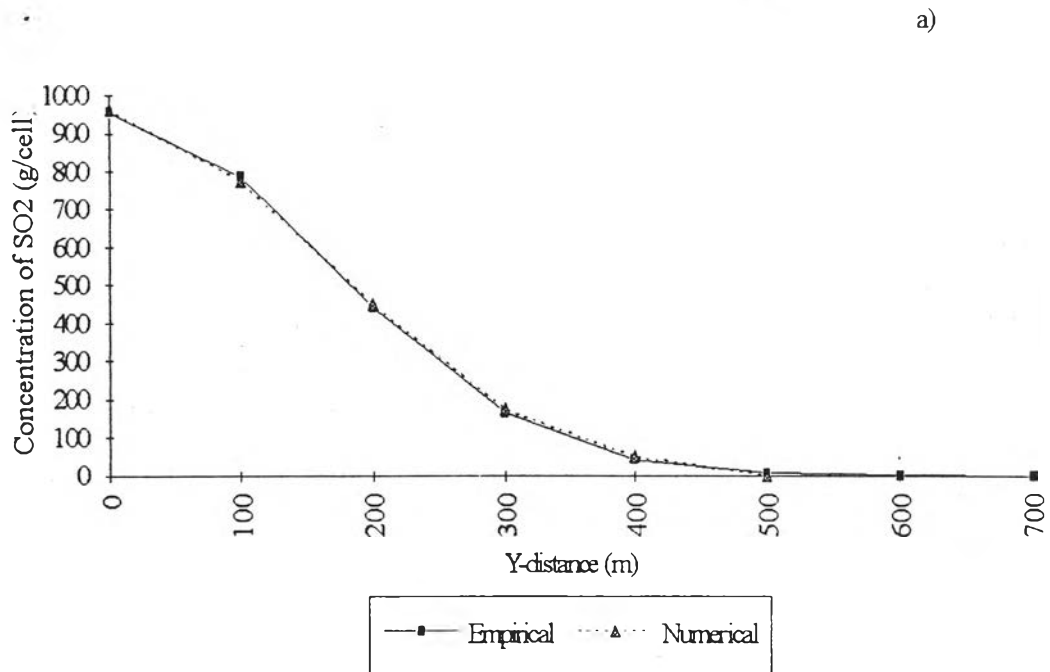


Figure 4.2 Comparison of the Empirical and Numerical Concentrations of SO_2 for Atmospheric Stability Class B at 1 km Downwind from the Source

a) Varying Y-Distance (Fixed $z=0$)

b) Varying Z-Distance (Fixed $y=0$)

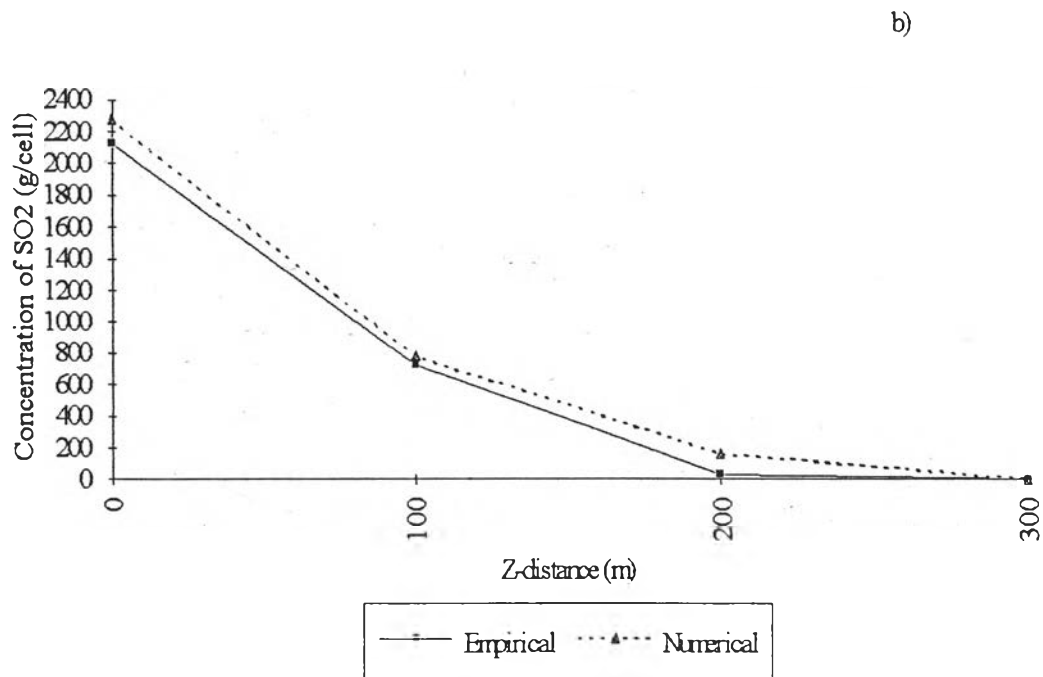
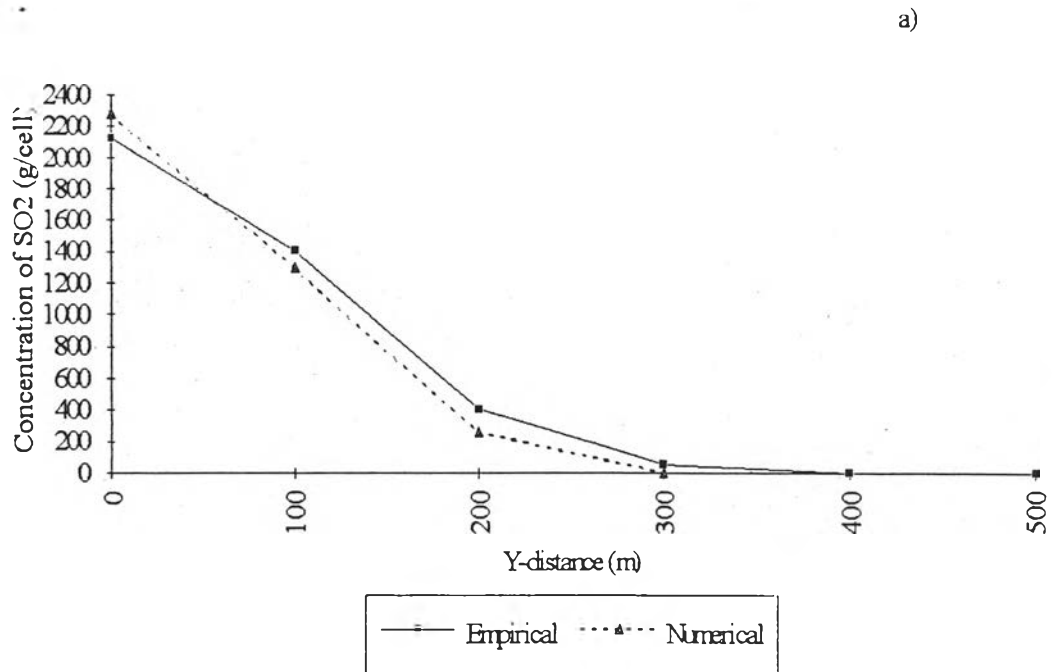
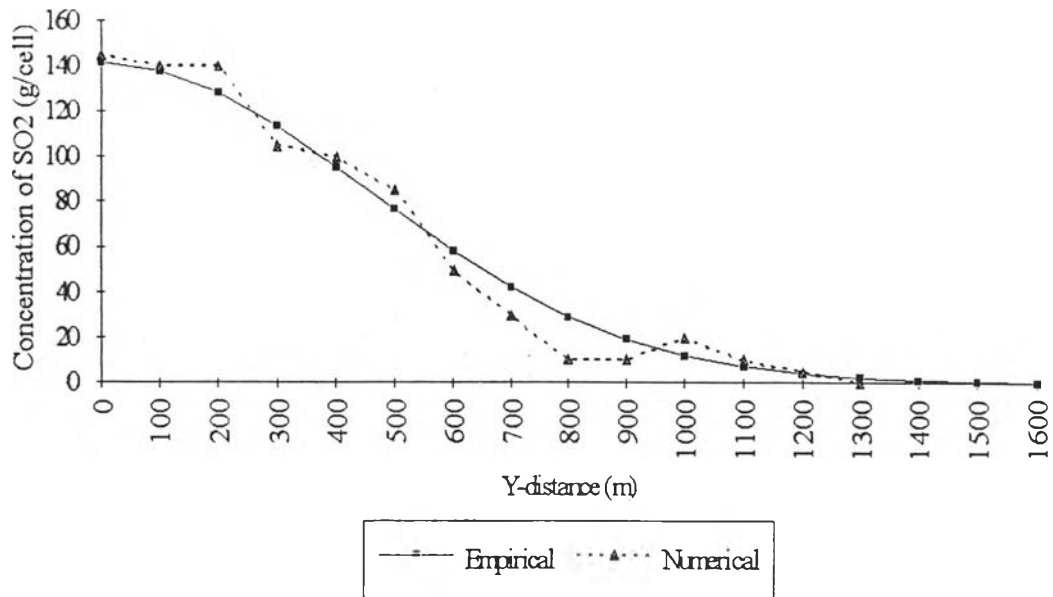


Figure 4.3 Comparison of the Empirical and Numerical Concentrations of SO₂ for Atmospheric Stability Class C at 1 km Downwind from the Source
 a) Varying Y-Distance (Fixed z=0) b) Varying Z-Distance (Fixed y=0)

a)



b)

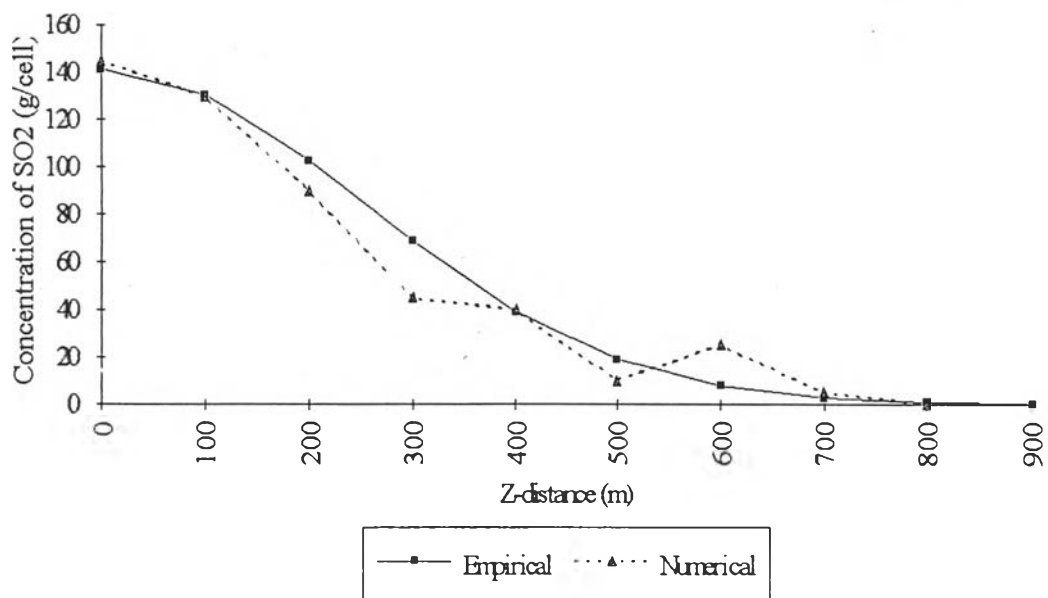


Figure 4.4 Comparison of the Empirical and Numerical Concentrations of SO_2 for Atmospheric Stability Class C at 5 km Downwind from the Source
 a) Varying Y-Distance (Fixed $z=0$) b) Varying Z-Distance (Fixed $y=0$)

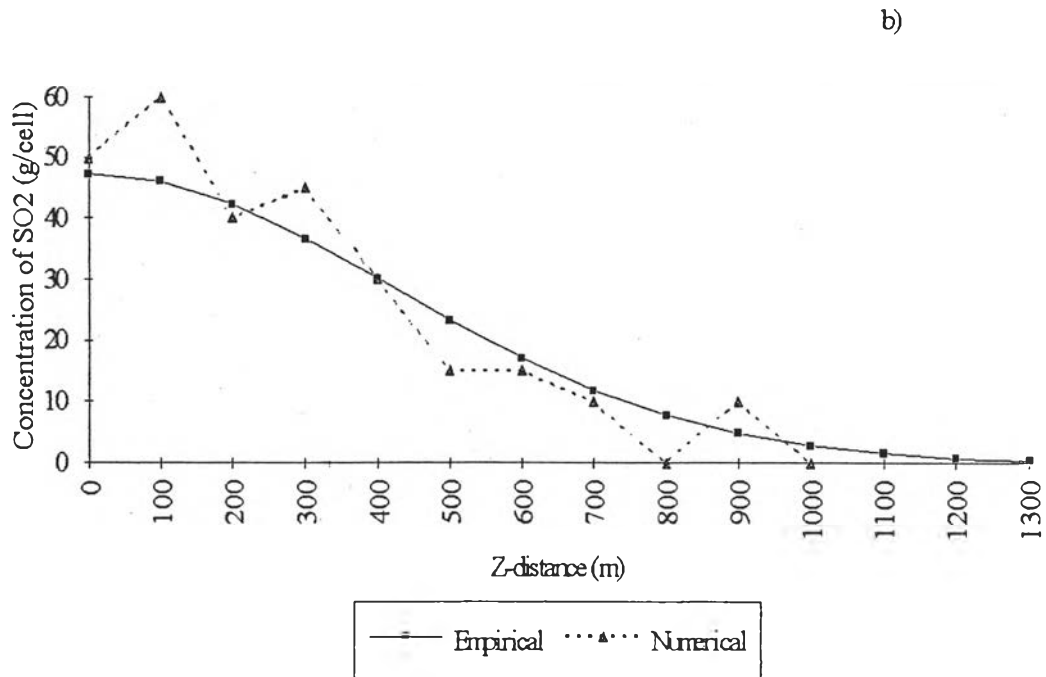
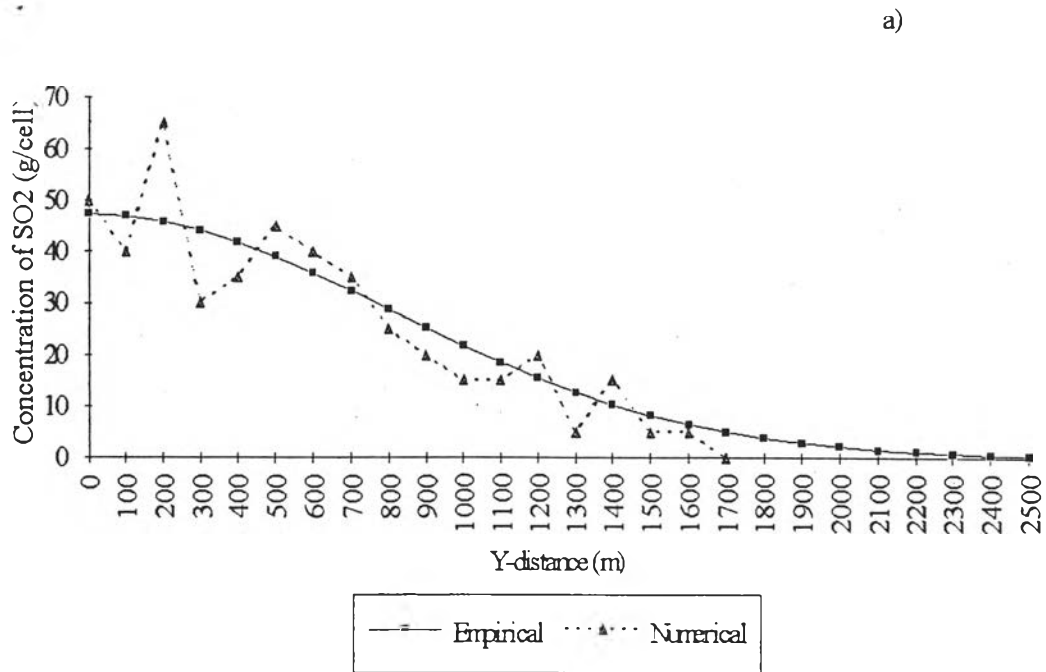


Figure 4.5 Comparison of the Empirical and Numerical Concentrations of SO_2 for Atmospheric Stability Class C at 10 km Downwind from the Source

a) Varying Y-Distance (Fixed $z=0$)

b) Varying Z-Distance (Fixed $y=0$)

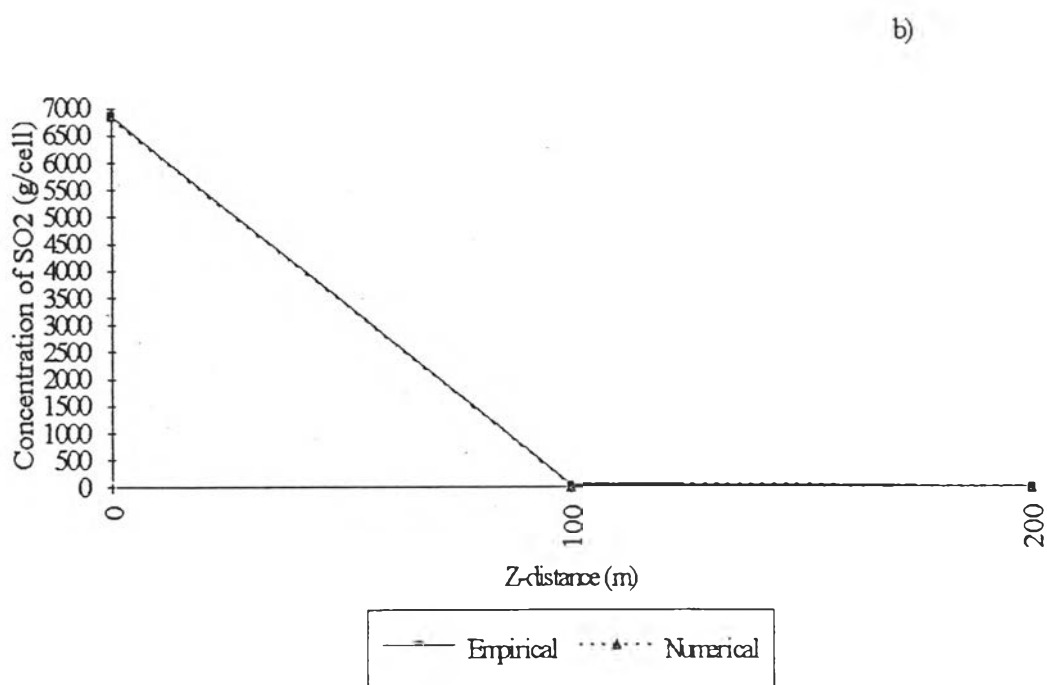
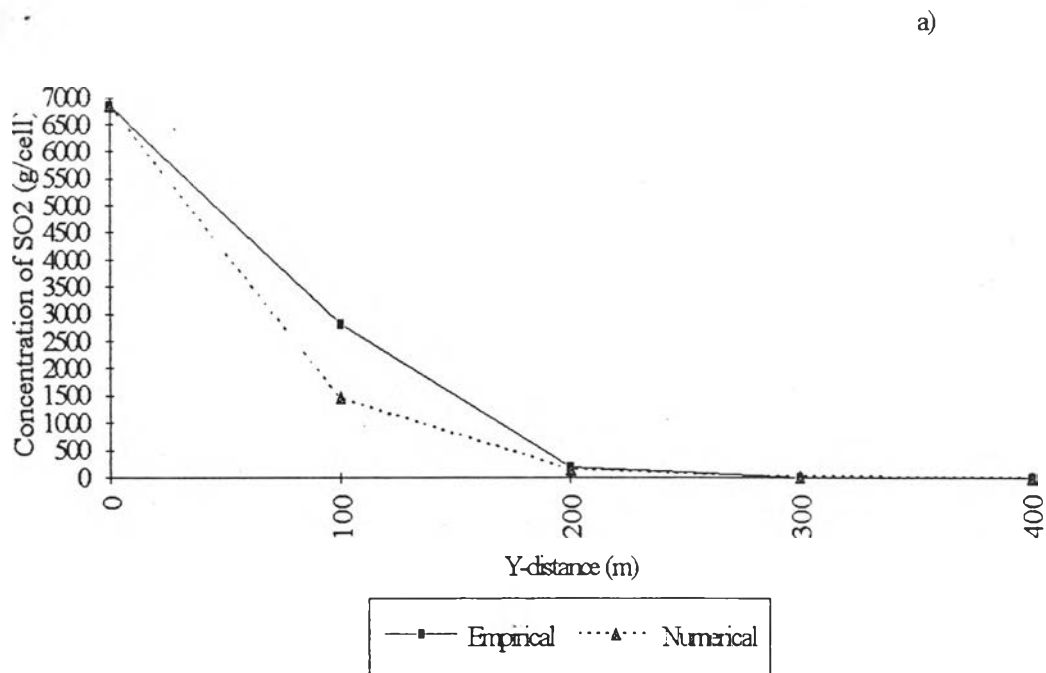


Figure 4.6 Comparison of the Empirical and Numerical Concentrations of SO_2 for Atmospheric Stability Class D at 1 km Downwind from the Source

a) Varying Y-Distance (Fixed $z=0$) b) Varying Z-Distance (Fixed $y=0$)

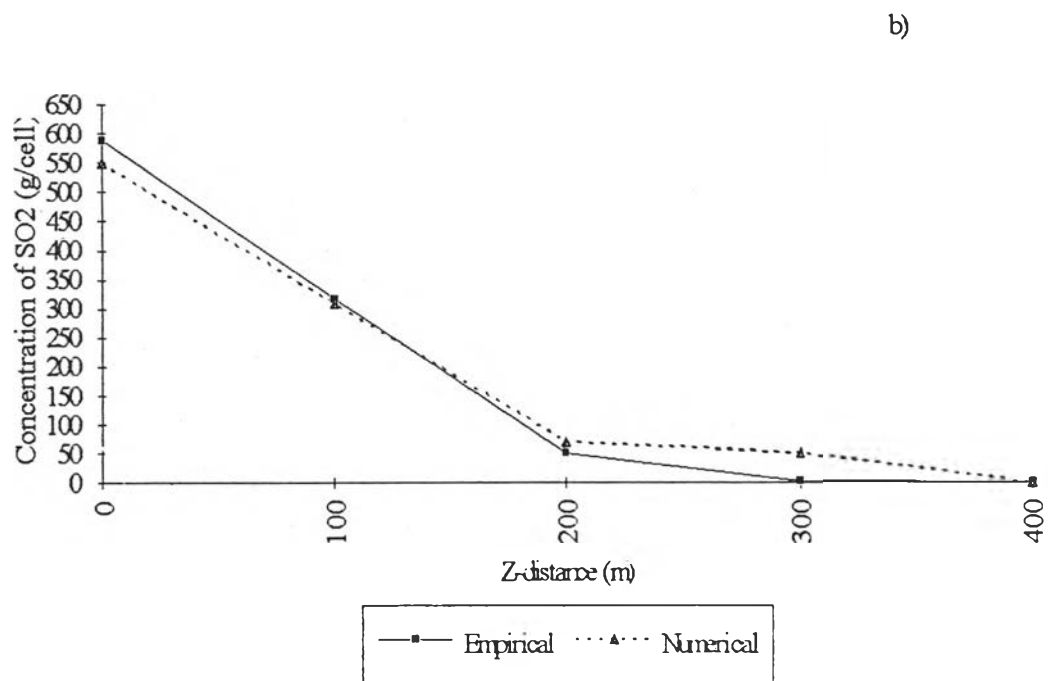
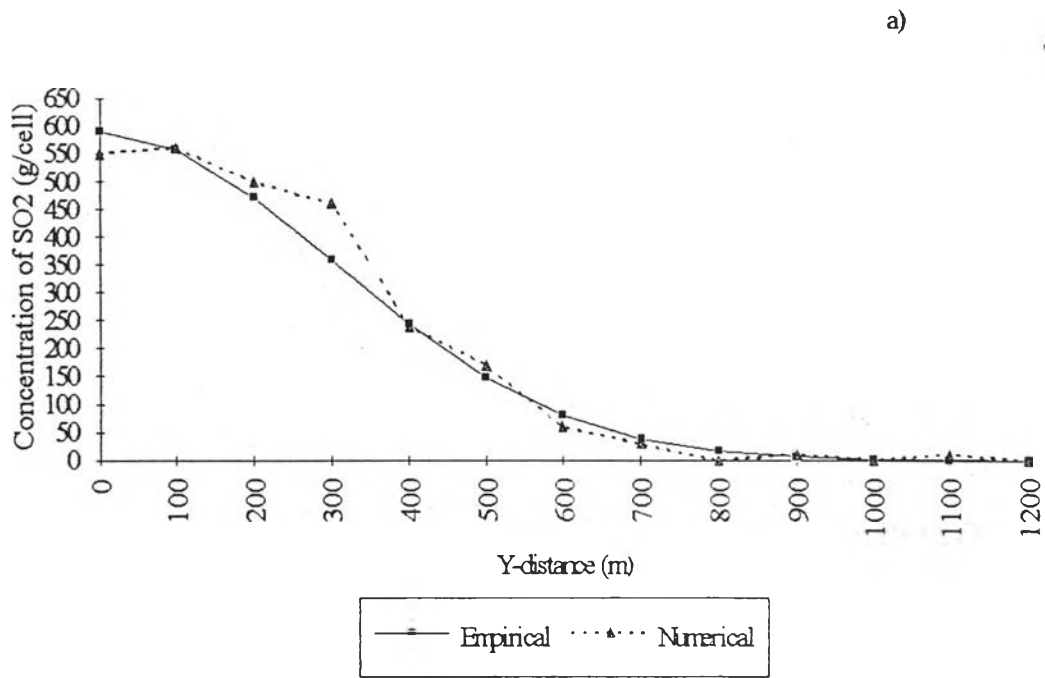


Figure 4.7 Comparison of the Empirical and Numerical Concentrations of SO₂ for Atmospheric Stability Class D at 5 km Downwind from the Source
 a) Varying Y-Distance (Fixed z=0) b) Varying Z-Distance (Fixed y=0)

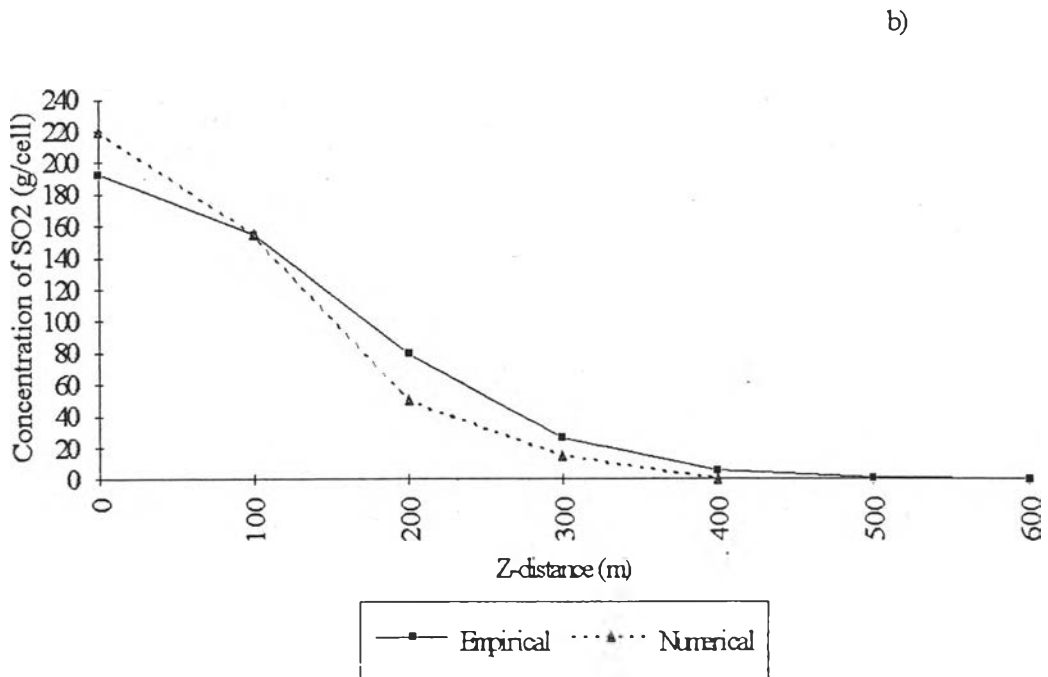
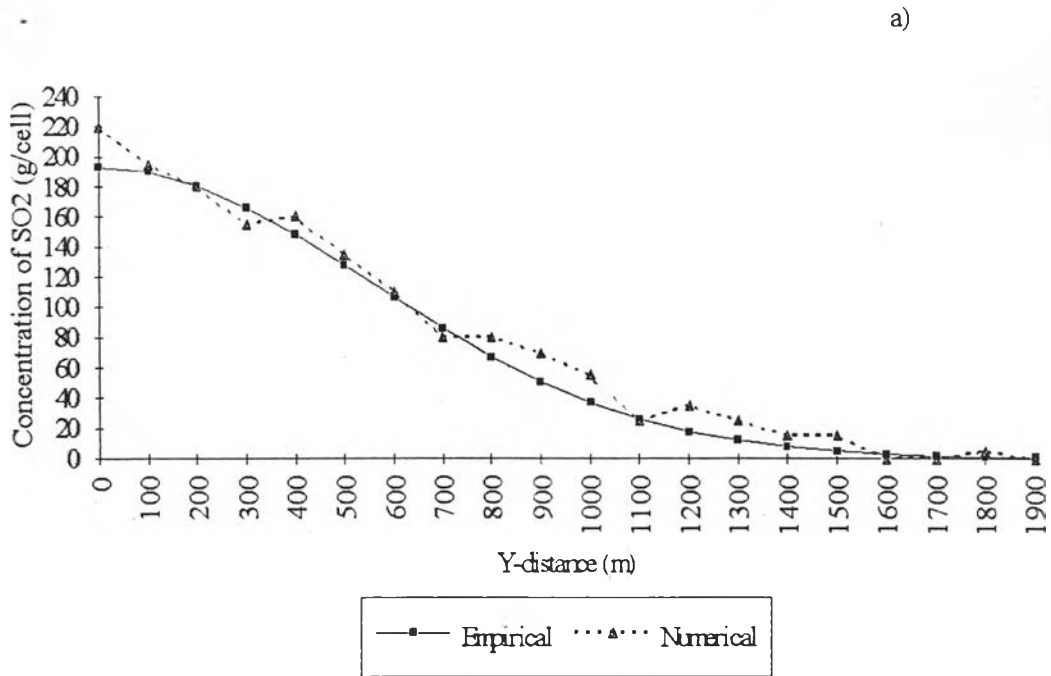


Figure 4.8 Comparison of the Empirical and Numerical Concentrations of SO₂ for Atmospheric Stability Class D at 10 km Downwind from the Source

a) Varying Y-Distance (Fixed z=0)

b) Varying Z-Distance (Fixed y=0)

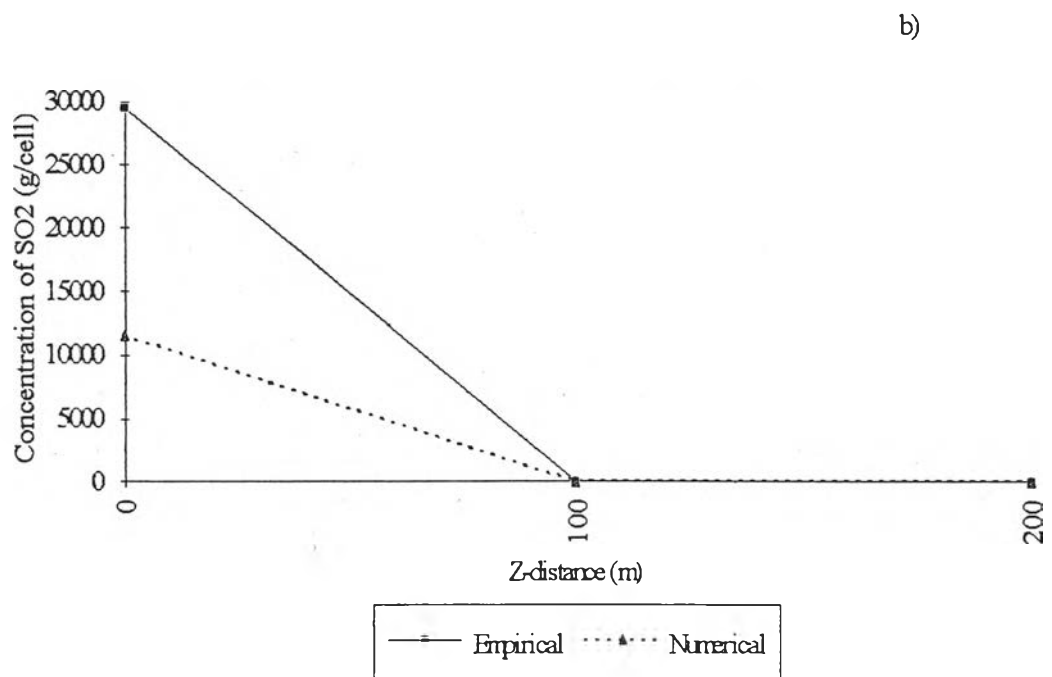
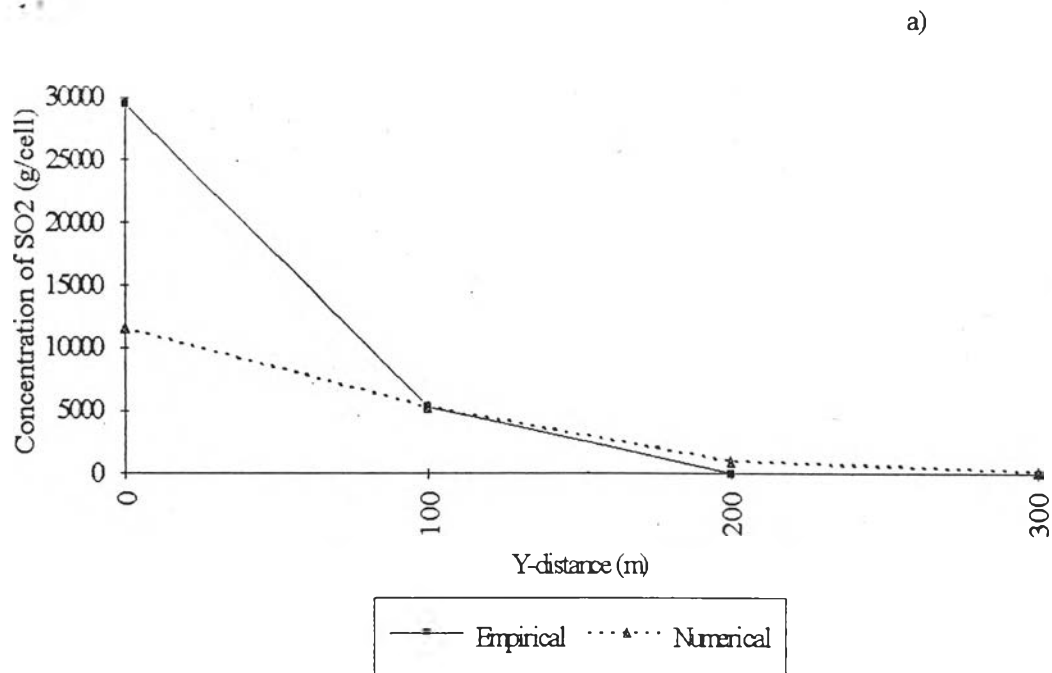
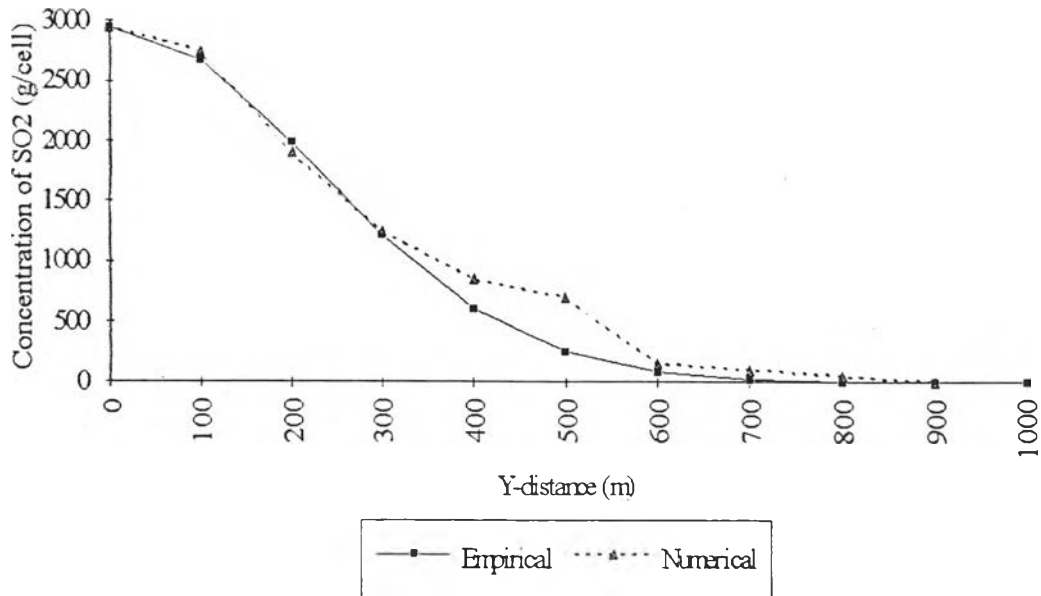


Figure 4.9 Comparison of the Empirical and Numerical Concentrations of SO_2 for Atmospheric Stability Class E at 1 km Downwind from the Source

a) Varying Y-Distance (Fixed $z=0$)

b) Varying Z-Distance (Fixed $y=0$)

a)



b)

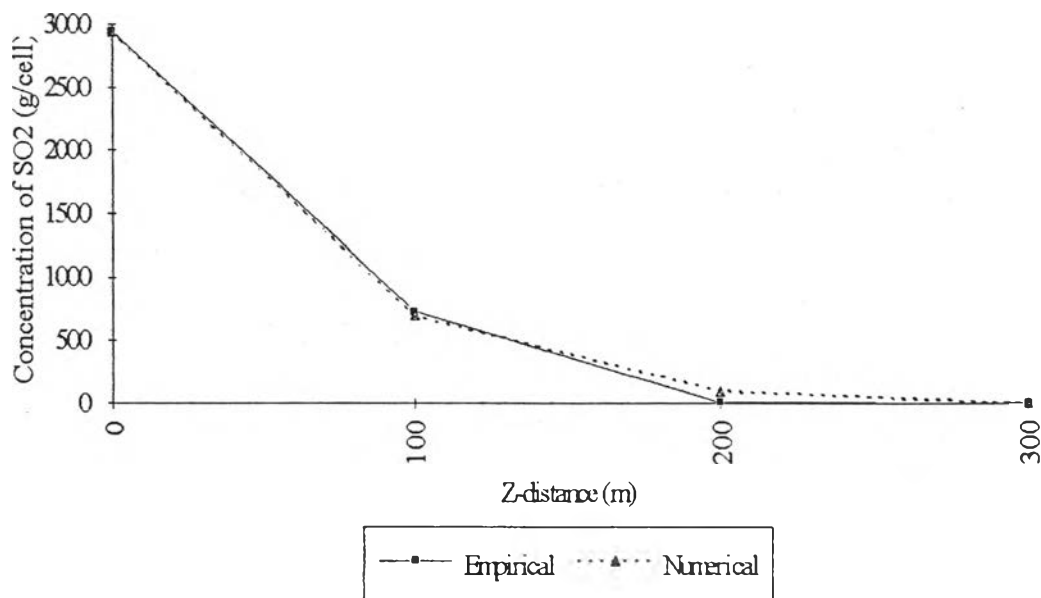


Figure 4.10 Comparison of the Empirical and Numerical Concentrations of SO₂ for Atmospheric Stability Class E at 5 km Downwind from the Source

a) Varying Y-Distance (Fixed z=0)

b) Varying Z-Distance (Fixed y=0)

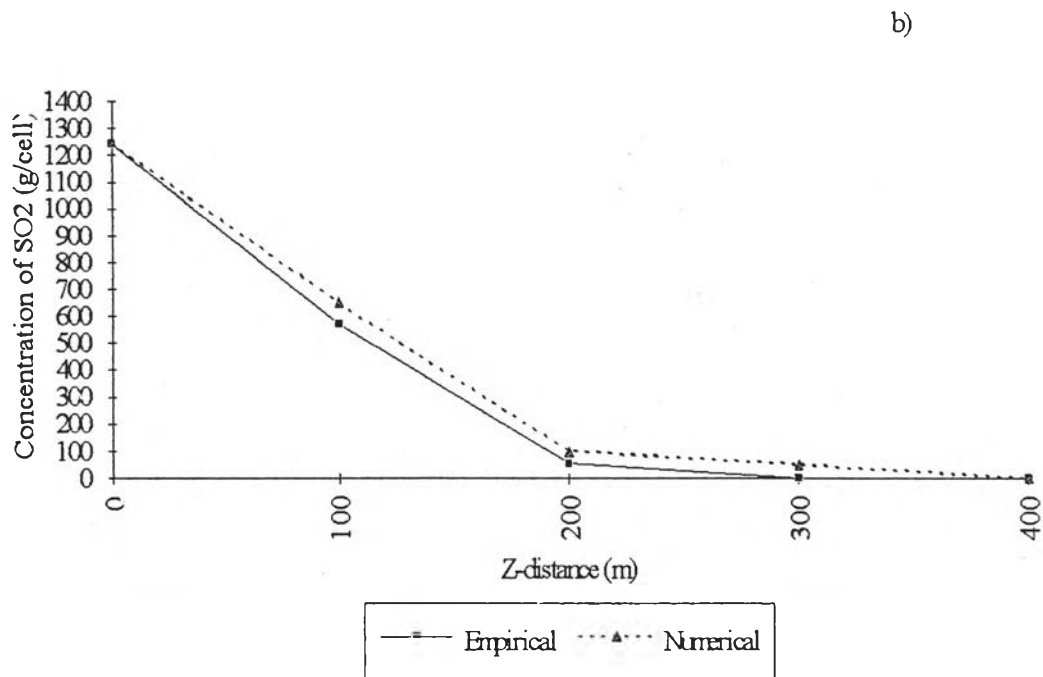
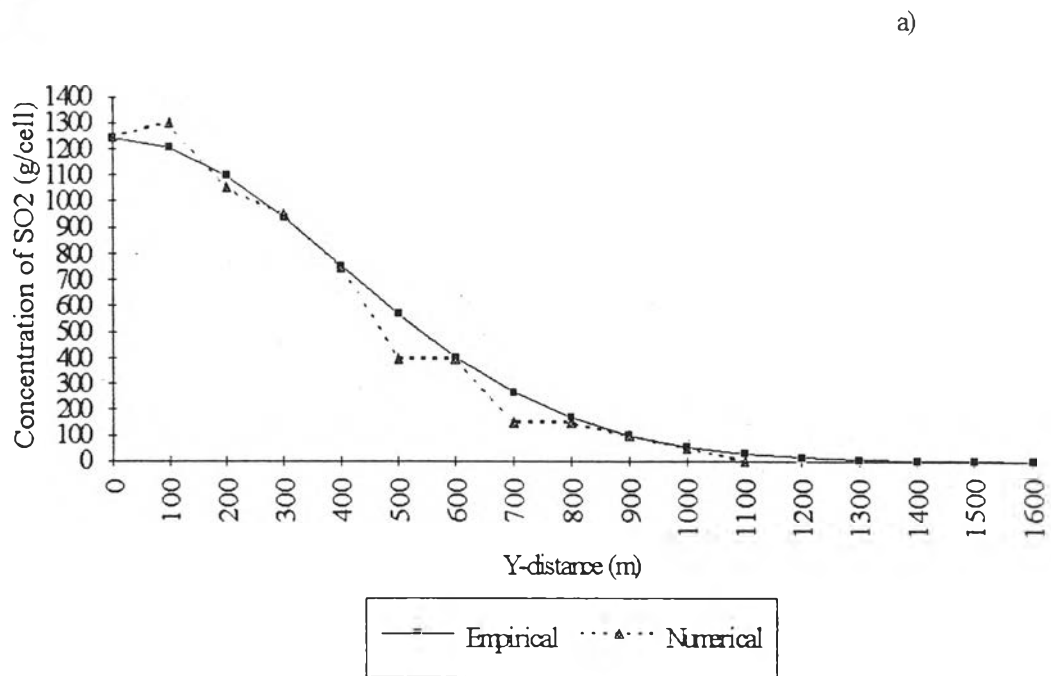
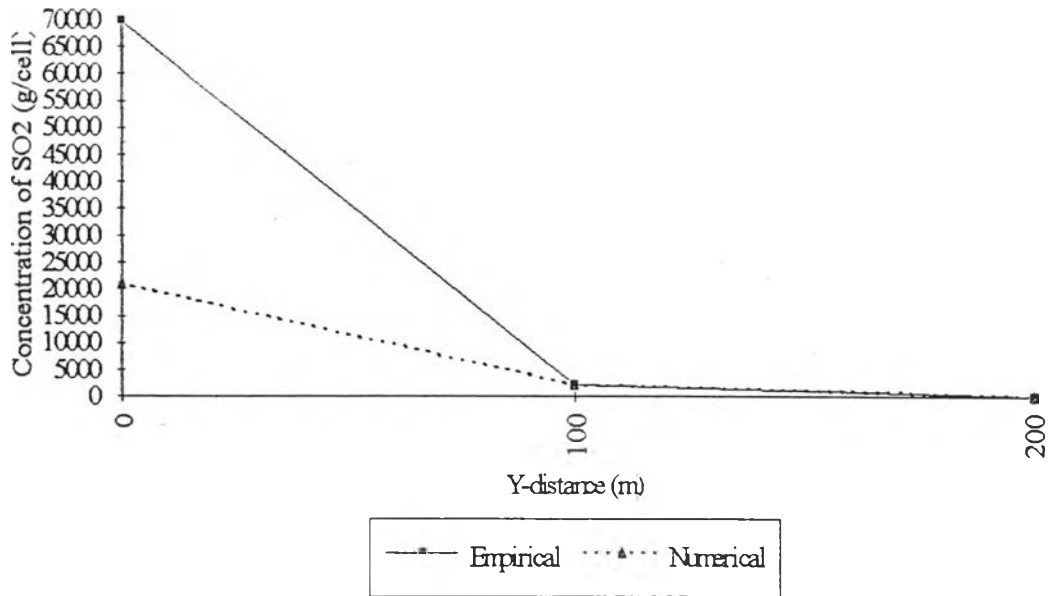


Figure 4.11 Comparison of the Empirical and Numerical Concentrations of SO₂ for Atmospheric Stability Class E at 10 km Downwind from the Source

a) Varying Y-Distance (Fixed z=0)

b) Varying Z-Distance (Fixed y=0)

a)



b)

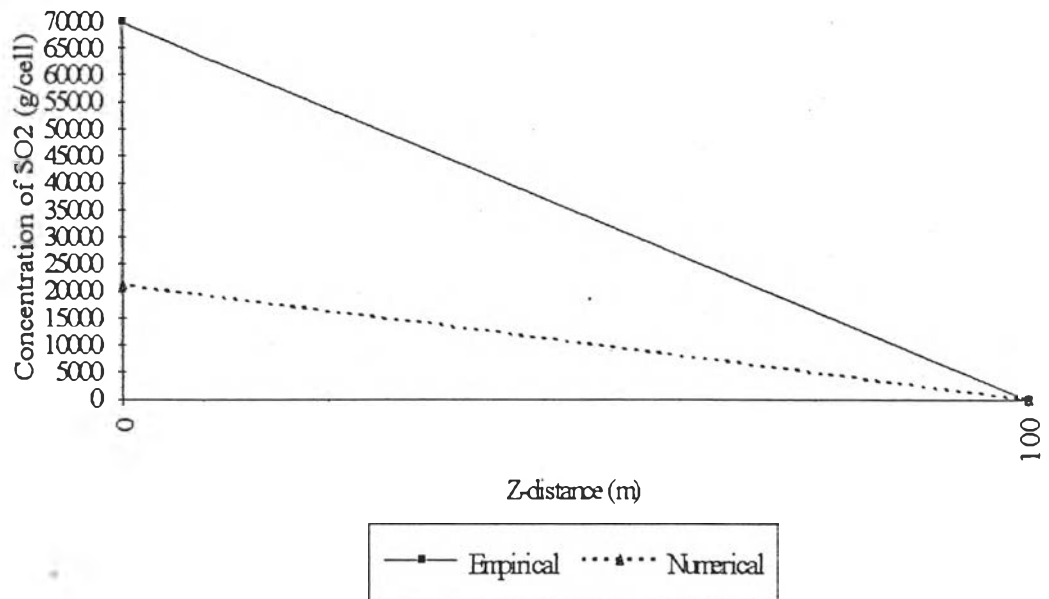


Figure 4.12 Comparison of the Empirical and Numerical Concentrations of SO₂ for Atmospheric Stability Class F at 1 km Downwind from the Source

a) Varying Y-Distance (Fixed z=0)

b) Varying Z-Distance (Fixed y=0)

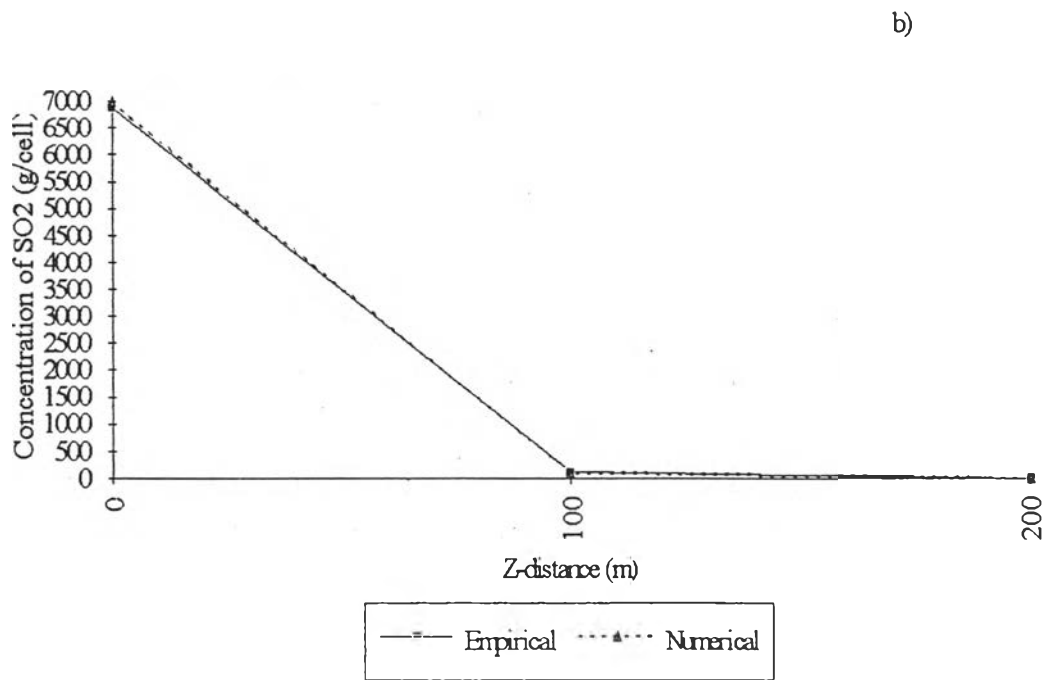
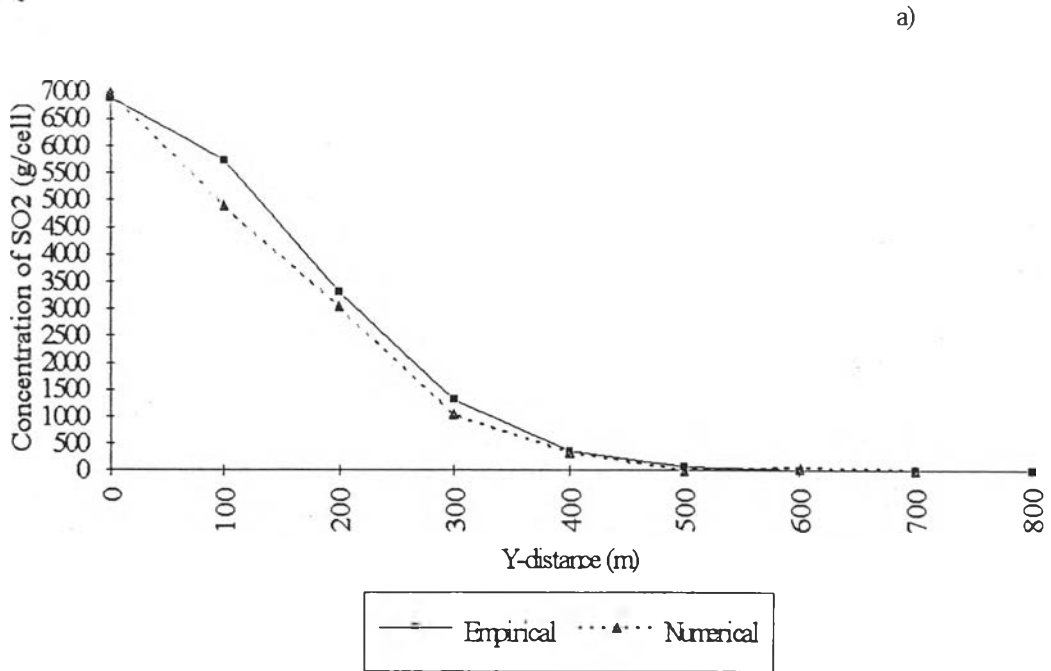
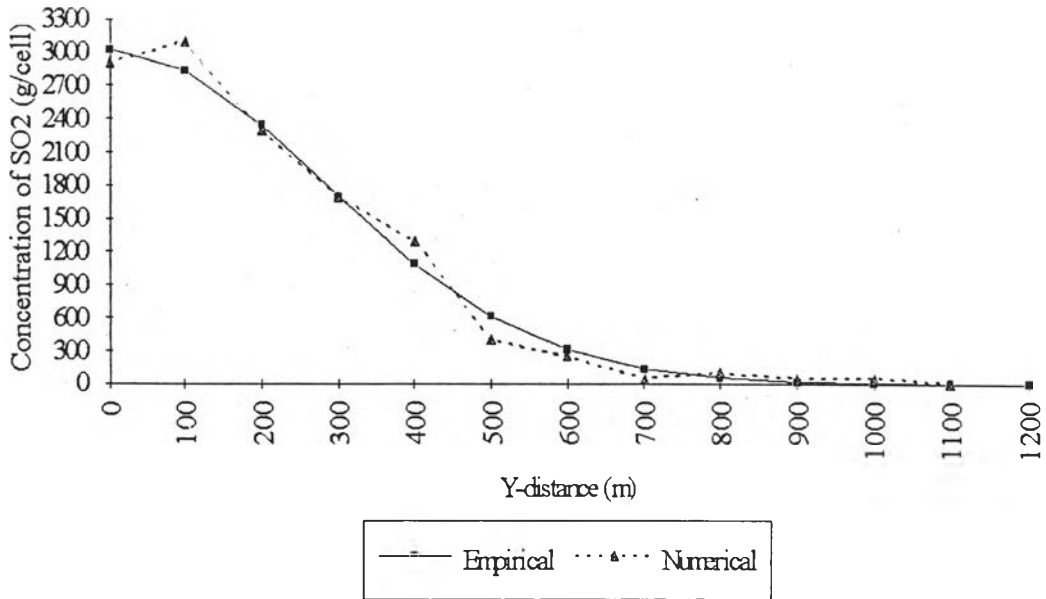


Figure 4.13 Comparison of the Empirical and Numerical Concentrations of SO₂ for Atmospheric Stability Class F at 5 km Downwind from the Source

a) Varying Y-Distance (Fixed z=0)

b) Varying Z-Distance (Fixed y=0)

a)



b)

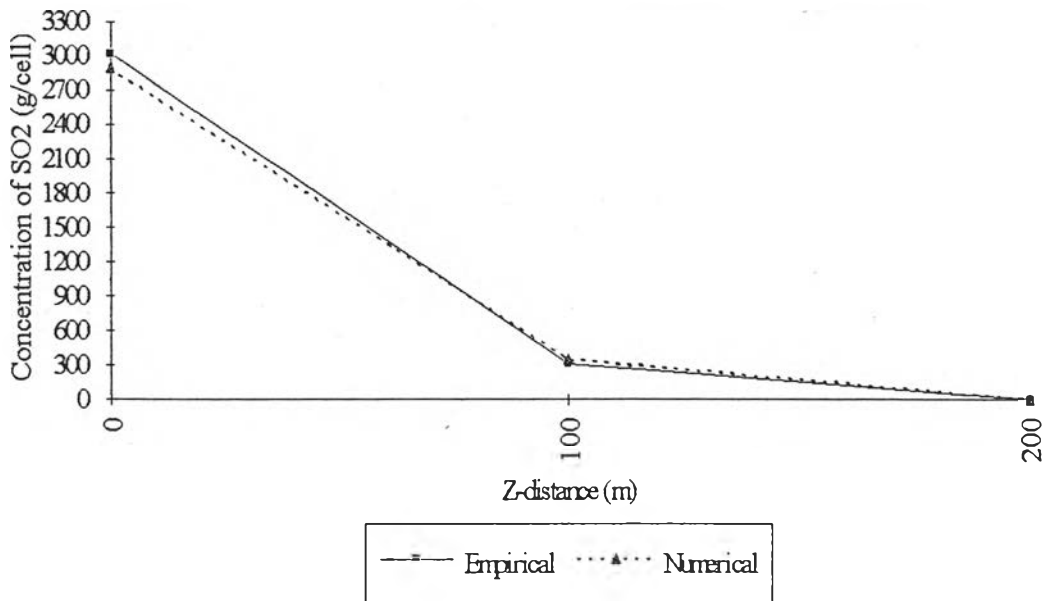


Figure 4.14 Comparison of the Empirical and Numerical Concentrations of SO₂ for Atmospheric Stability Class F at 10 km Downwind from the Source

a) Varying Y-Distance (Fixed z=0)

b) Varying Z-Distance (Fixed y=0)

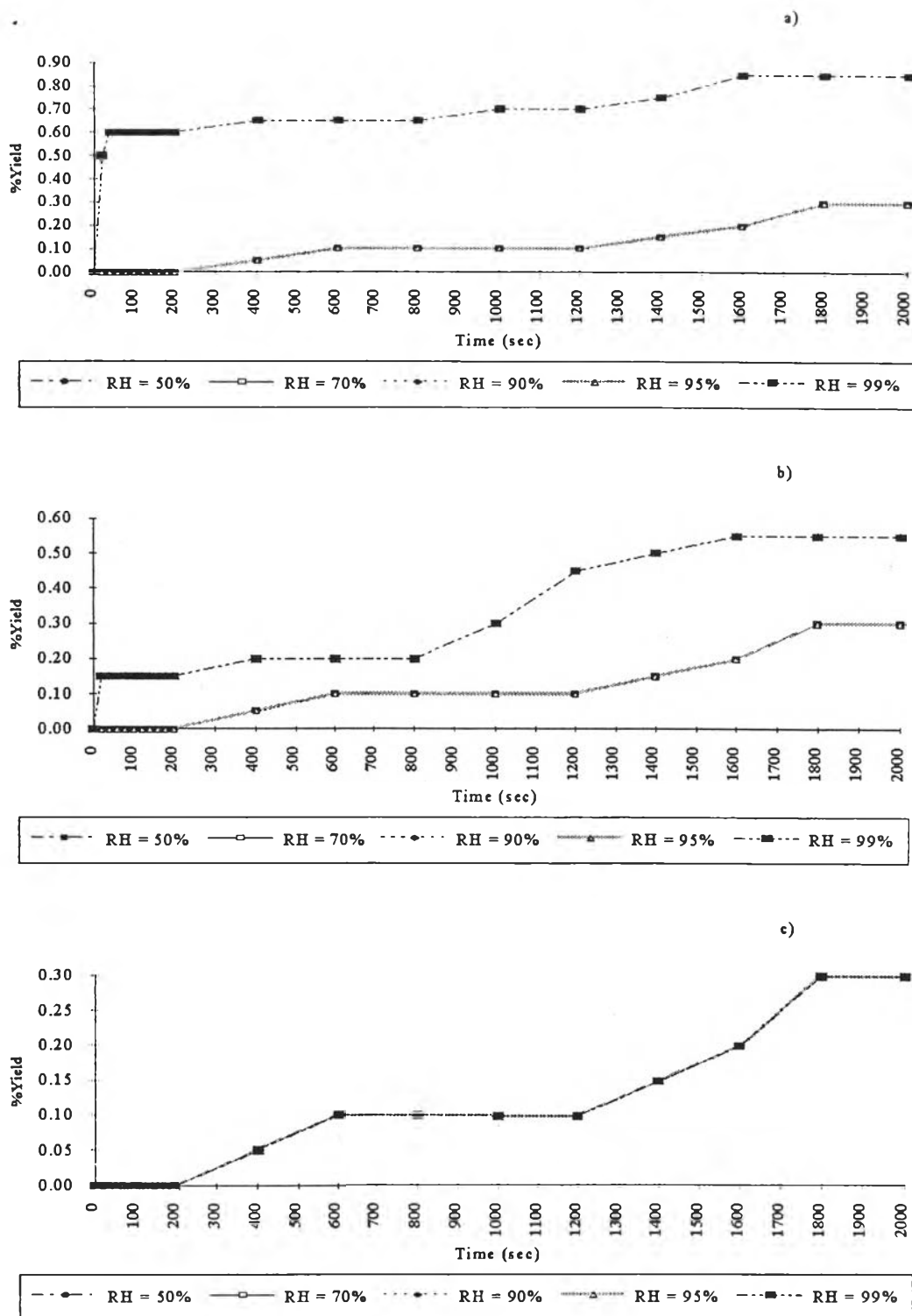


Figure 4.15 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class C at $[\text{Fe}] = 1201 \text{ ng/m}^3$ and $[\text{NH}_3] = 50 \text{ ppb}$

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

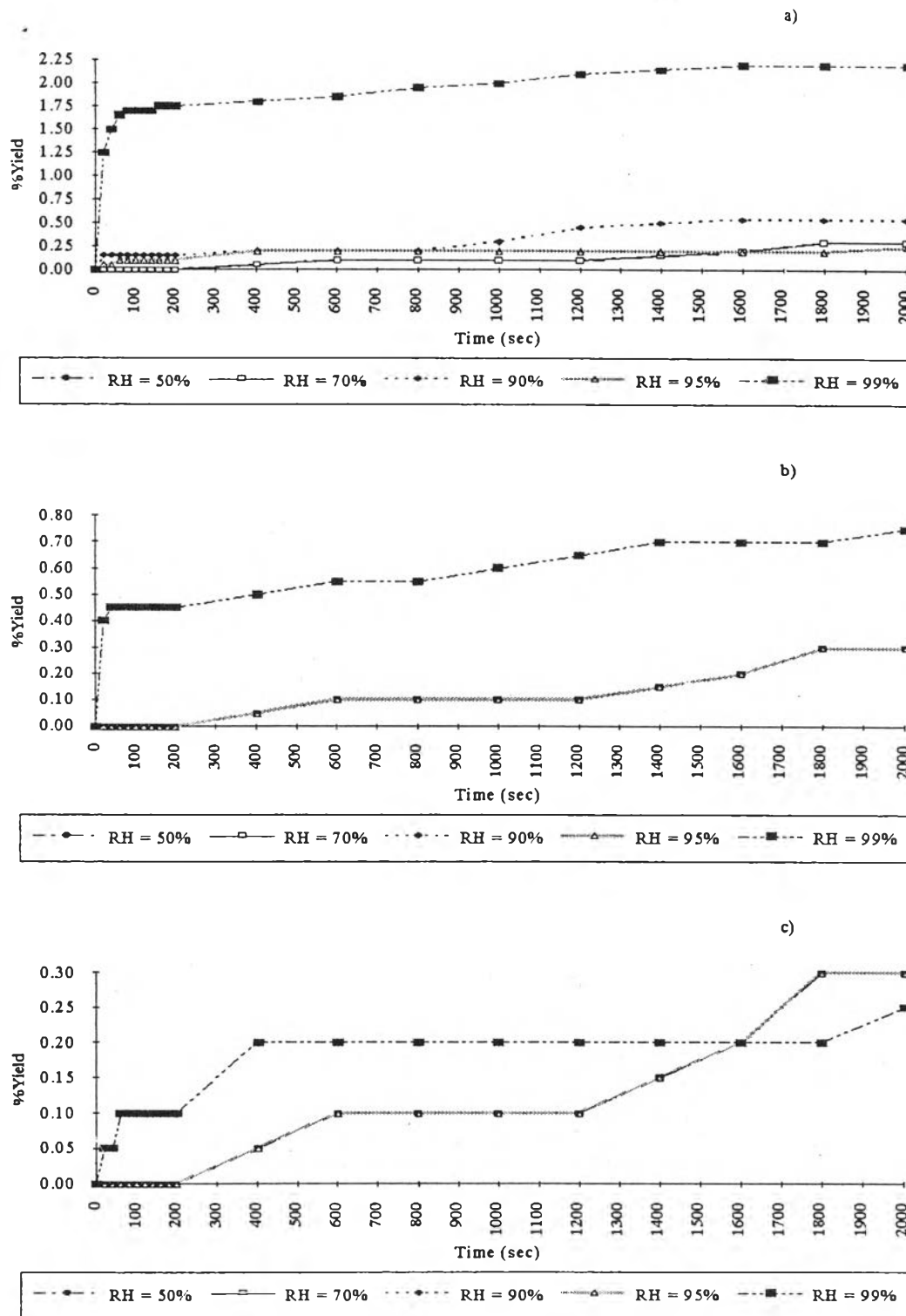


Figure 4.16 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class C at [Fe] = 1201 ng/m³ and [NH₃] = 80 ppb

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

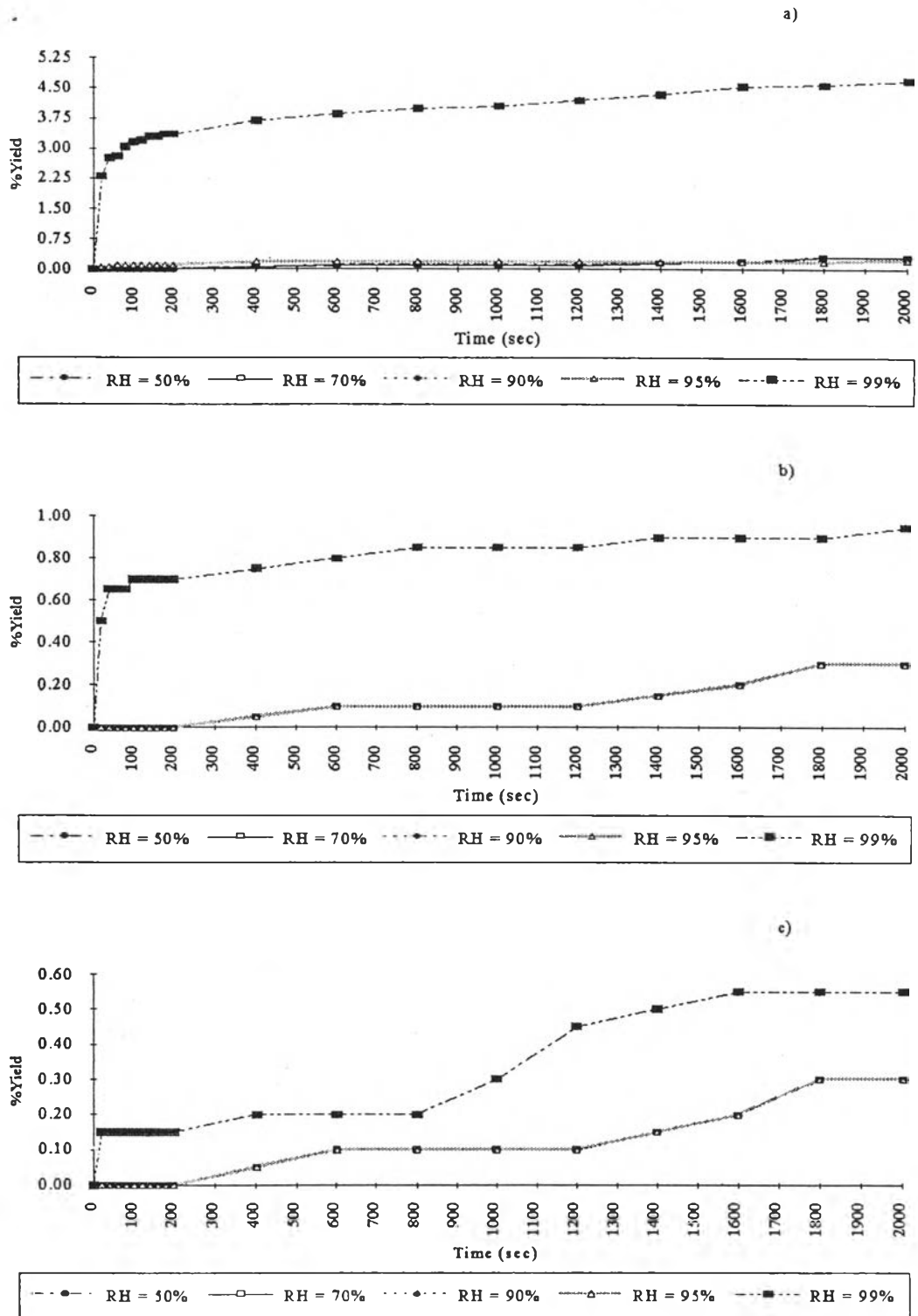


Figure 4.17 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class C at $[\text{Fe}] = 1201 \text{ ng/m}^3$ and $[\text{NH}_3] = 100 \text{ ppb}$

a) $T = 20 \text{ }^\circ\text{C}$

b) $T = 25 \text{ }^\circ\text{C}$

c) $T = 30 \text{ }^\circ\text{C}$

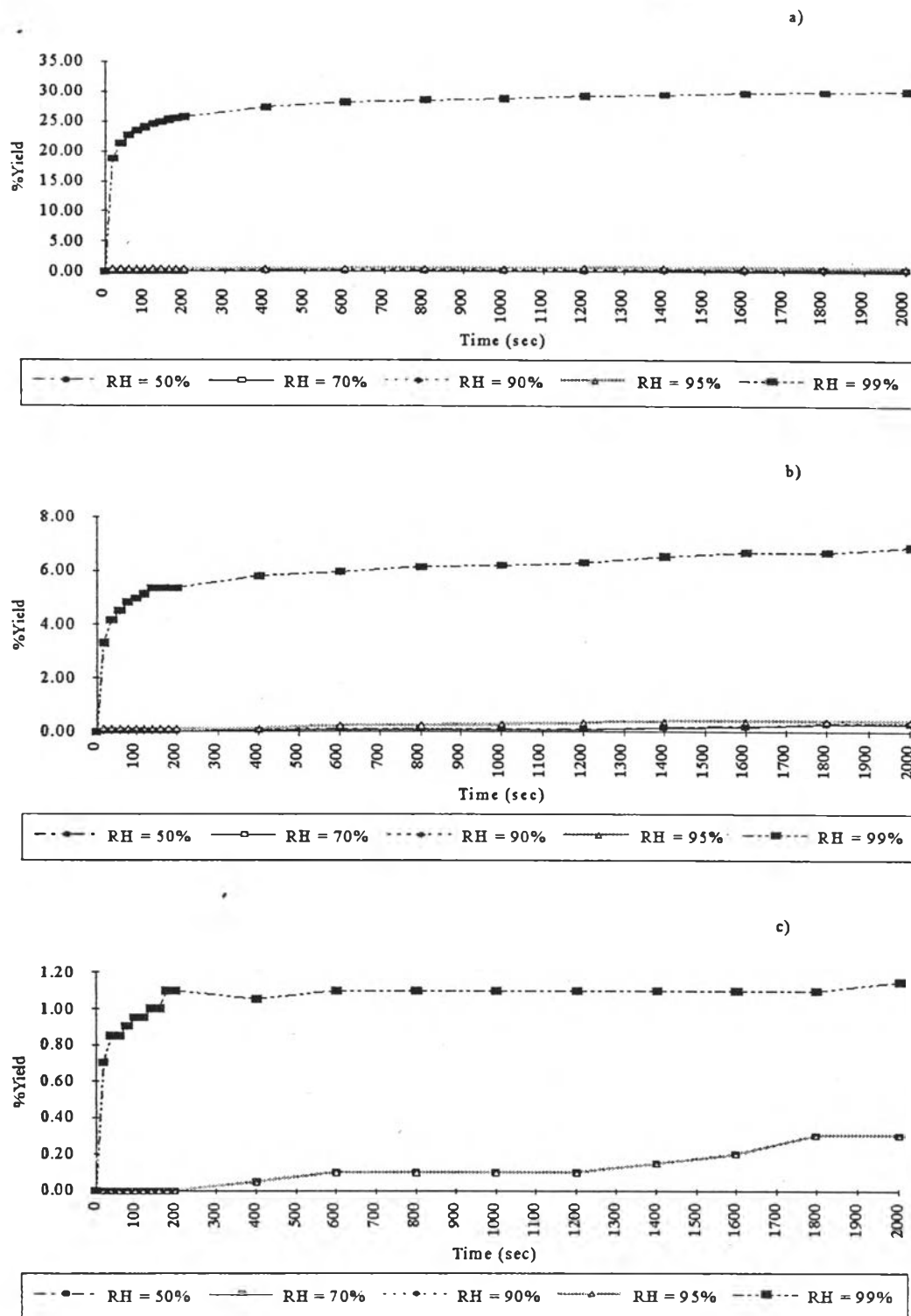


Figure 4.18 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class C at [Fe] = 0.1 mg/m³ and [NH₃] = 50 ppb

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

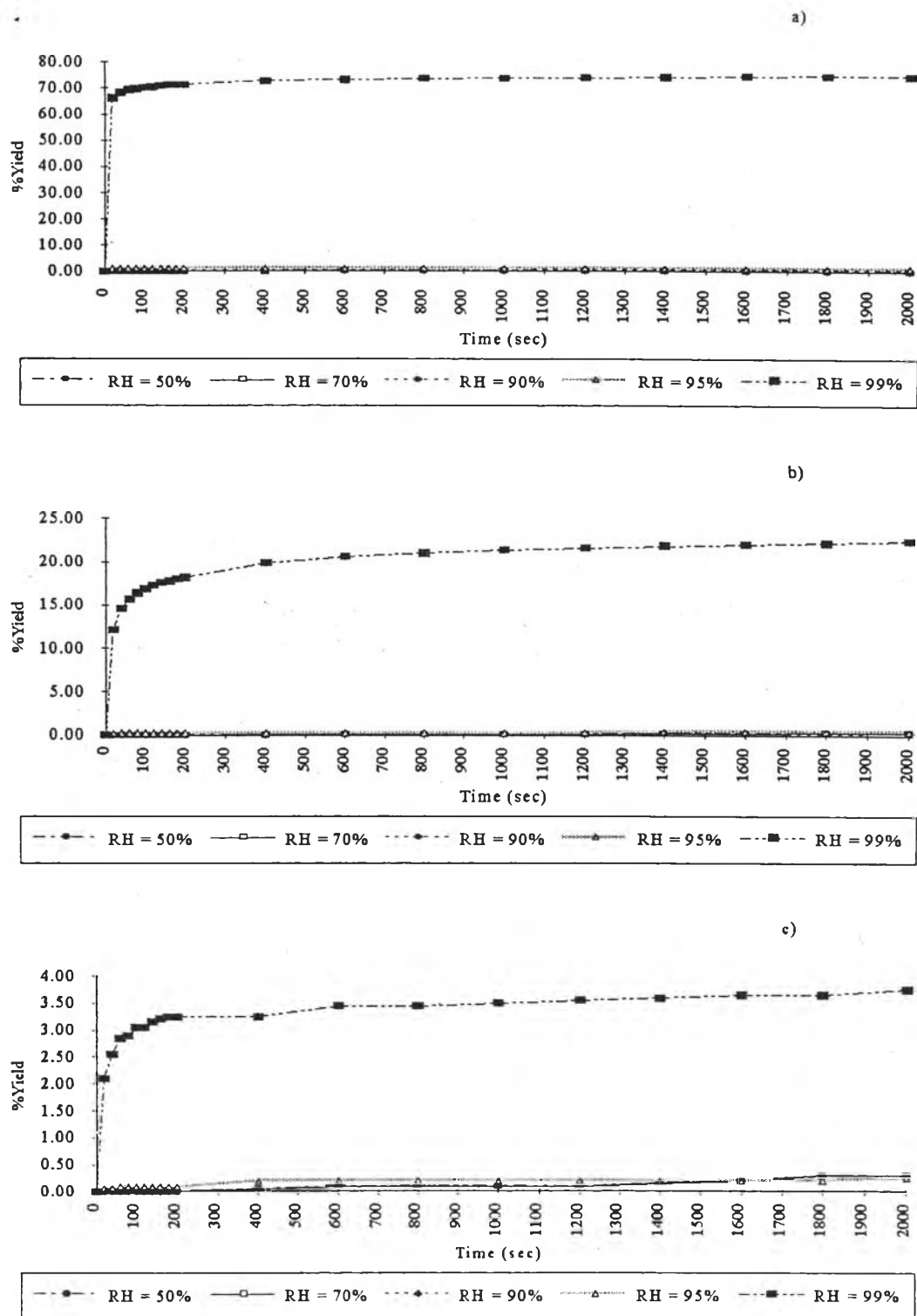


Figure 4.19 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class C at $[\text{Fe}] = 0.1 \text{ mg/m}^3$ and $[\text{NH}_3] = 80 \text{ ppb}$

a) $T = 20 \text{ }^\circ\text{C}$

b) $T = 25 \text{ }^\circ\text{C}$

c) $T = 30 \text{ }^\circ\text{C}$

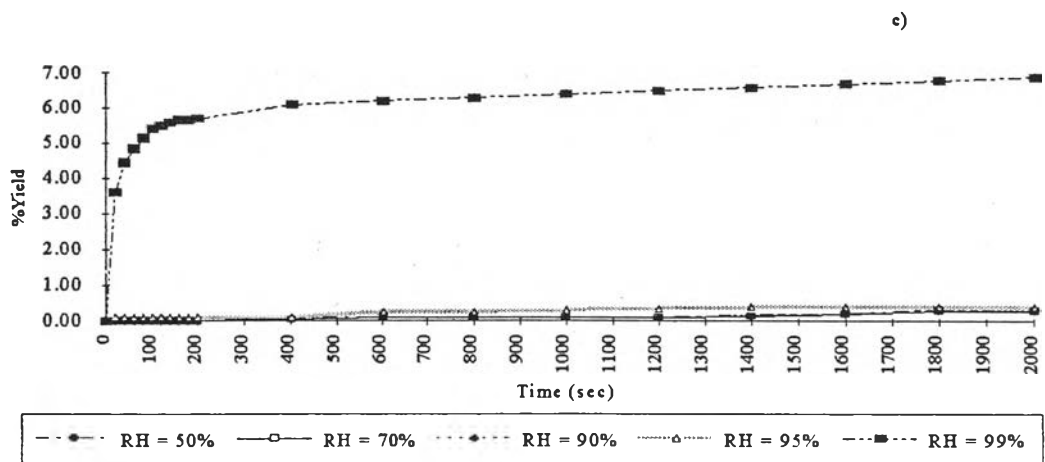
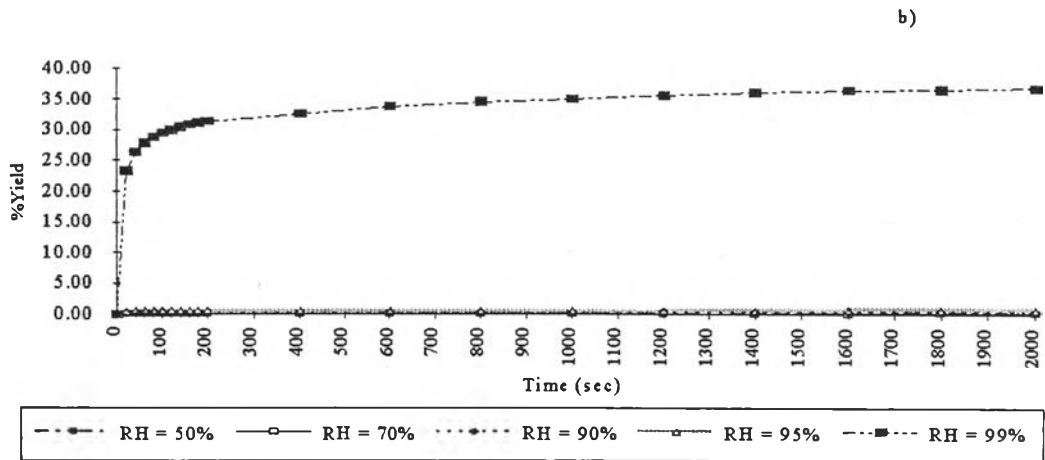
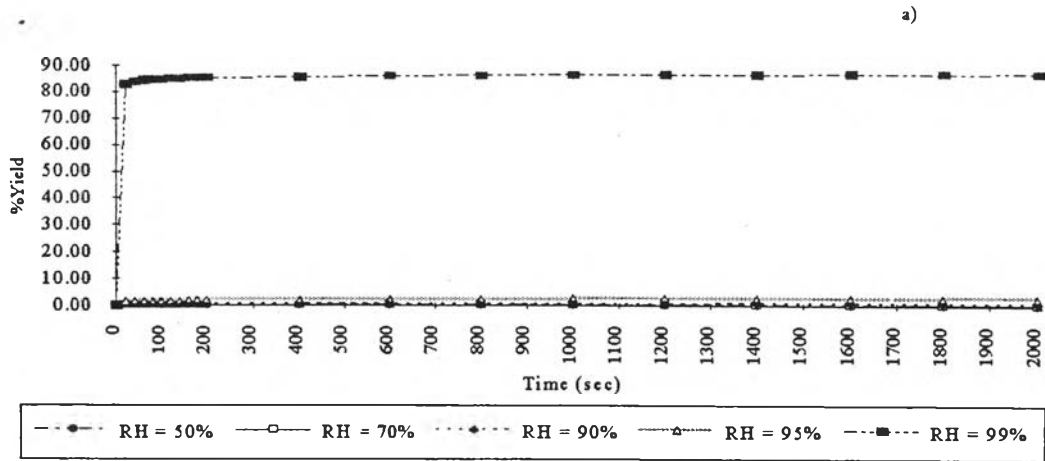


Figure 4.20 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class C at $[Fe] = 0.1 \text{ mg/m}^3$ and $[NH_3] = 100 \text{ ppb}$

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

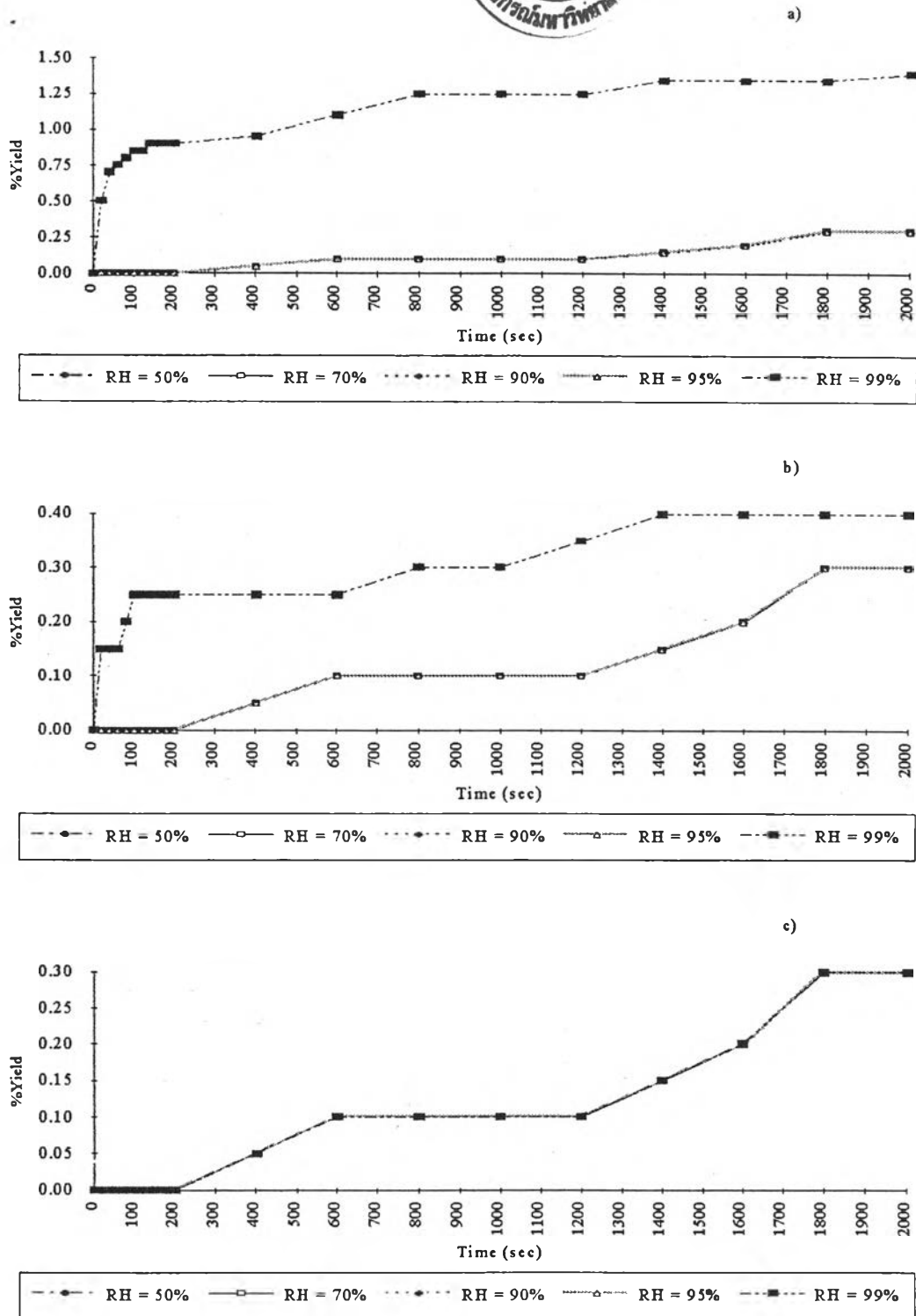


Figure 4.21 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class D at [Fe] = 1201 ng/m³ and [NH₃] = 50 ppb

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

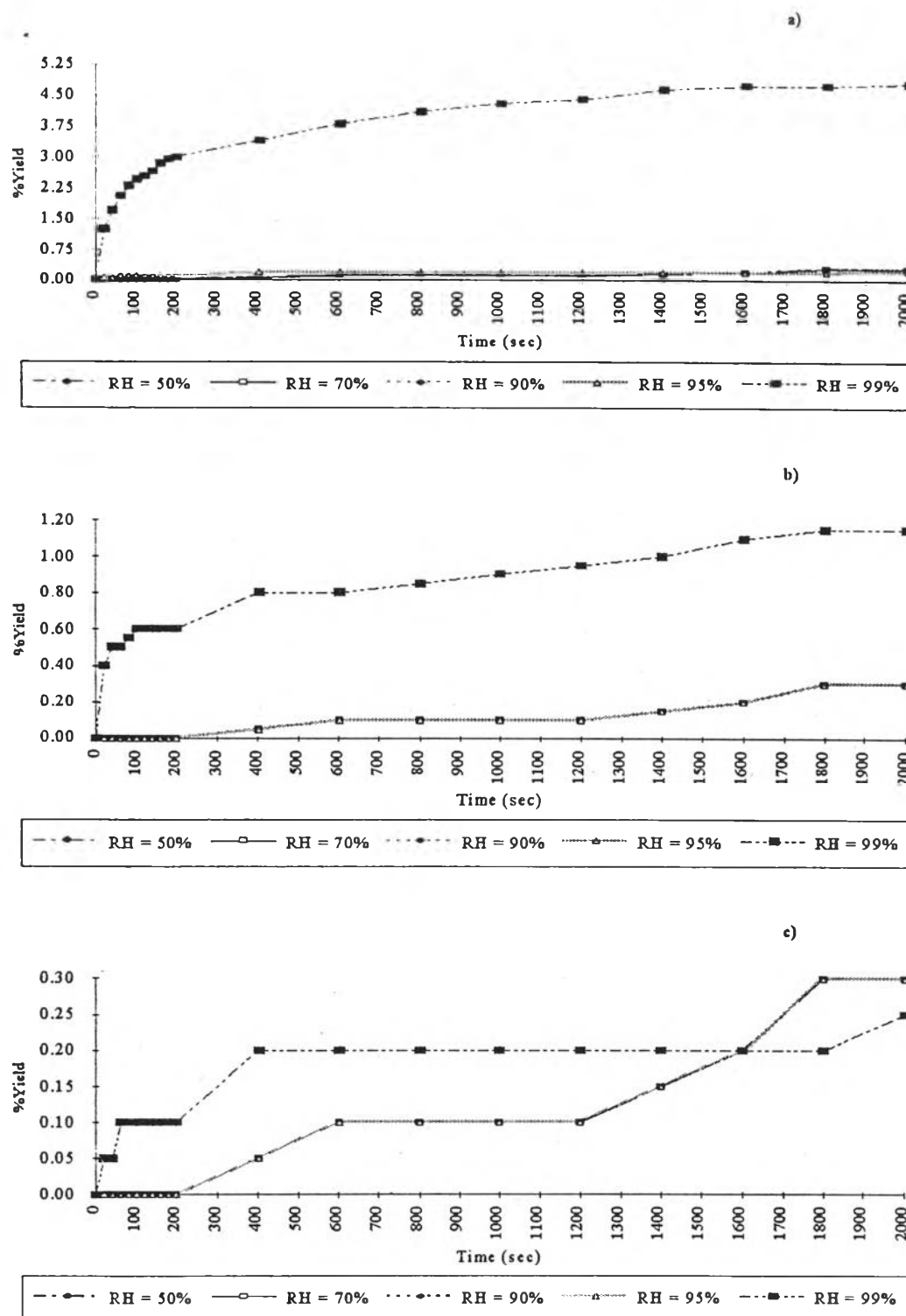


Figure 4.22 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class D at $[\text{Fe}] = 1201 \text{ ng/m}^3$ and $[\text{NH}_3] = 80 \text{ ppb}$

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

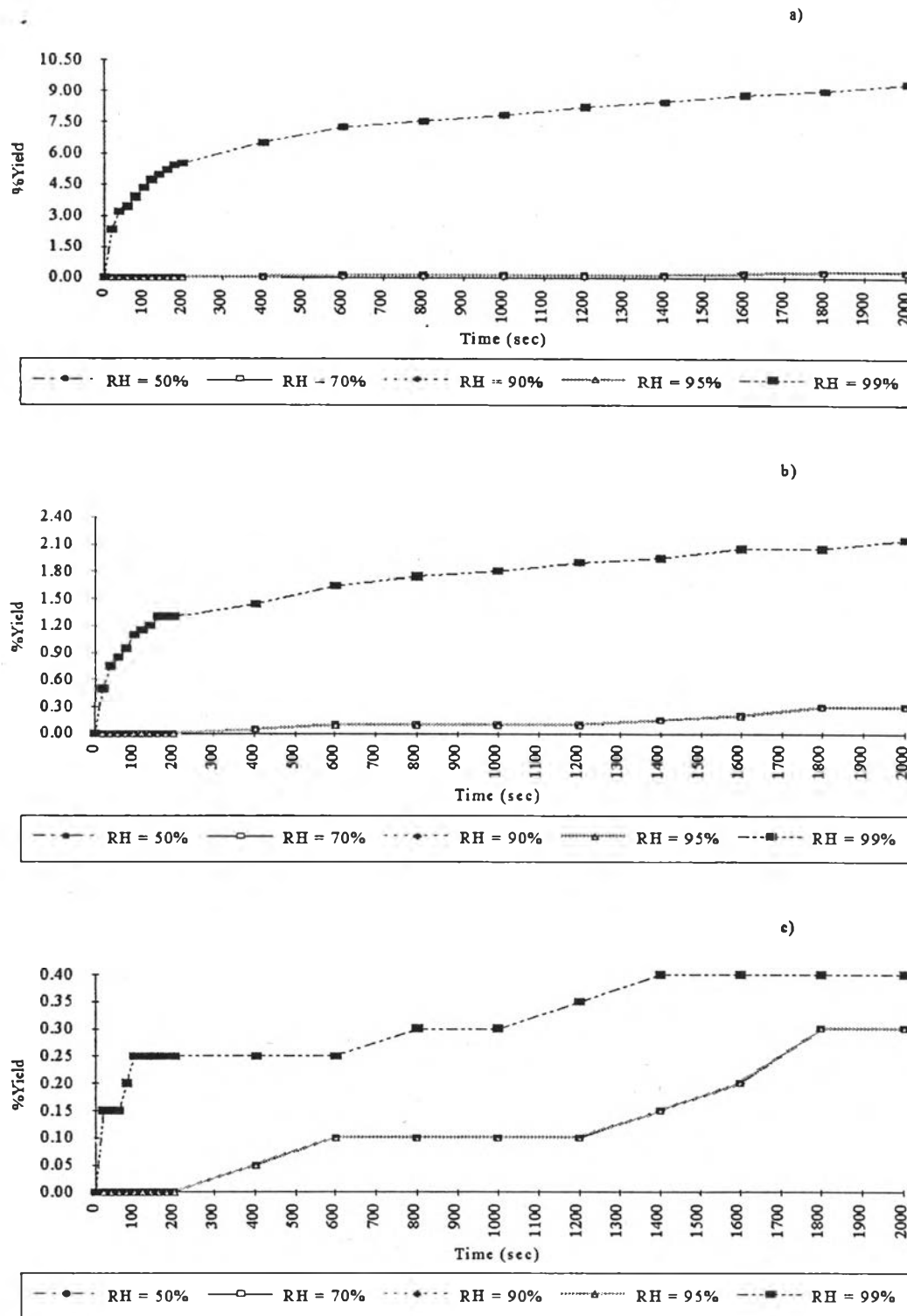


Figure 4.23 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class D at $[Fe] = 1201 \text{ ng/m}^3$ and $[NH_3] = 100 \text{ ppb}$

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

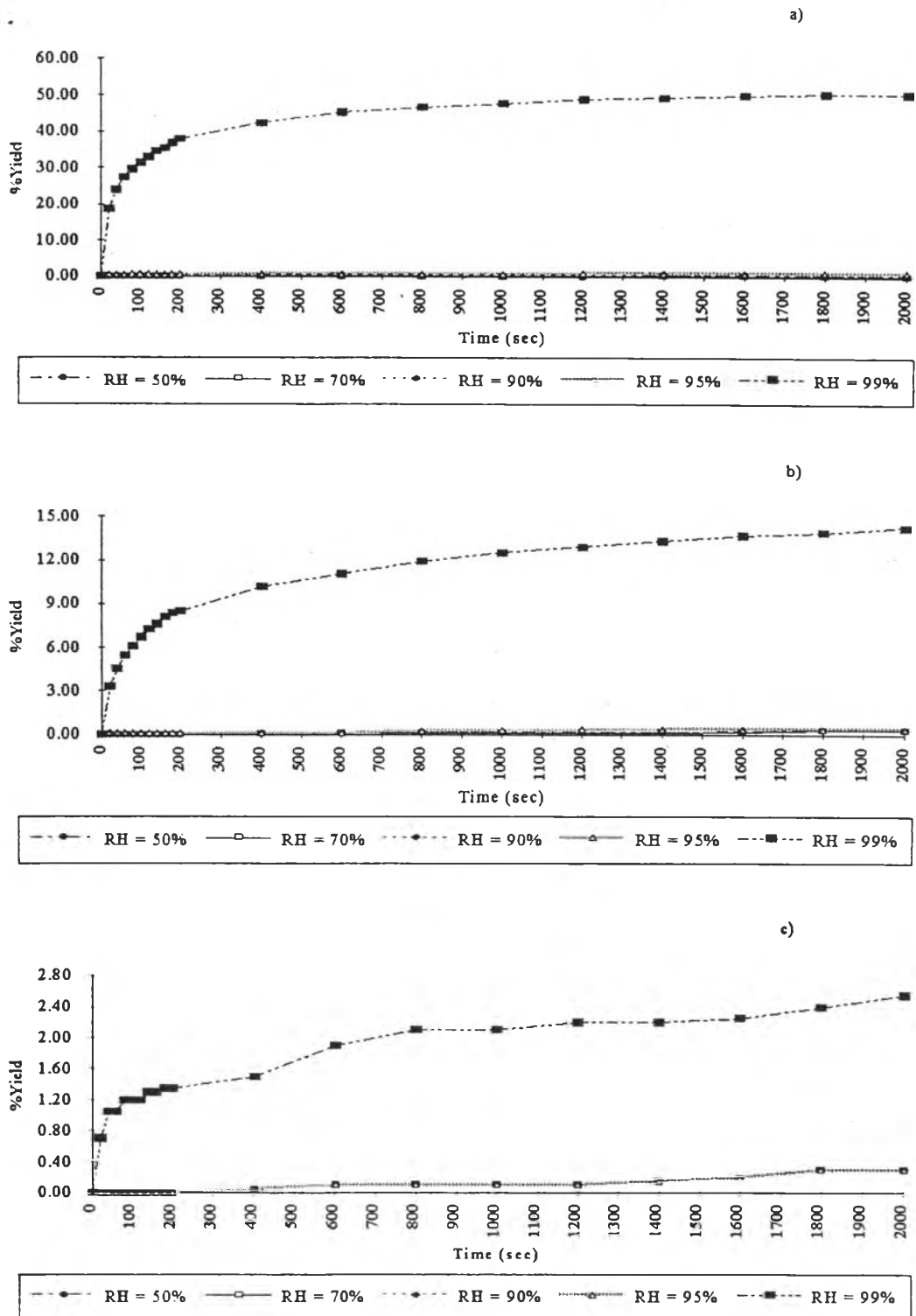


Figure 4.24 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class D at $[\text{Fe}] = 0.1 \text{ mg/m}^3$ and $[\text{NH}_3] = 50 \text{ ppb}$

a) $T = 20 \text{ }^\circ\text{C}$

b) $T = 25 \text{ }^\circ\text{C}$

c) $T = 30 \text{ }^\circ\text{C}$

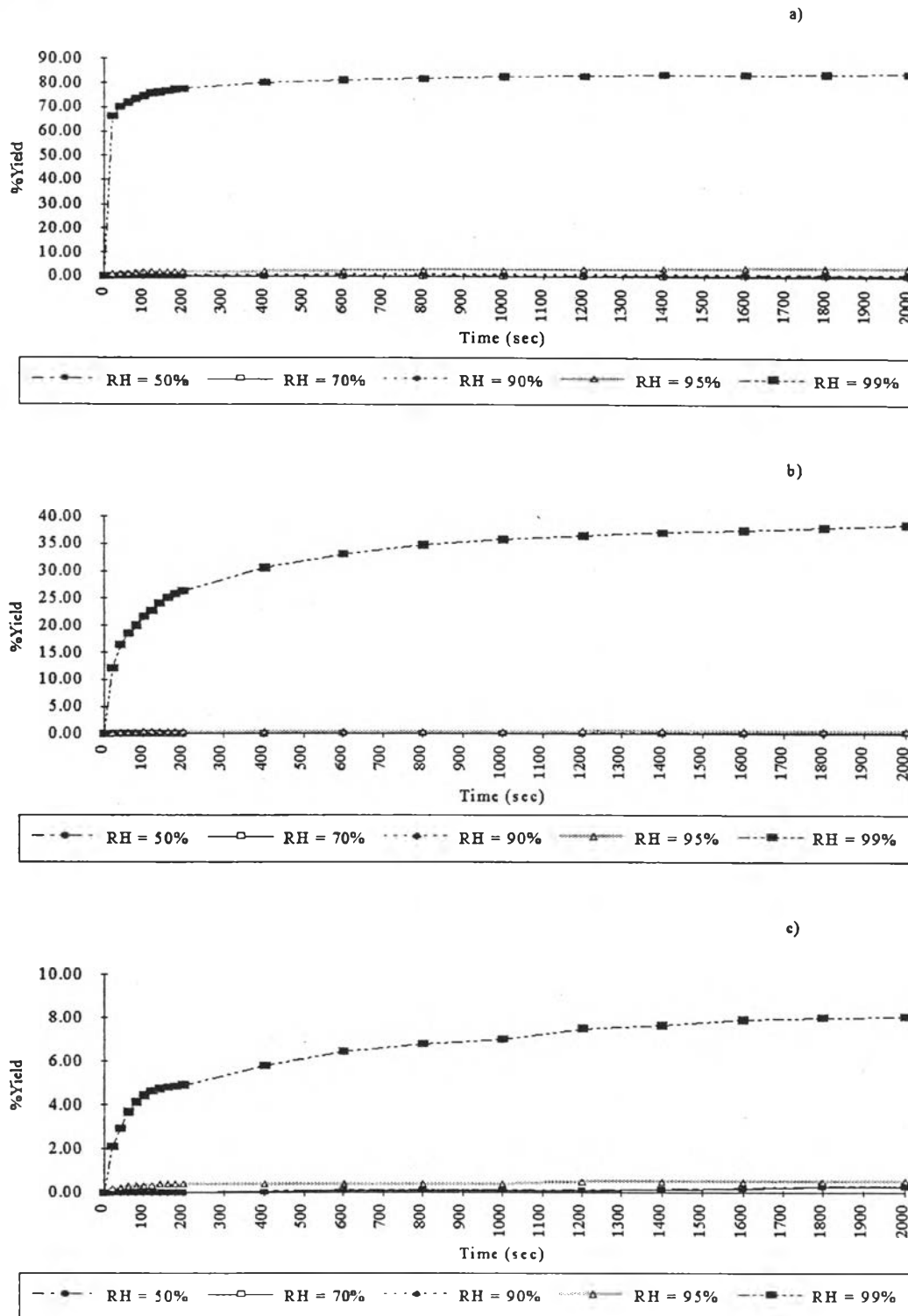


Figure 4.25 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class D at $[Fe] = 0.1 \text{ mg/m}^3$ and $[NH_3] = 80 \text{ ppb}$

a) $T = 20 \text{ }^\circ\text{C}$

b) $T = 25 \text{ }^\circ\text{C}$

c) $T = 30 \text{ }^\circ\text{C}$

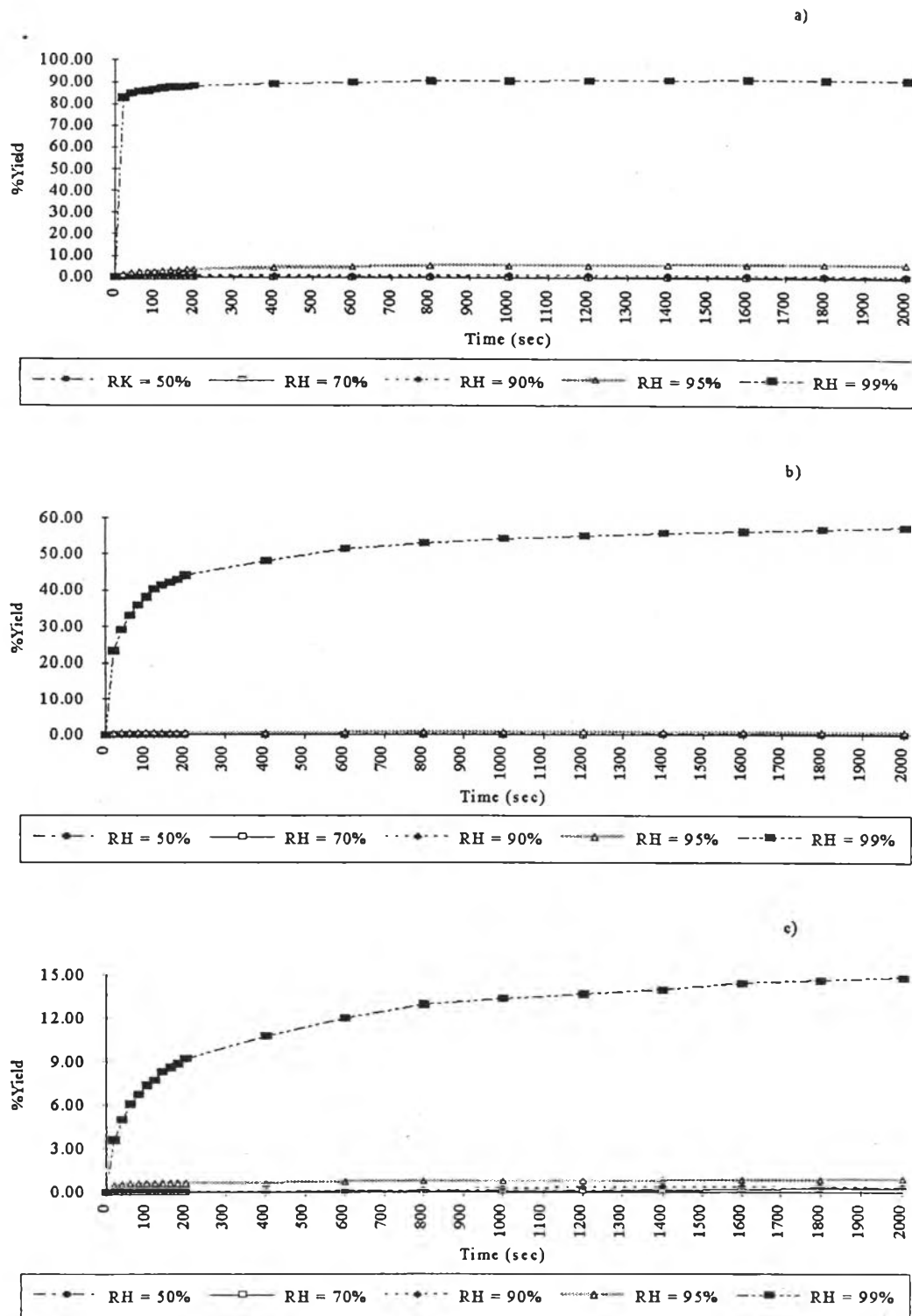


Figure 4.26 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class D at $[Fe] = 0.1 \text{ mg/m}^3$ and $[NH_3] = 100 \text{ ppb}$

a) $T = 20 \text{ }^\circ\text{C}$

b) $T = 25 \text{ }^\circ\text{C}$

c) $T = 30 \text{ }^\circ\text{C}$

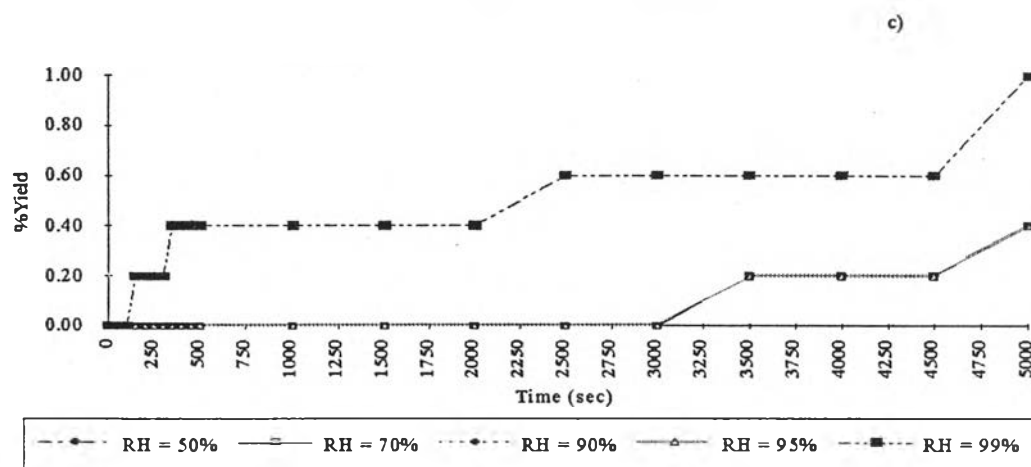
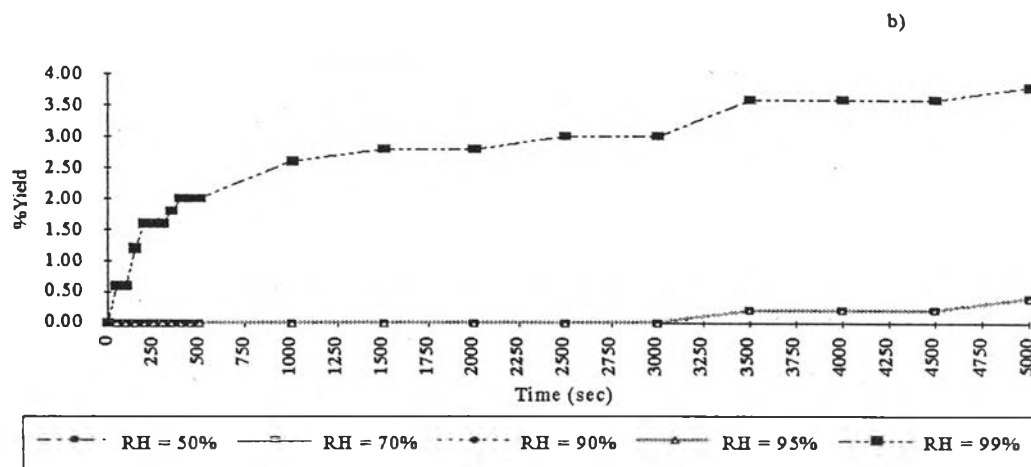
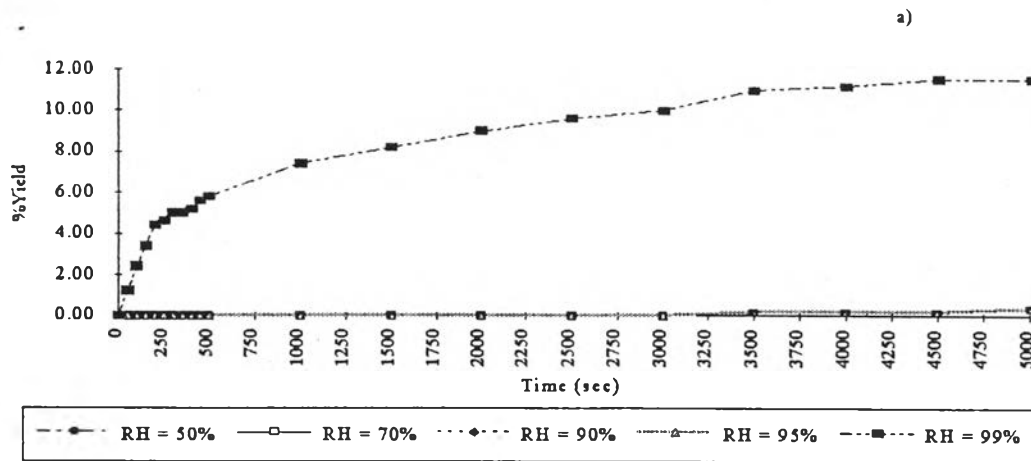


Figure 4.27 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class E at $[Fe] = 1201 \text{ ng/m}^3$ and $[NH_3] = 50 \text{ ppb}$

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

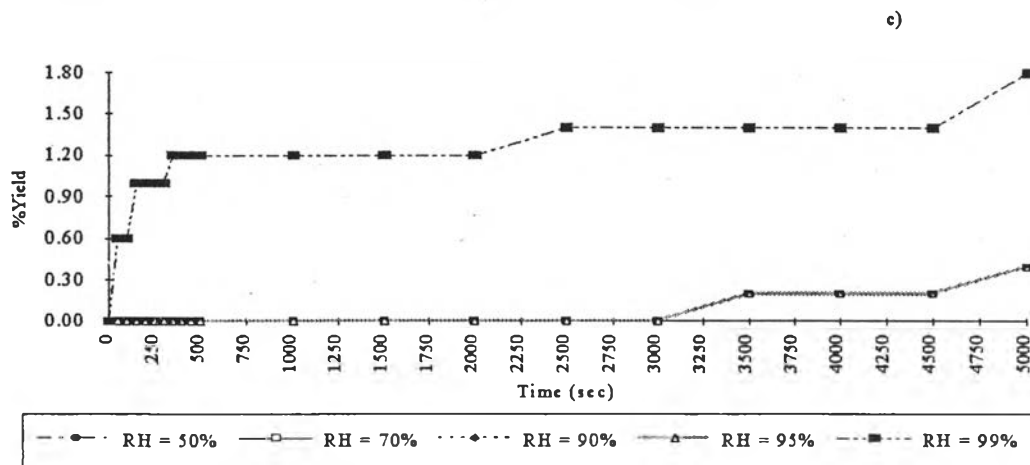
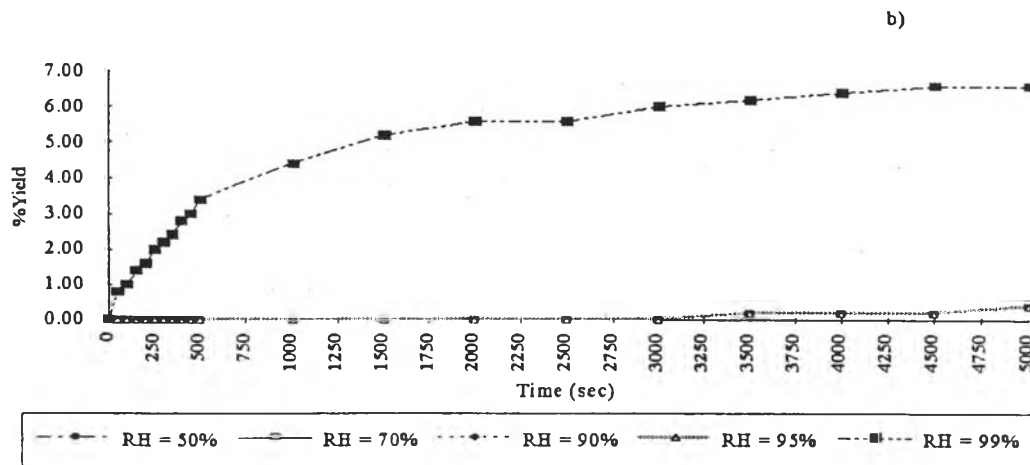
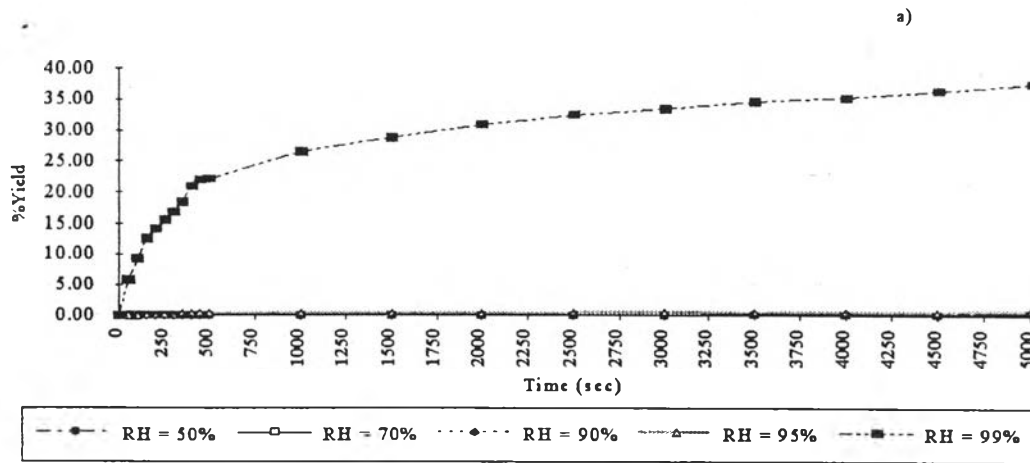


Figure 4.28 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class E at $[Fe] = 1201 \text{ ng/m}^3$ and $[NH_3] = 80 \text{ ppb}$

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

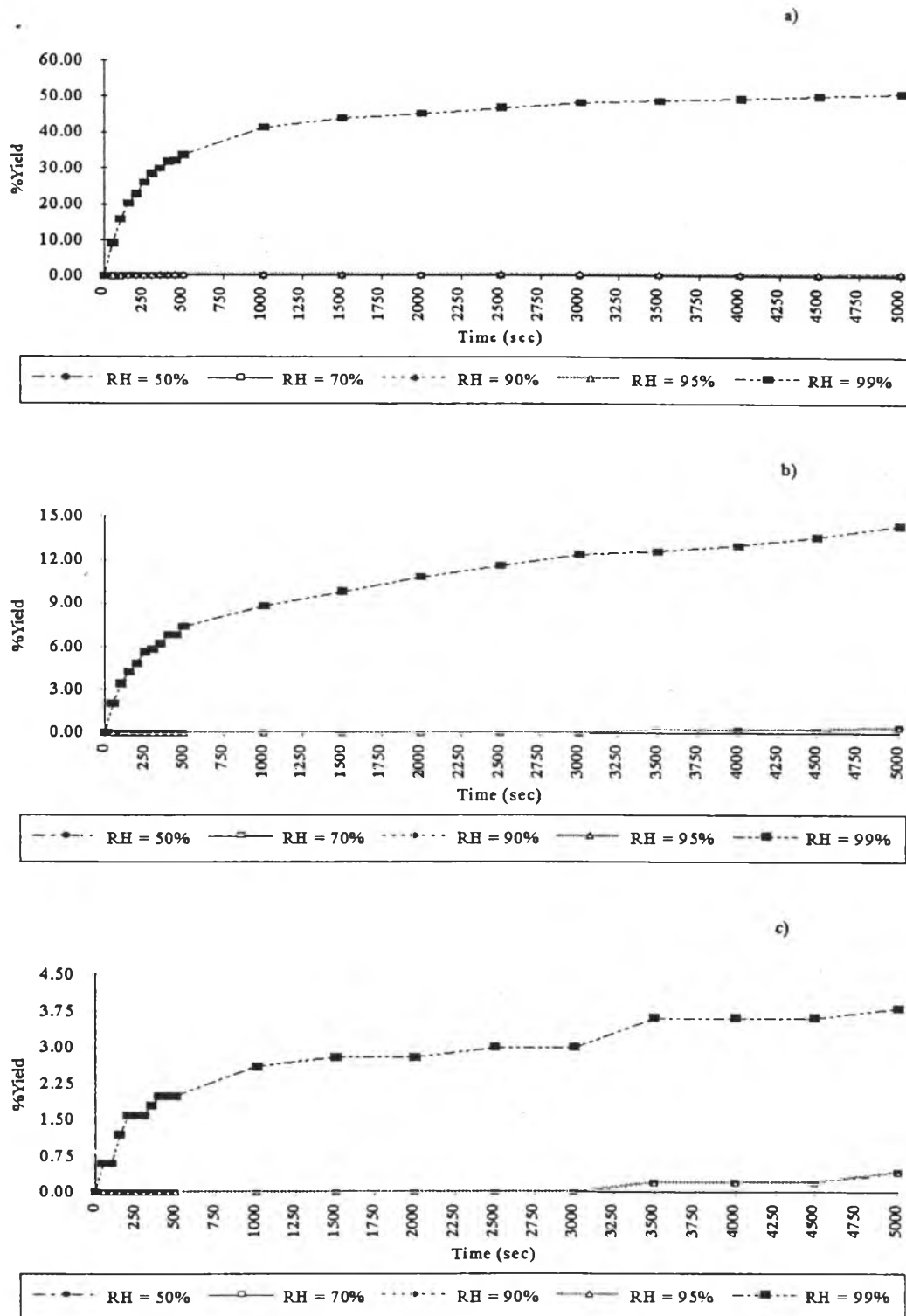


Figure 4.29 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class E at $[Fe] = 1201 \text{ ng/m}^3$ and $[NH_3] = 100 \text{ ppb}$

a) $T = 20 \text{ }^\circ\text{C}$

b) $T = 25 \text{ }^\circ\text{C}$

c) $T = 30 \text{ }^\circ\text{C}$

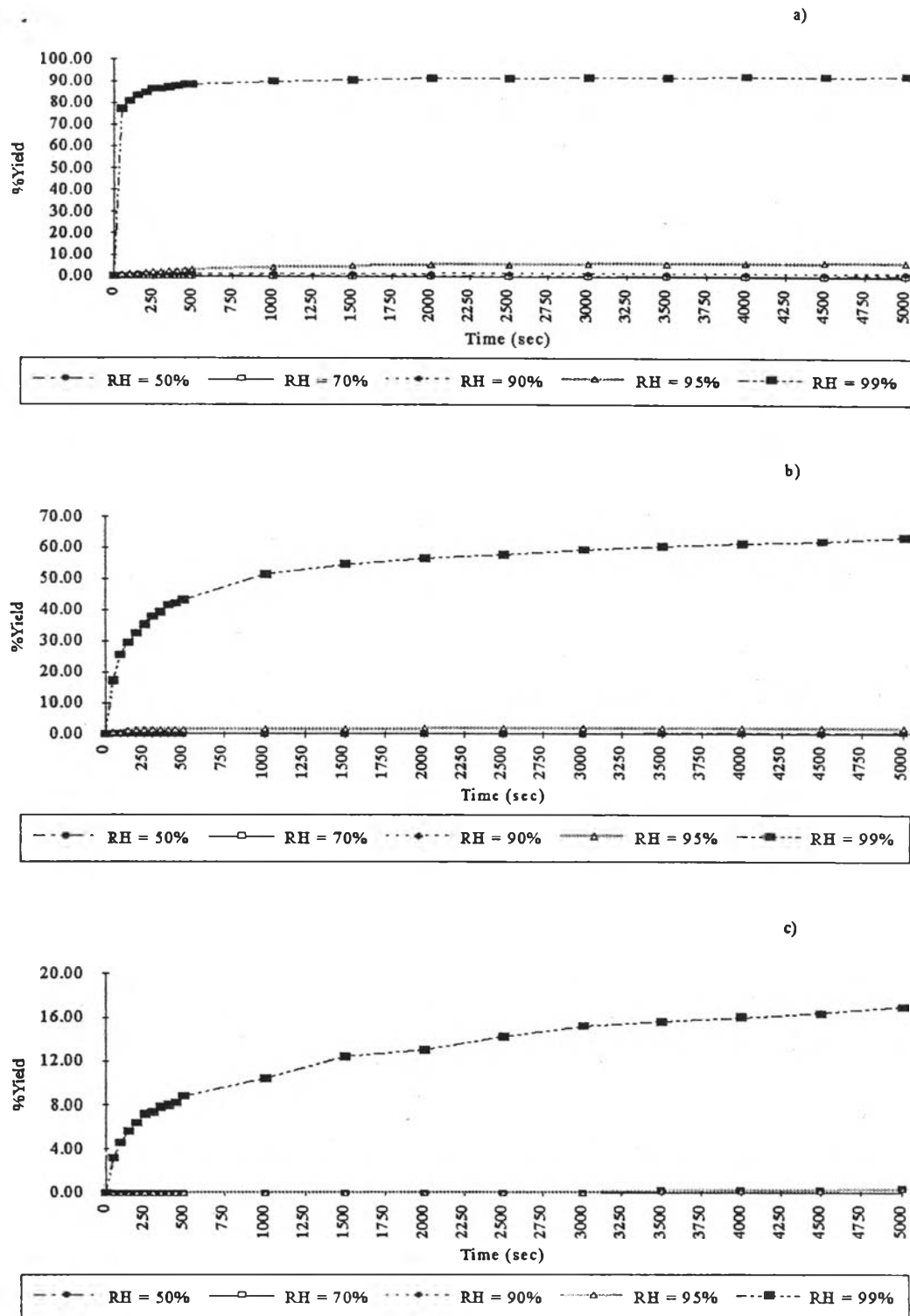


Figure 4.30 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class E at $[\text{Fe}] = 0.1 \text{ mg/m}^3$ and $[\text{NH}_3] = 50 \text{ ppb}$

a) $T = 20 \text{ }^\circ\text{C}$

b) $T = 25 \text{ }^\circ\text{C}$

c) $T = 30 \text{ }^\circ\text{C}$

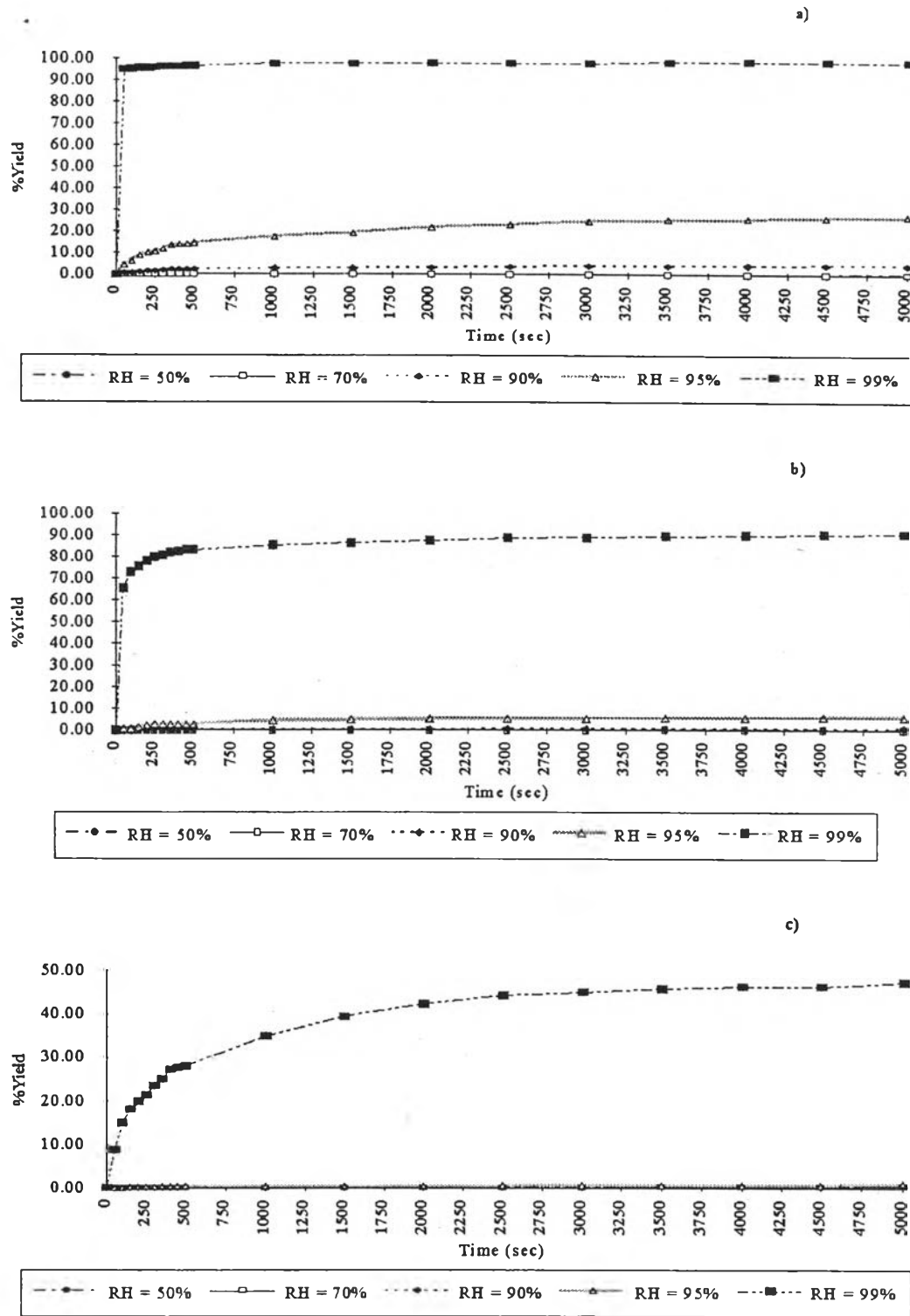


Figure 4.31 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class E at $[\text{Fe}] = 0.1 \text{ mg/m}^3$ and $[\text{NH}_3] = 80 \text{ ppb}$

a) $T = 20 \text{ }^\circ\text{C}$

b) $T = 25 \text{ }^\circ\text{C}$

c) $T = 30 \text{ }^\circ\text{C}$

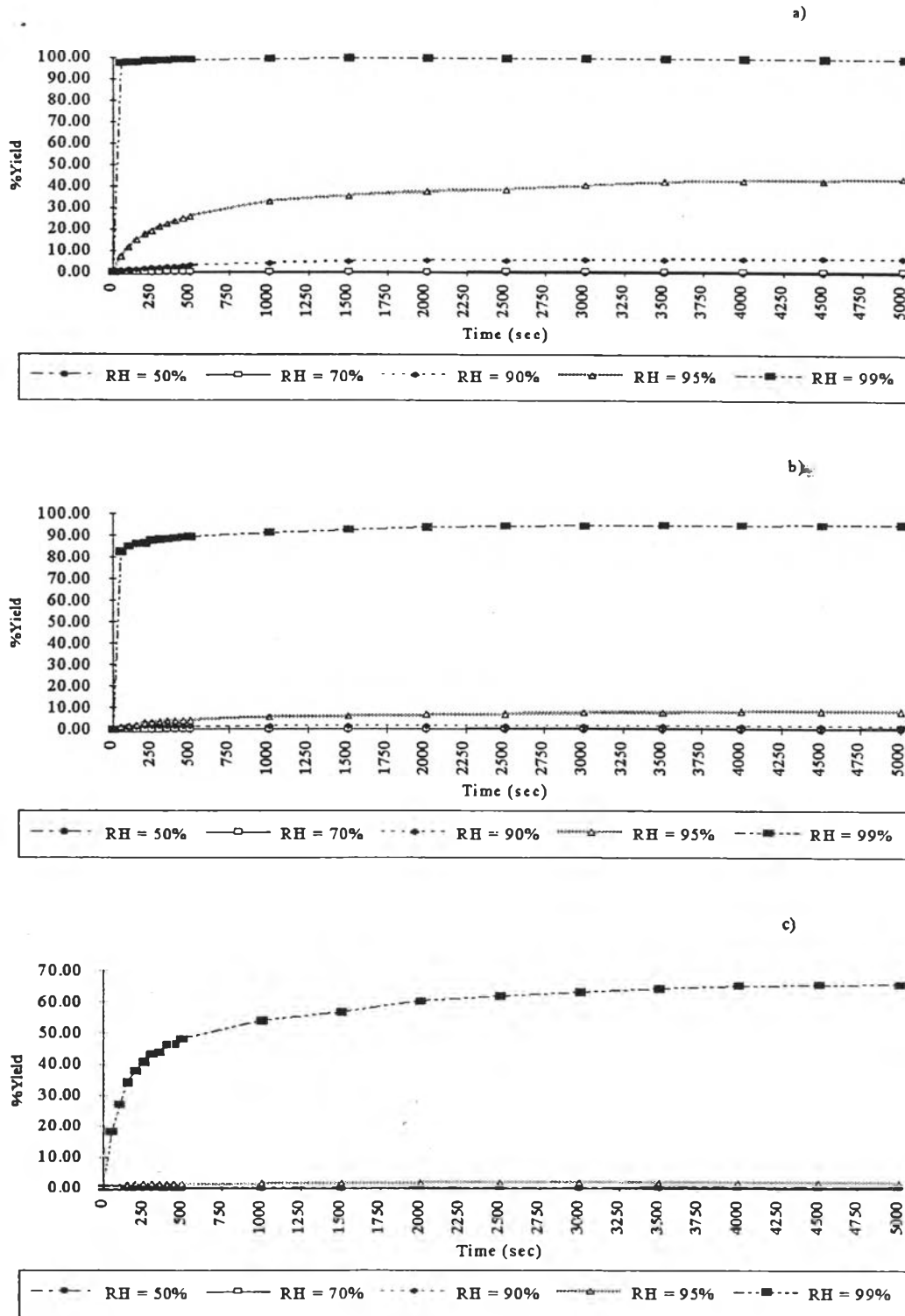


Figure 4.32 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class E at [Fe] = 0.1 mg/m³ and [NH₃] = 100 ppb

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

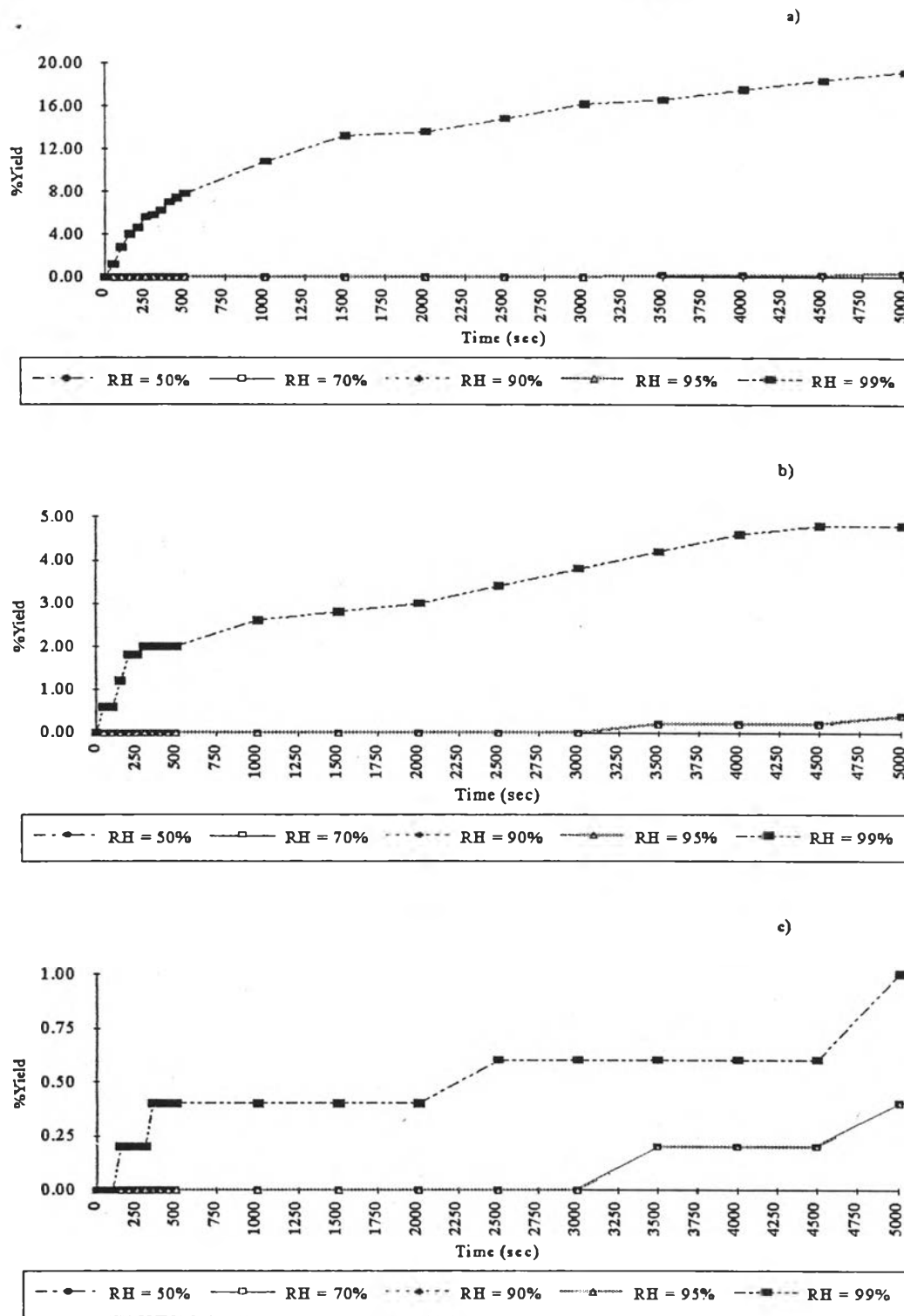


Figure 4.33 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class F at [Fe] = 1201 ng/m³ and [NH₃] = 50 ppb

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

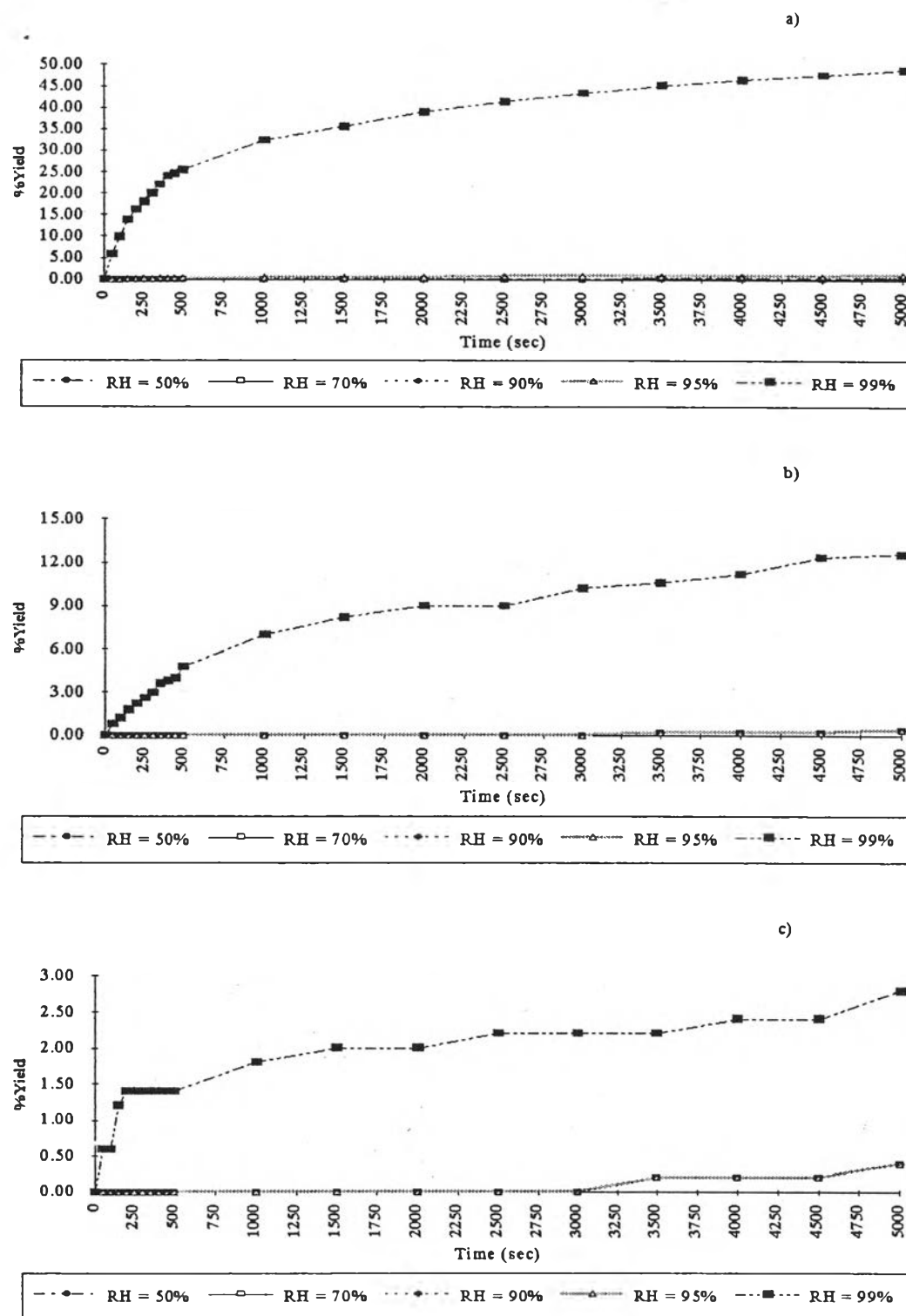


Figure 4.34 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class F at $[Fe] = 1201 \text{ ng/m}^3$ and $[NH_3] = 80 \text{ ppb}$

a) $T = 20 \text{ }^\circ\text{C}$

b) $T = 25 \text{ }^\circ\text{C}$

c) $T = 30 \text{ }^\circ\text{C}$

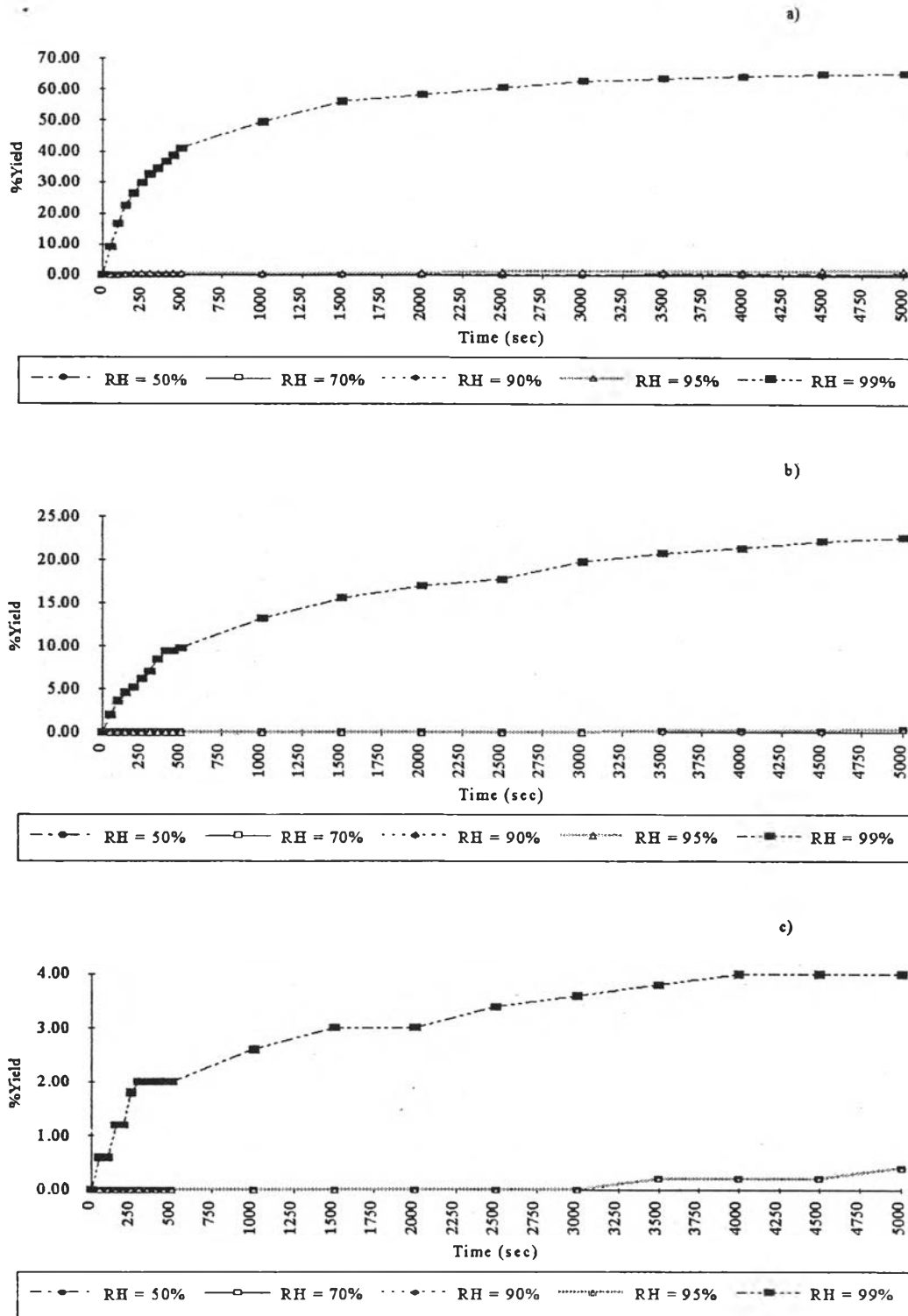


Figure 4.35 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class F at [Fe] = 1201 ng/m³ and [NH₃] = 100 ppb

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

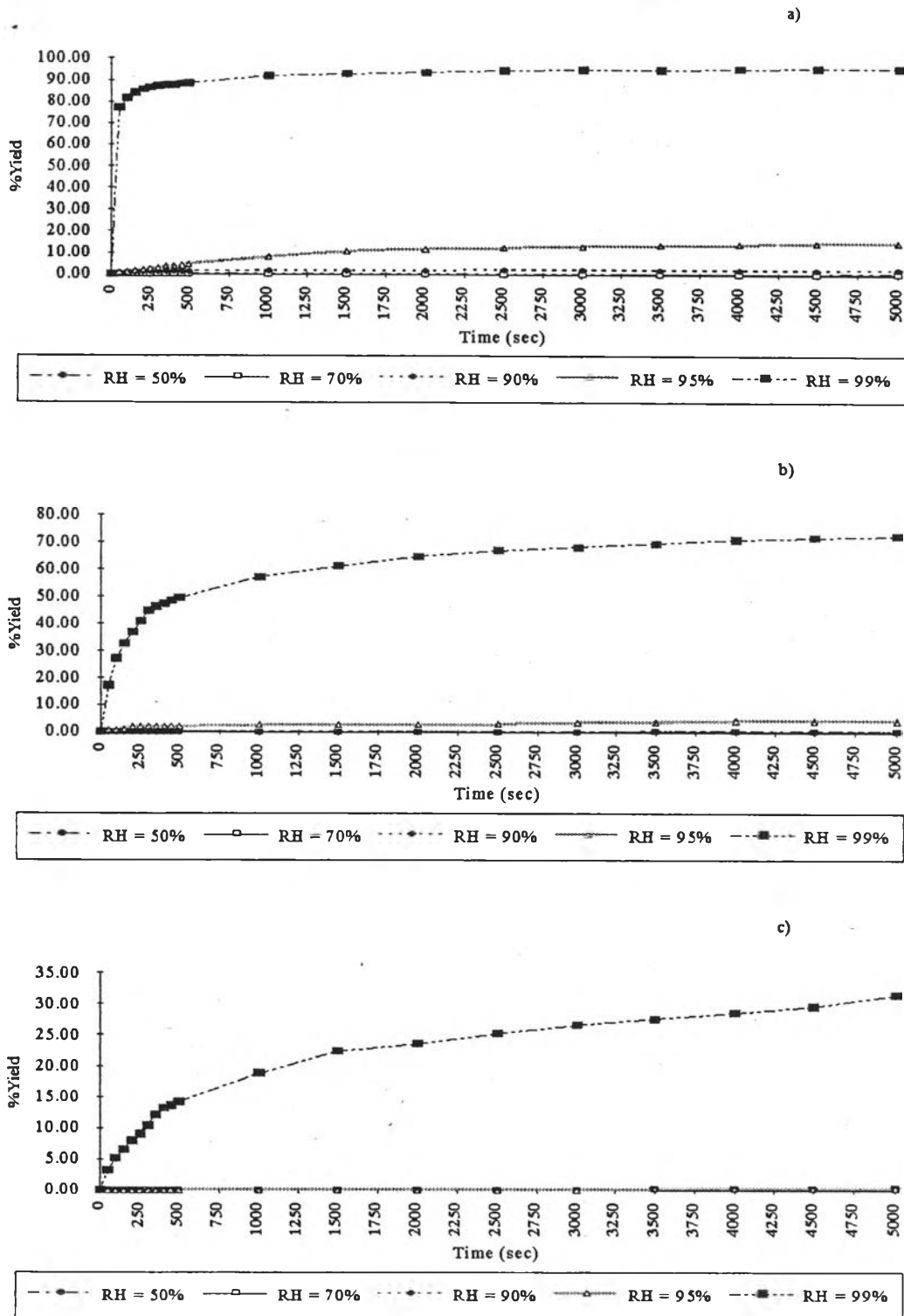


Figure 4.36 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class F at [Fe] = 0.1 mg/m³ and [NH₃] = 50 ppb

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

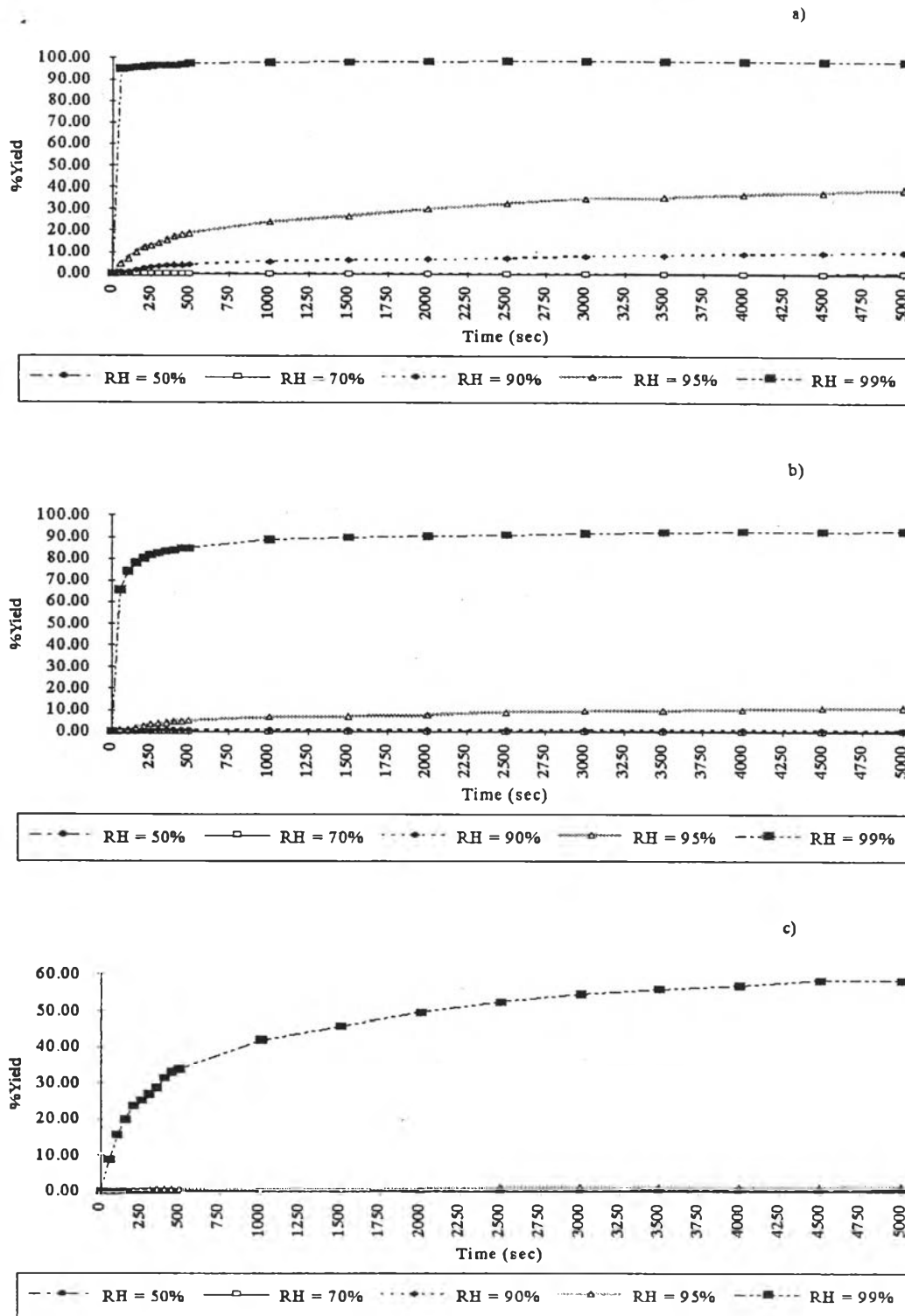


Figure 4.37 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class F at [Fe] = 0.1 mg/m³ and [NH₃] = 80 ppb

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

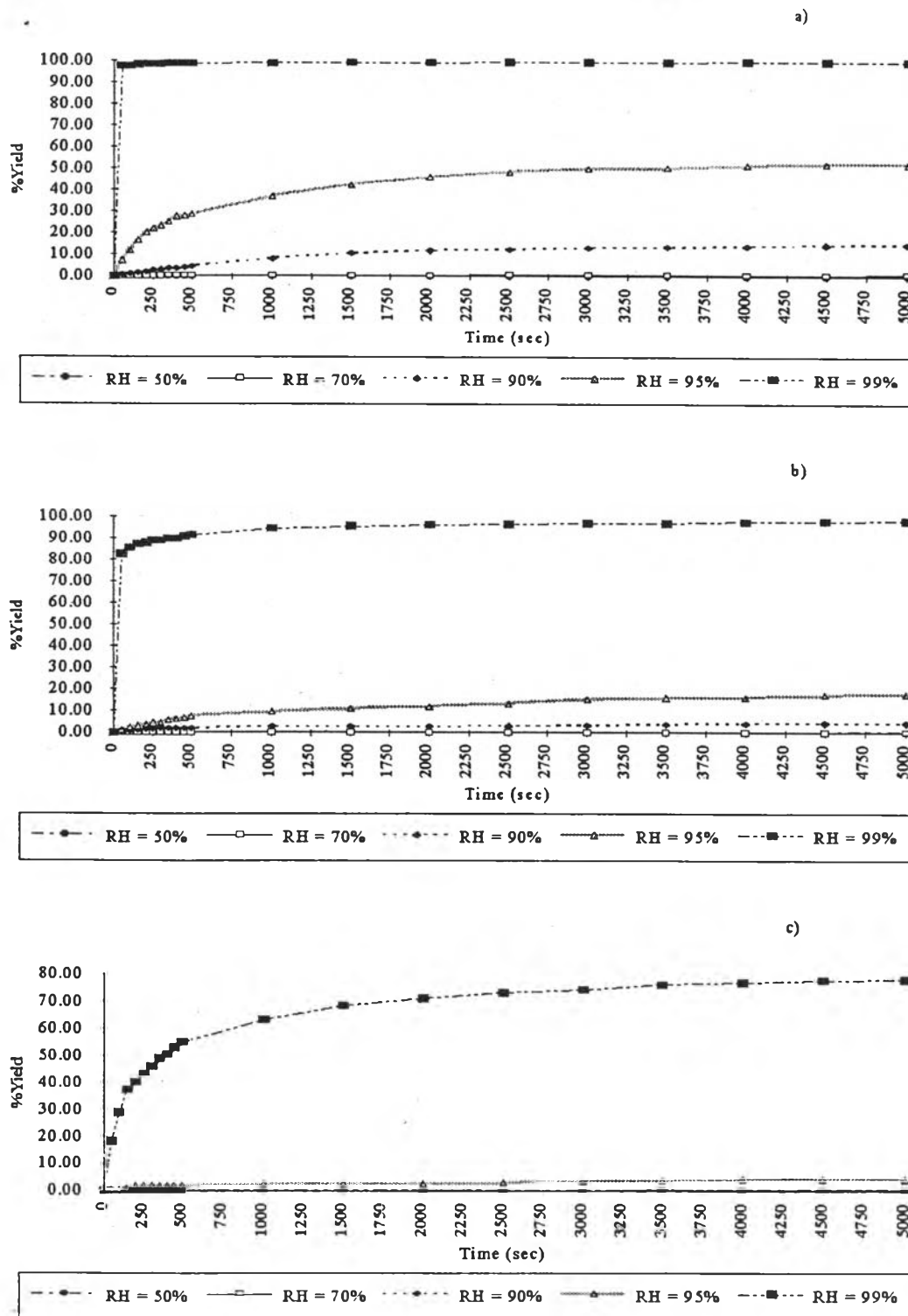


Figure 4.38 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Rich Environment for Atmospheric Stability Class F at $[Fe] = 0.1 \text{ mg/m}^3$ and $[NH_3] = 100 \text{ ppb}$

a) $T = 20 \text{ }^\circ\text{C}$

b) $T = 25 \text{ }^\circ\text{C}$

c) $T = 30 \text{ }^\circ\text{C}$

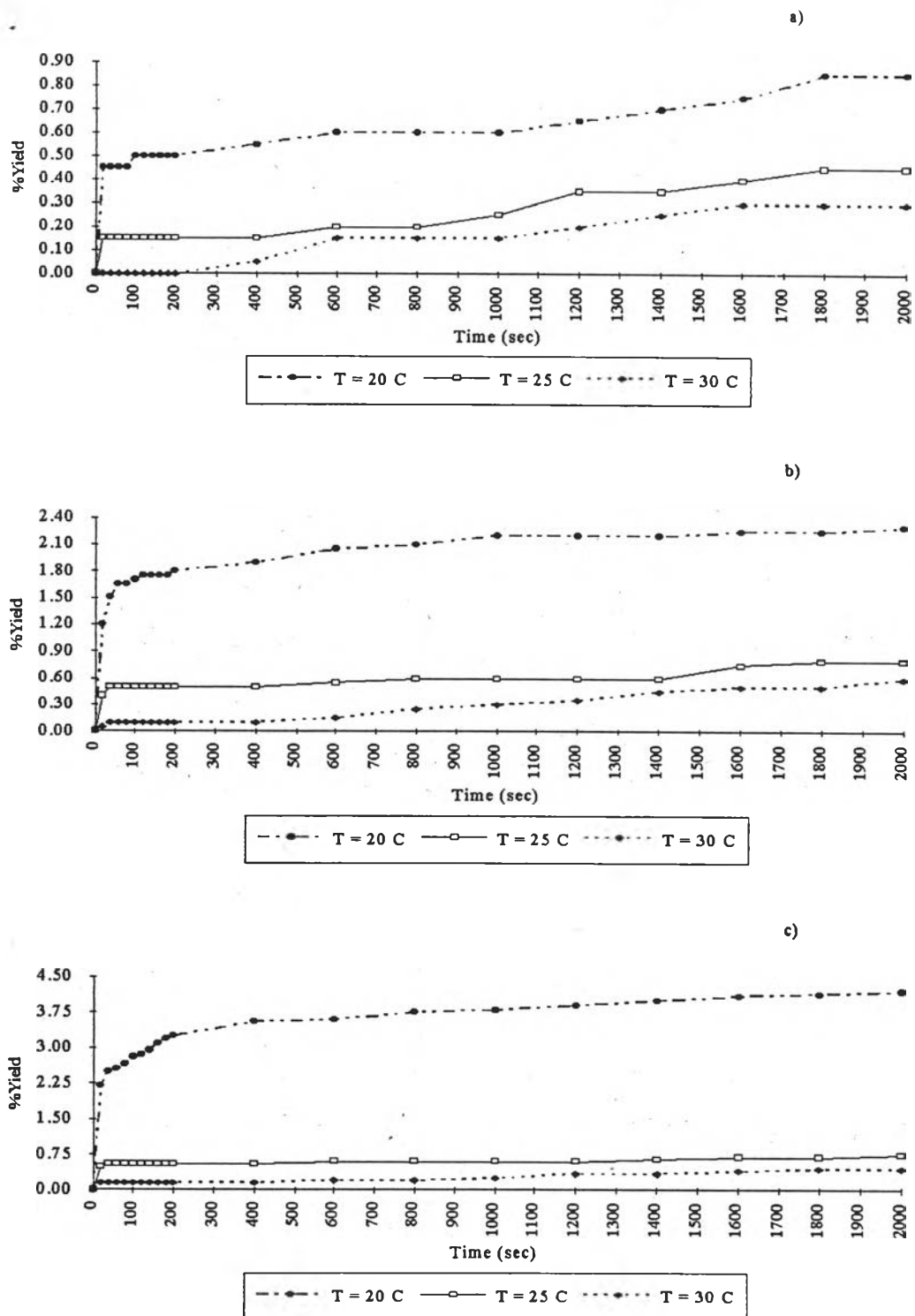


Figure 4.39 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Deficient Environment for Atmospheric Stability Class C at Relative Humidity = 99% and $[Fe] = 1201 \text{ ng/m}^3$

a) $[NH_3] = 50 \text{ ppb}$

b) $[NH_3] = 80 \text{ ppb}$

c) $[NH_3] = 100 \text{ ppb}$

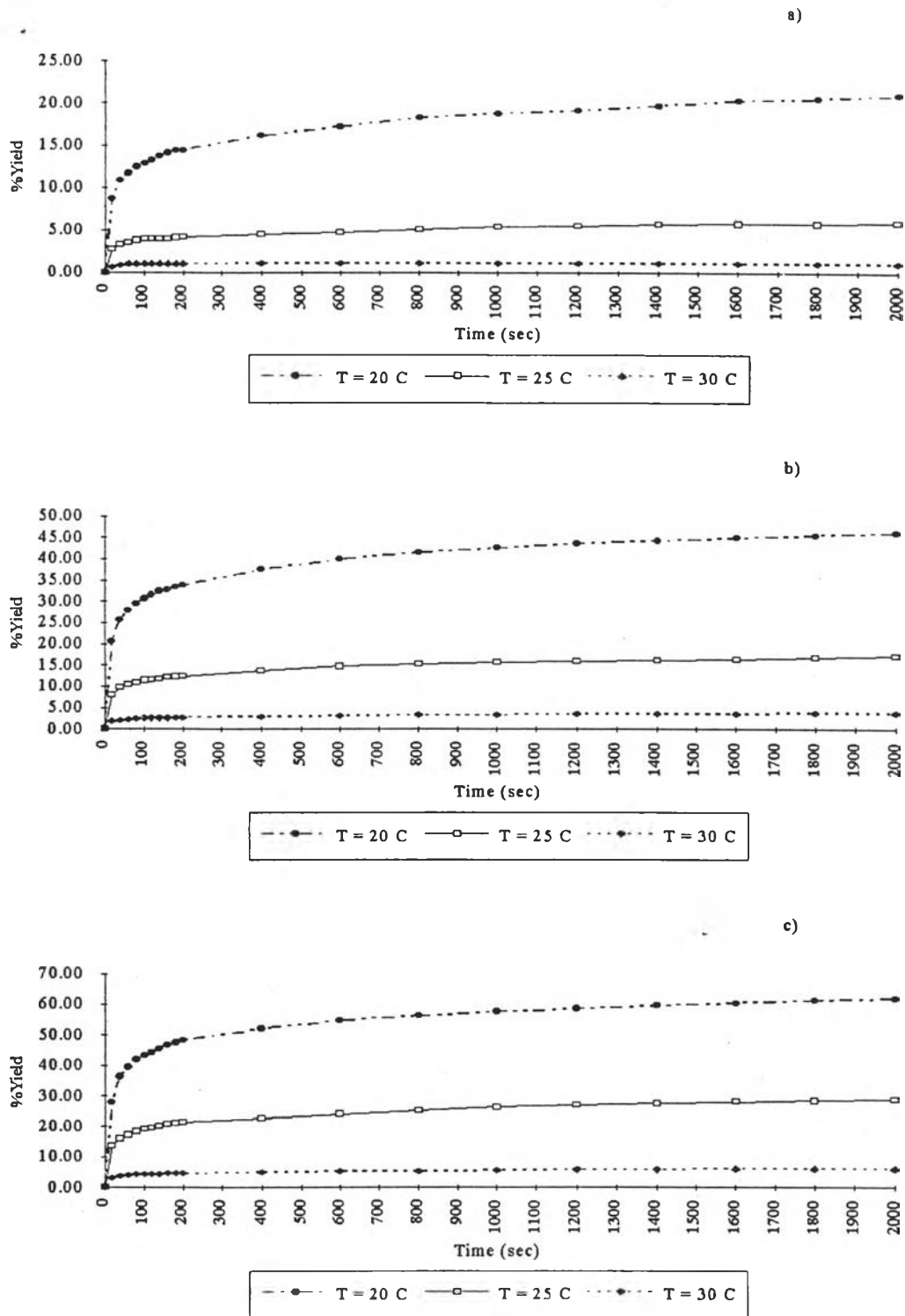


Figure 4.40 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Deficient Environment for Atmospheric Stability Class C at Relative Humidity = 99% and [Fe] = 0.1 mg/m³

a) [NH₃] = 50 ppb

b) [NH₃] = 80 ppb

c) [NH₃] = 100 ppb

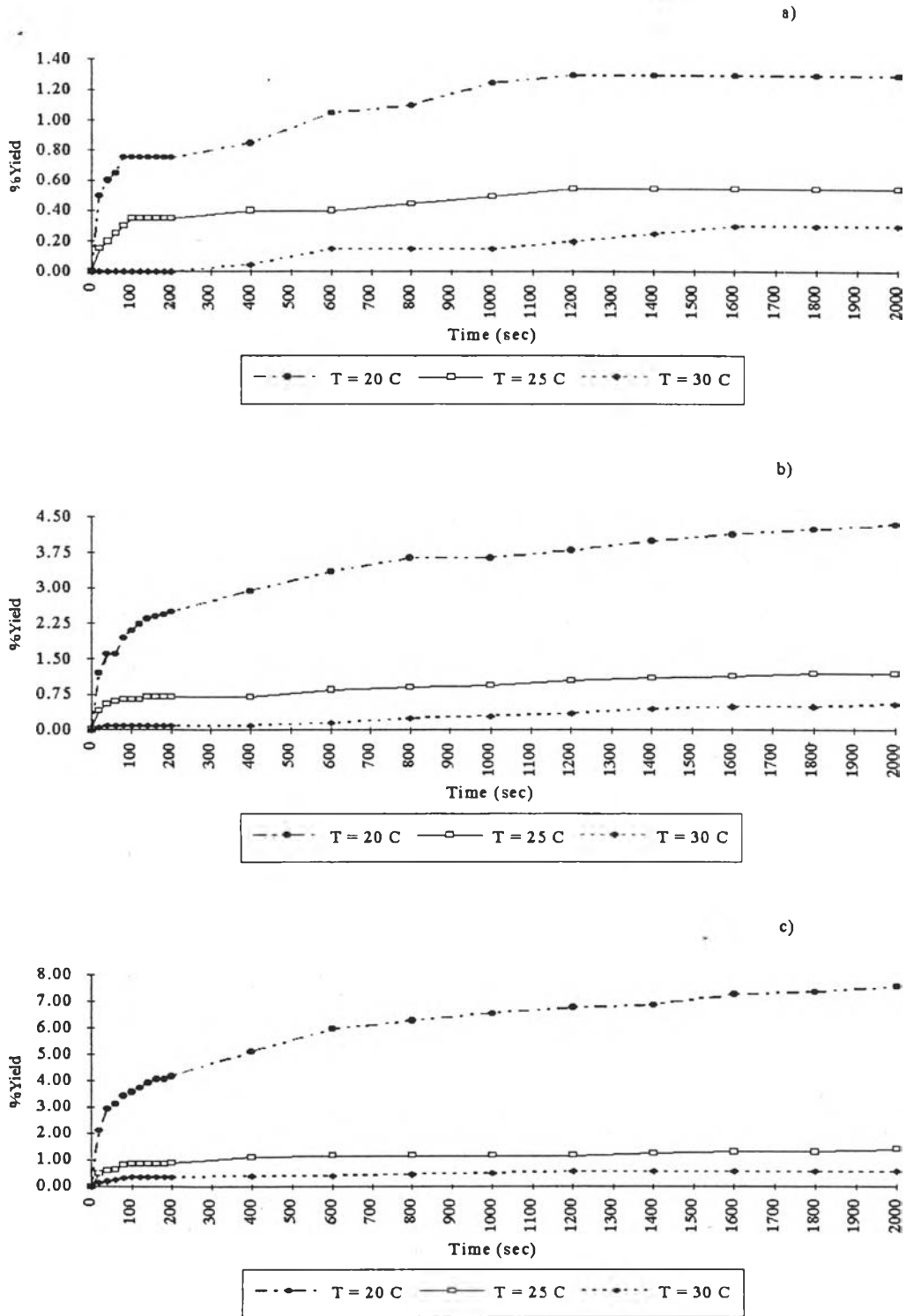


Figure 4.41 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Deficient Environment for Atmospheric Stability Class D at Relative Humidity = 99% and [Fe] = 1201 ng/m³

a) [NH₃] = 50 ppb

b) [NH₃] = 80 ppb

c) [NH₃] = 100 ppb

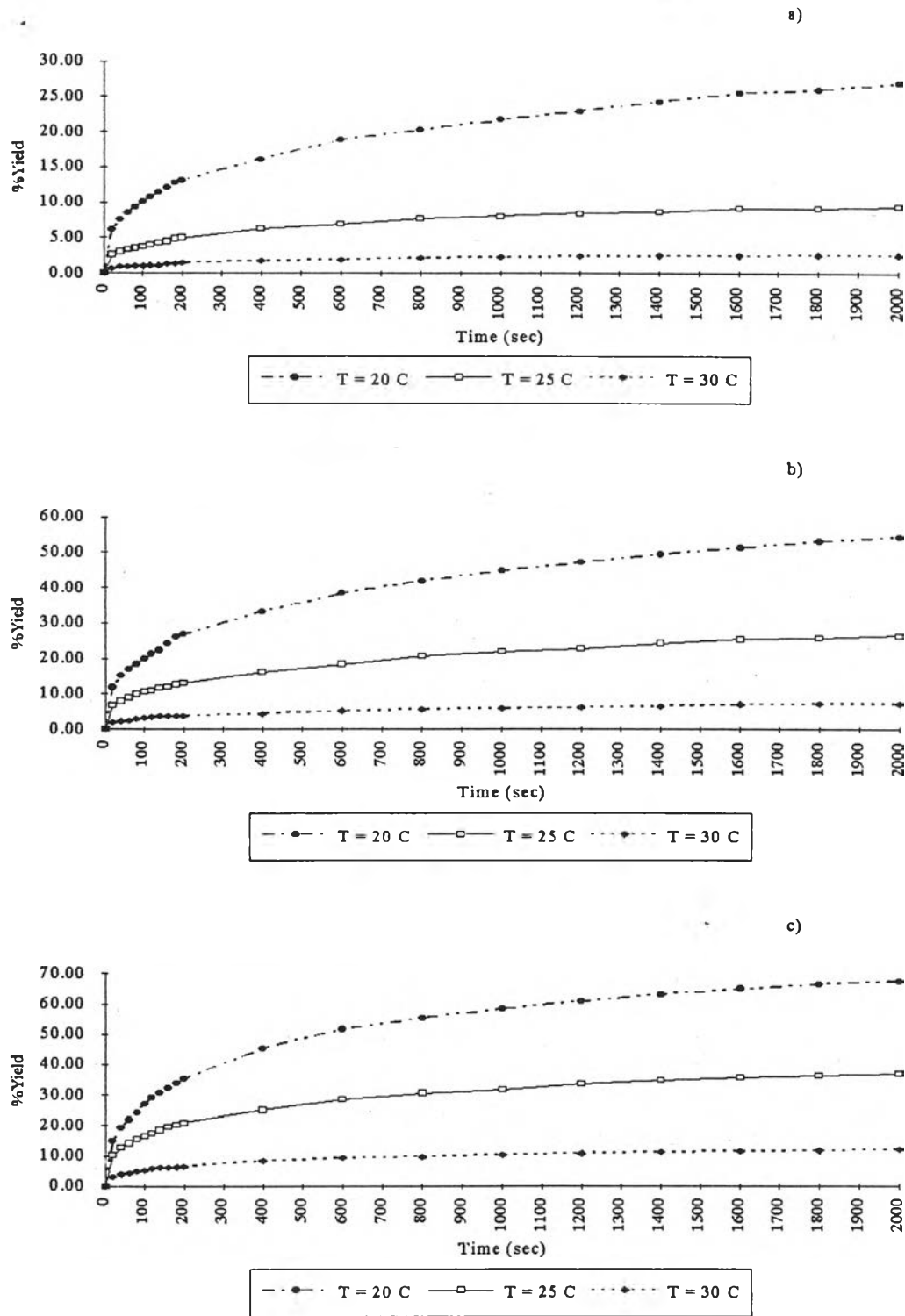


Figure 4.42 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Deficient Environment for Atmospheric Stability Class D at Relative Humidity = 99% and [Fe] = 0.1 mg/m³

a) [NH₃] = 50 ppb

b) [NH₃] = 80 ppb

c) [NH₃] = 100 ppb

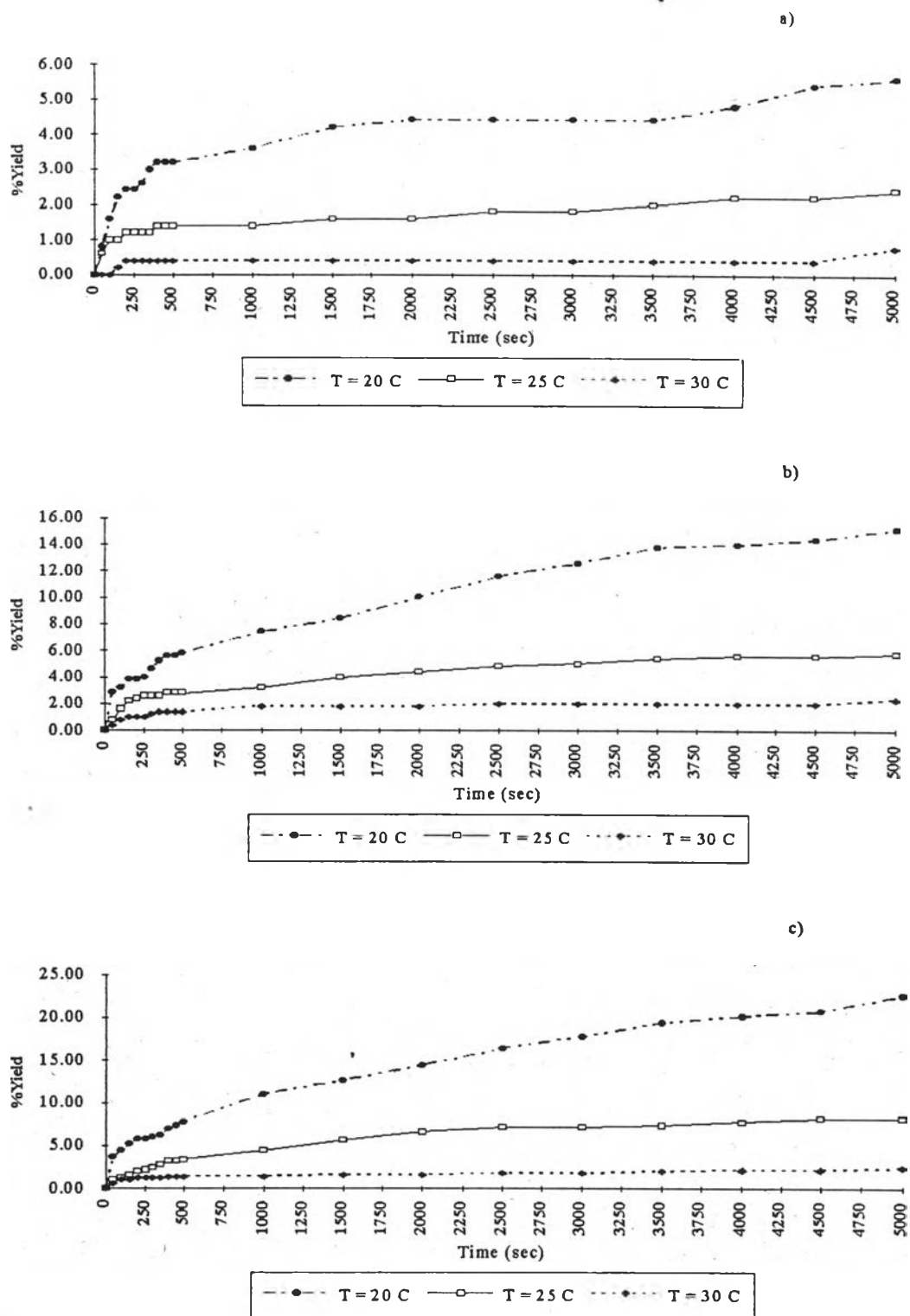


Figure 4.43 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Deficient Environment for Atmospheric Stability Class E at Relative Humidity = 99% and $[Fe] = 1201 \text{ ng/m}^3$

a) $[NH_3] = 50 \text{ ppb}$

b) $[NH_3] = 80 \text{ ppb}$

c) $[NH_3] = 100 \text{ ppb}$

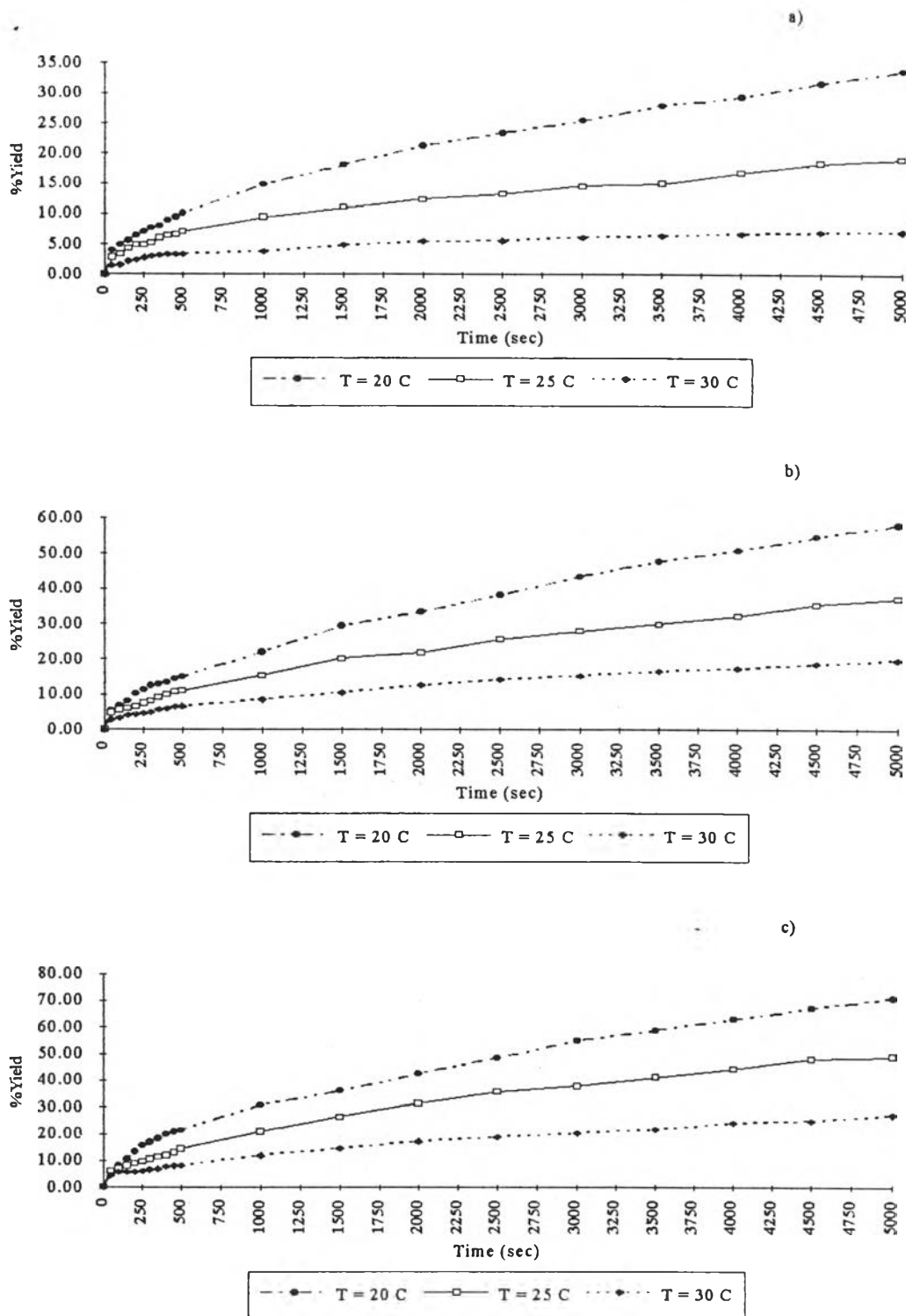


Figure 4.44 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Deficient Environment for Atmospheric Stability Class E at Relative Humidity = 99% and [Fe] = 0.1 mg/m³

a) [NH₃] = 50 ppb

b) [NH₃] = 80 ppb

c) [NH₃] = 100 ppb

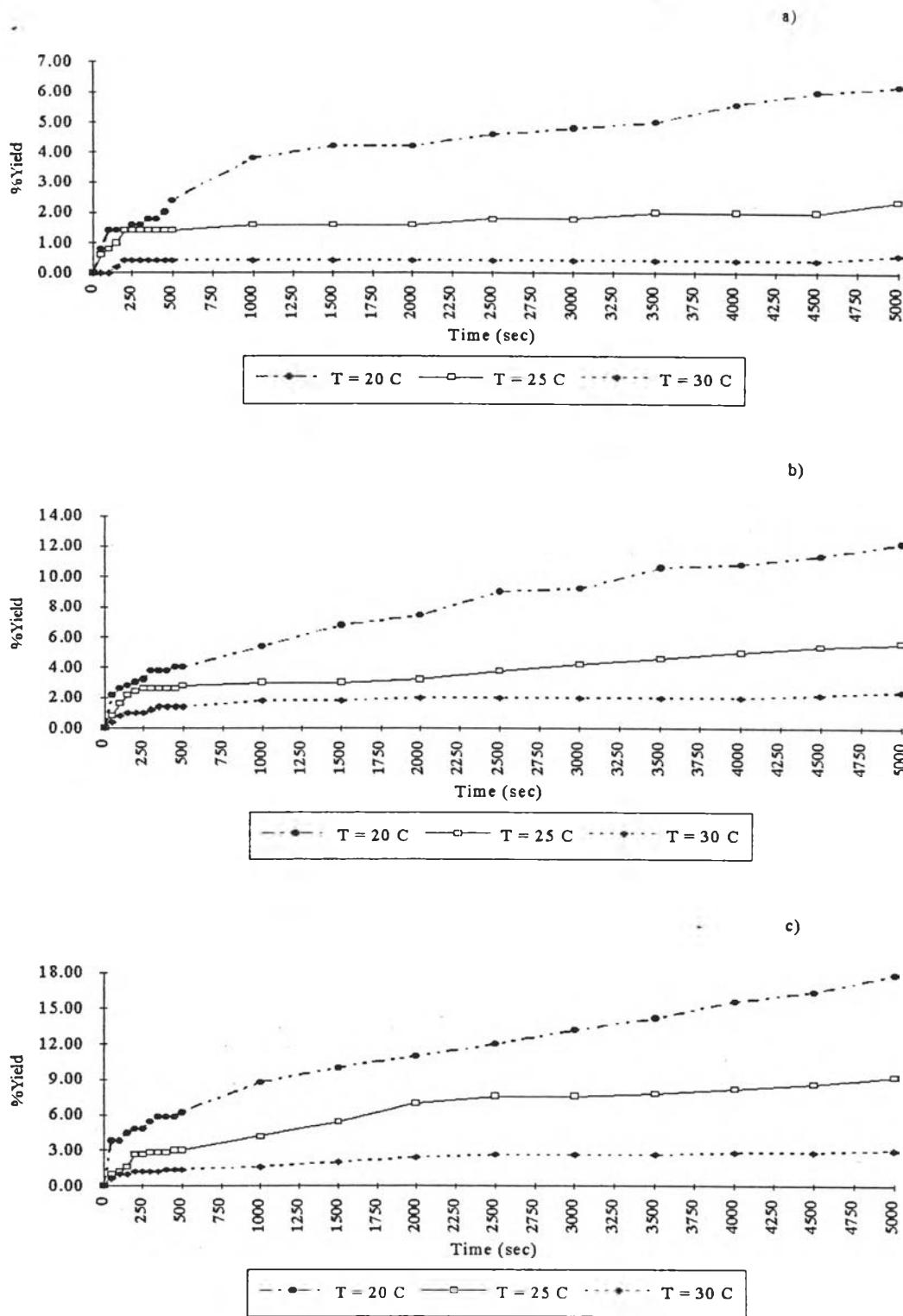


Figure 4.45 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Deficient Environment for Atmospheric Stability Class F at Relative Humidity = 99% and [Fe] = 1201 ng/m³

a) [NH₃] = 50 ppb

b) [NH₃] = 80 ppb

c) [NH₃] = 100 ppb

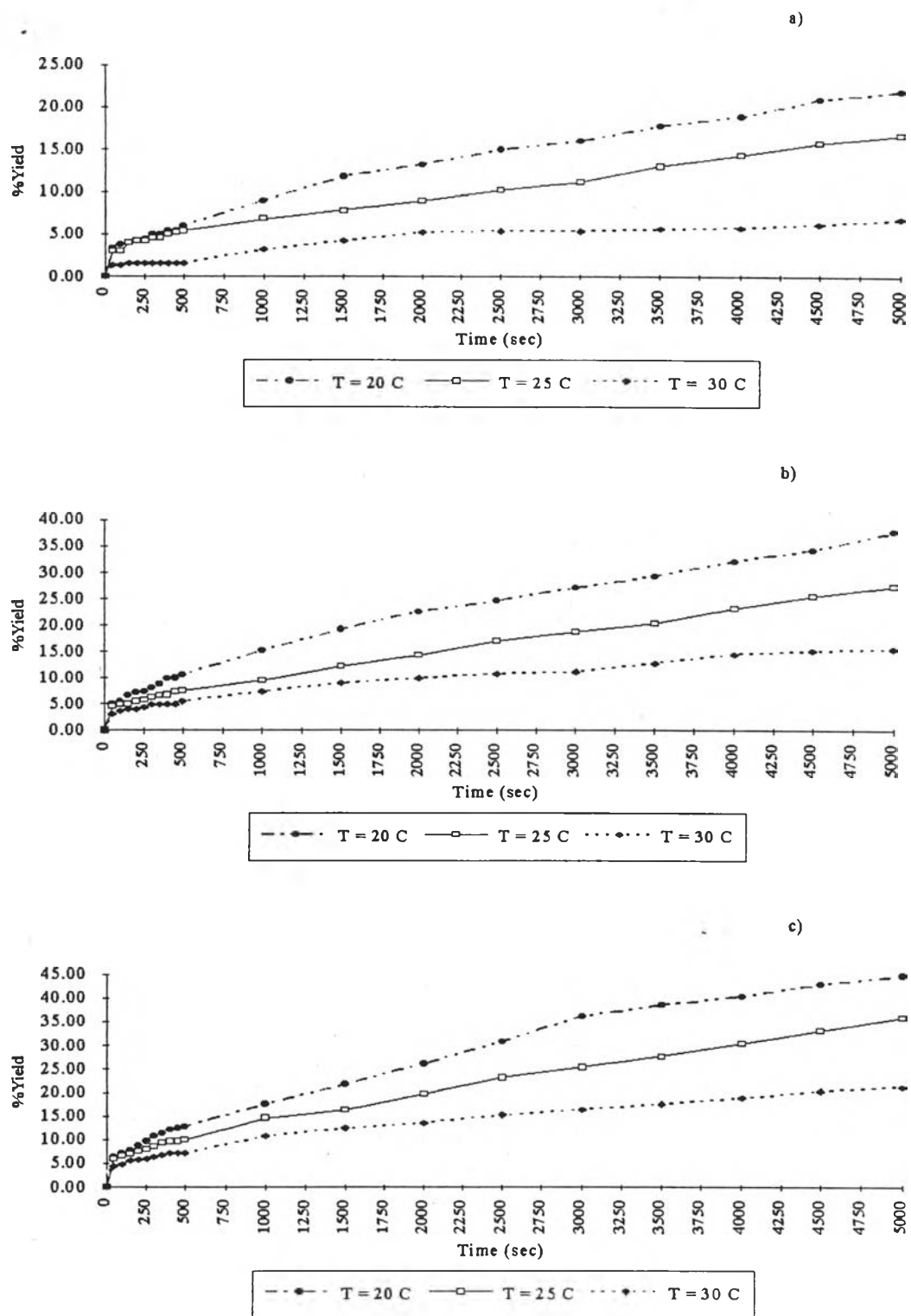


Figure 4.46 %Yield VS Time of Freiberg(1974)'s Reaction Rate in Ammonia-Deficient Environment for Atmospheric Stability Class F at Relative Humidity = 99% and [Fe] = 0.1 mg/m³

a) [NH₃] = 50 ppb b) [NH₃] = 80 ppb c) [NH₃] = 100 ppb

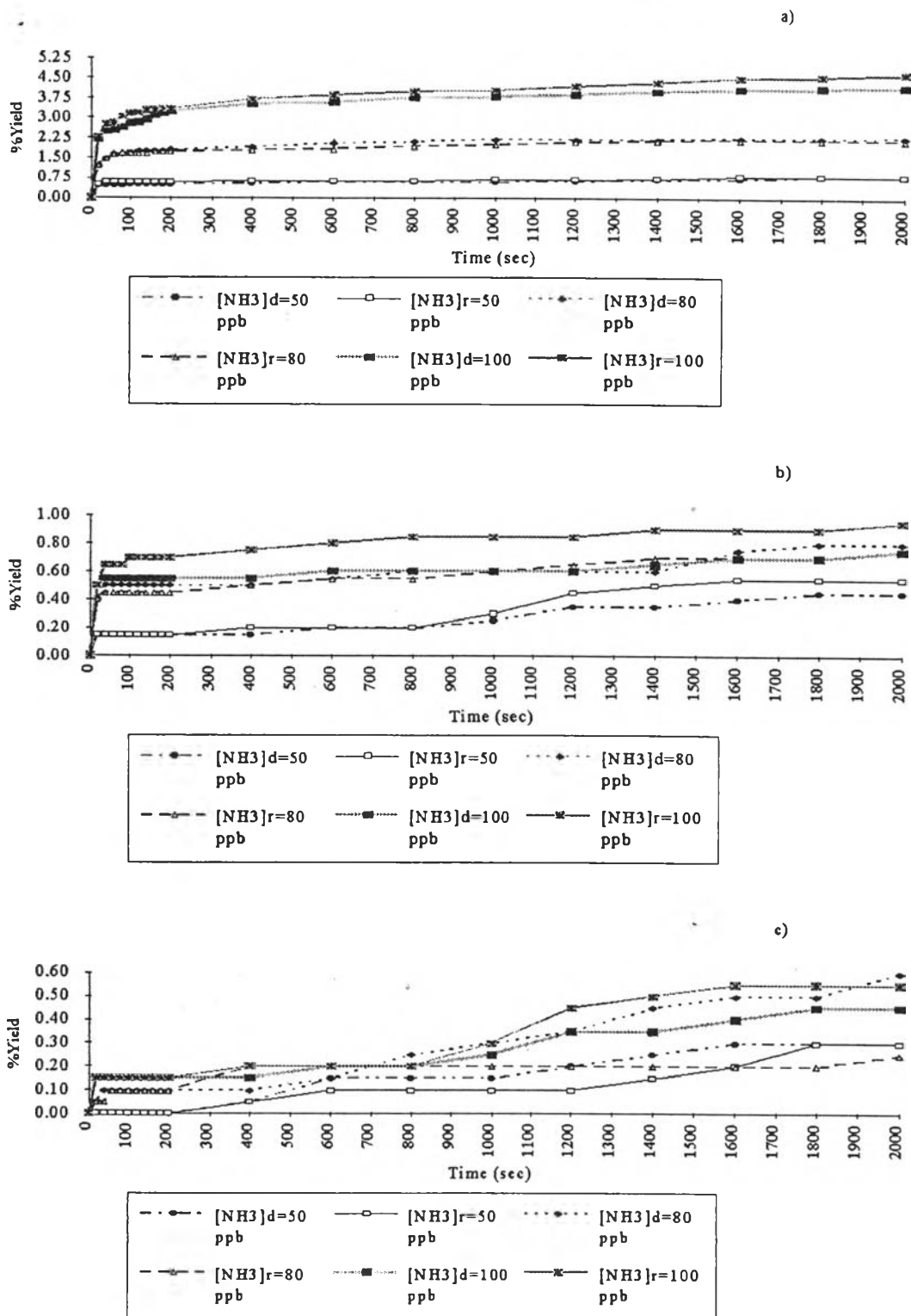


Figure 4.47 Comparison of Freiberg(1974)'s Yield between in Ammonia-Rich Environment and in Ammonia-Deficient Environment for Atmospheric Stability Class C at Relative Humidity = 99% and [Fe] = 1201 ng/m³

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

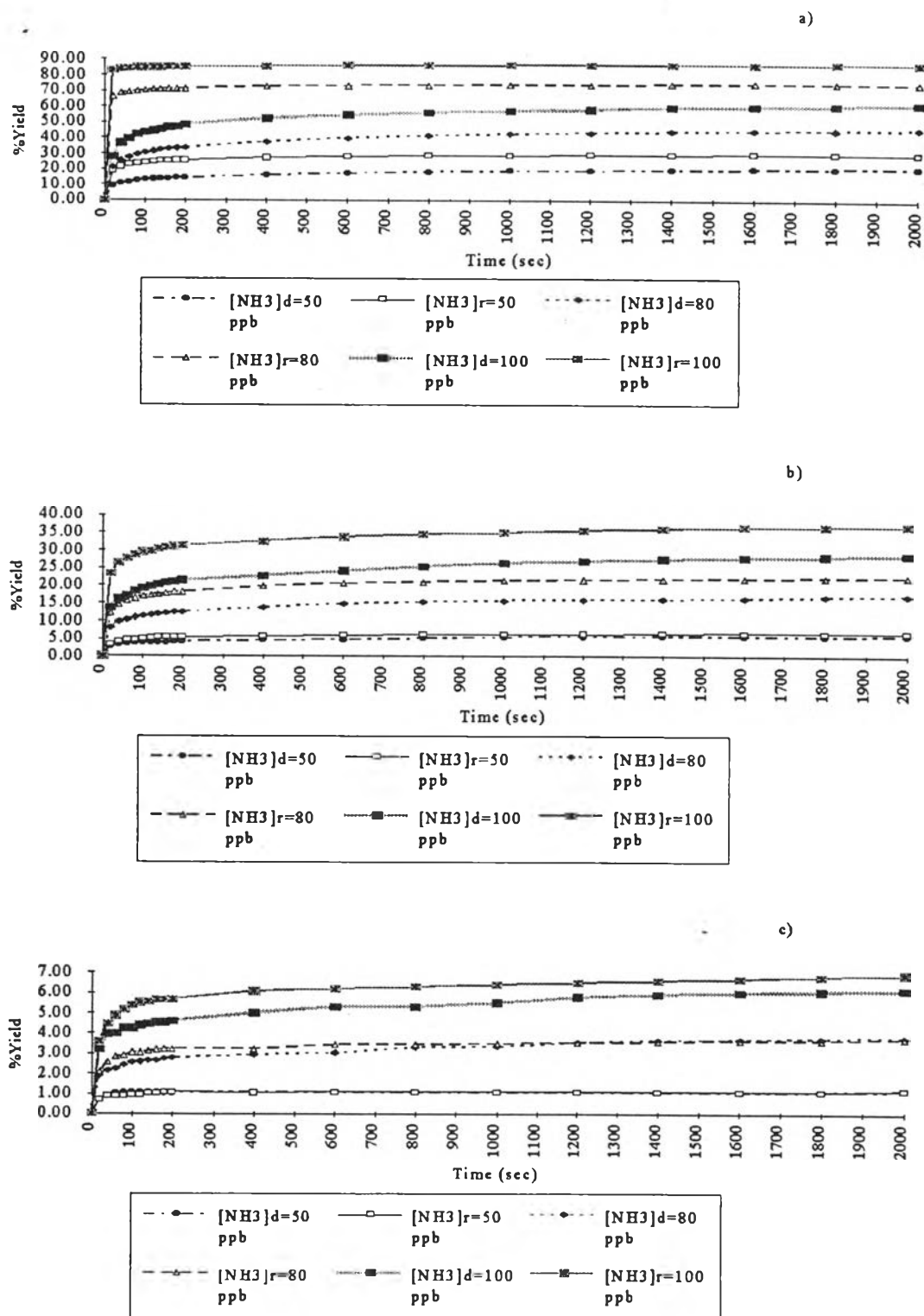


Figure 4.48 Comparison of Freiberg(1974)'s Yield between in Ammonia-Rich Environment and in Ammonia-Deficient Environment for Atmospheric Stability Class C at Relative Humidity = 99% and [Fe] = 0.1 mg/m³

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

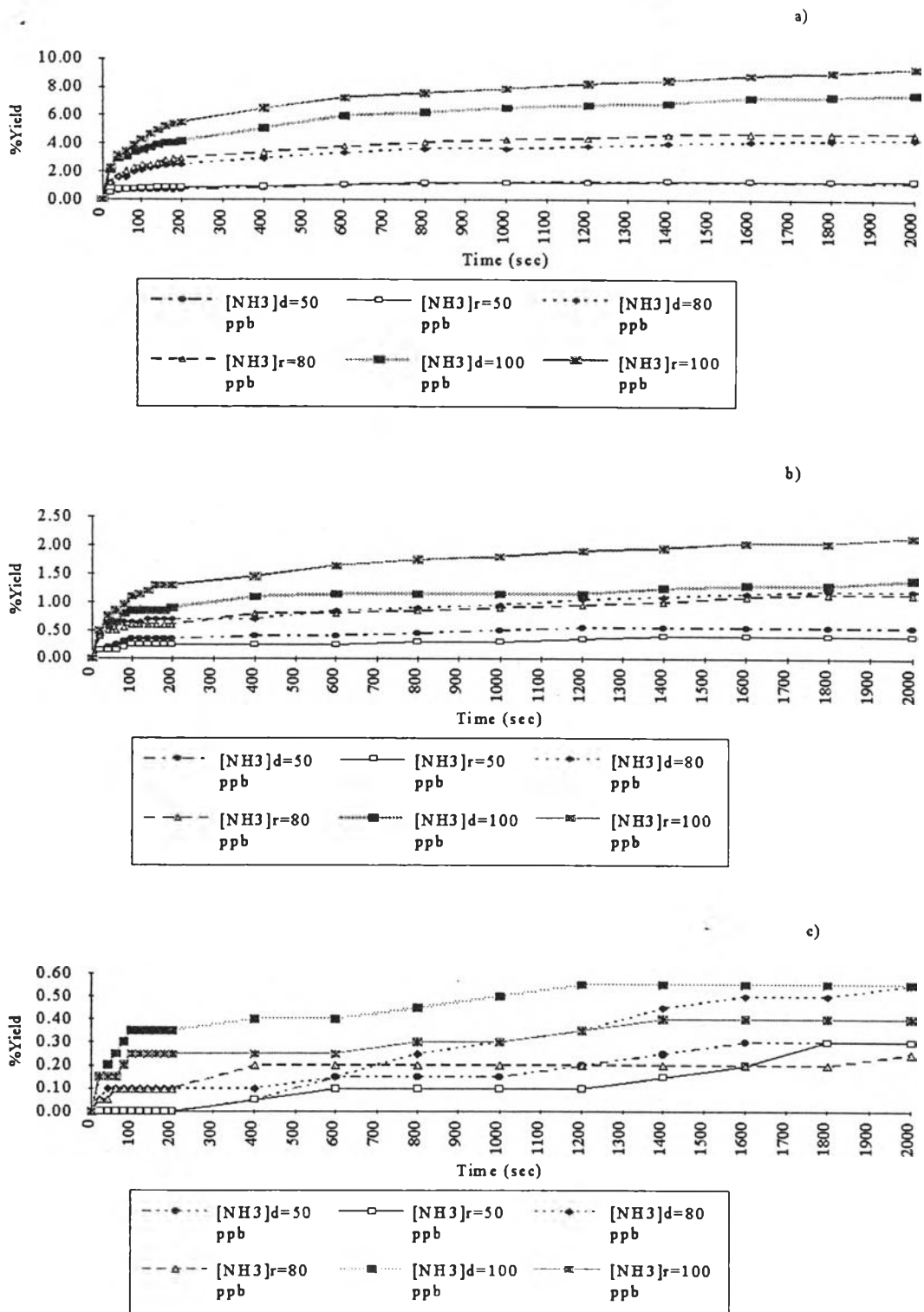


Figure 4.49 Comparison of Freiberg(1974)'s Yield between in Ammonia-Rich Environment and in Ammonia-Deficient Environment for Atmospheric Stability Class D at Relative Humidity = 99% and [Fe] = 1201 ng/m³

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

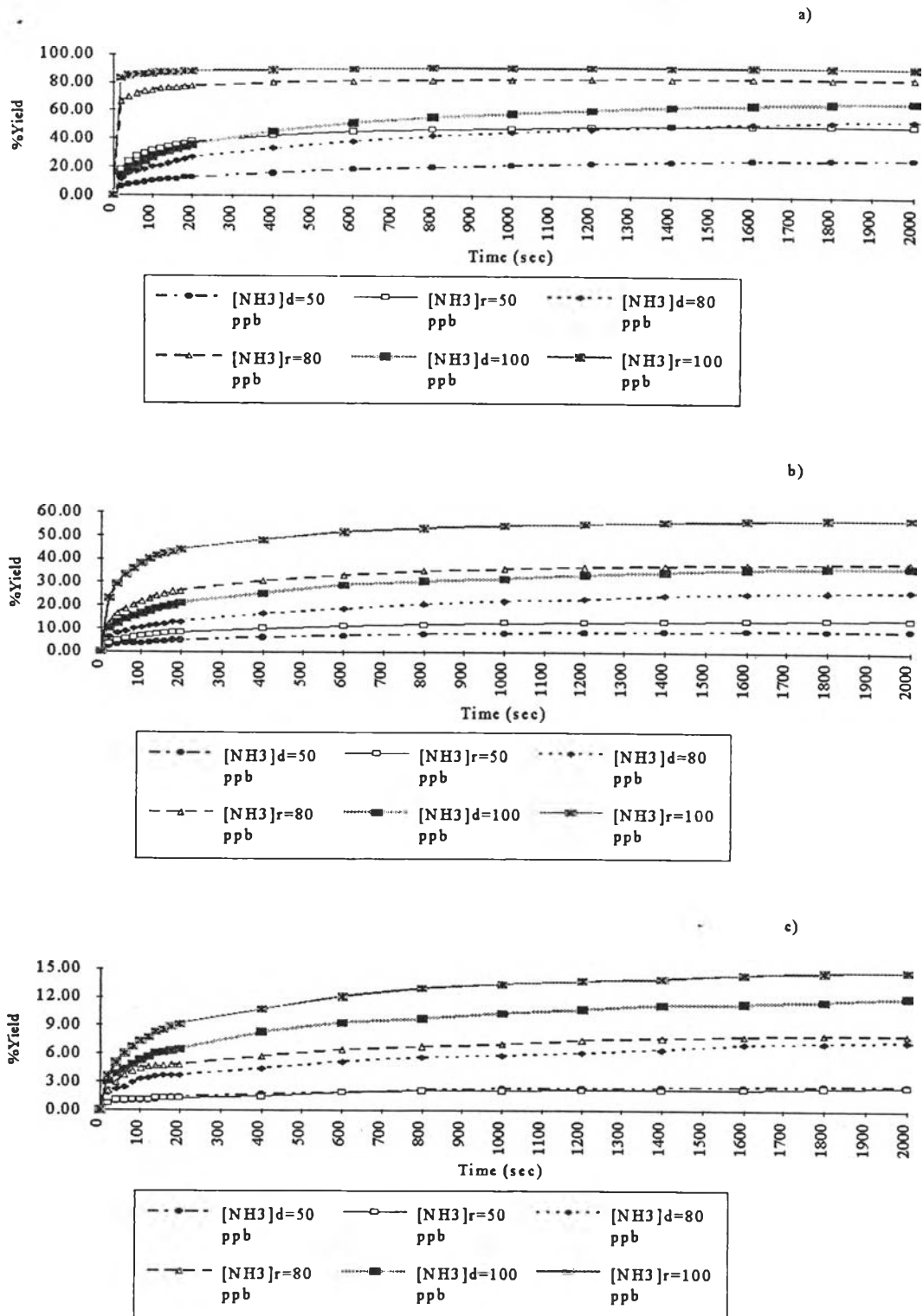


Figure 4.50 Comparison of Freiberg(1974)'s Yield between in Ammonia-Rich Environment and in Ammonia-Deficient Environment for Atmospheric Stability Class D at Relative Humidity = 99% and [Fe] = 0.1 mg/m³

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

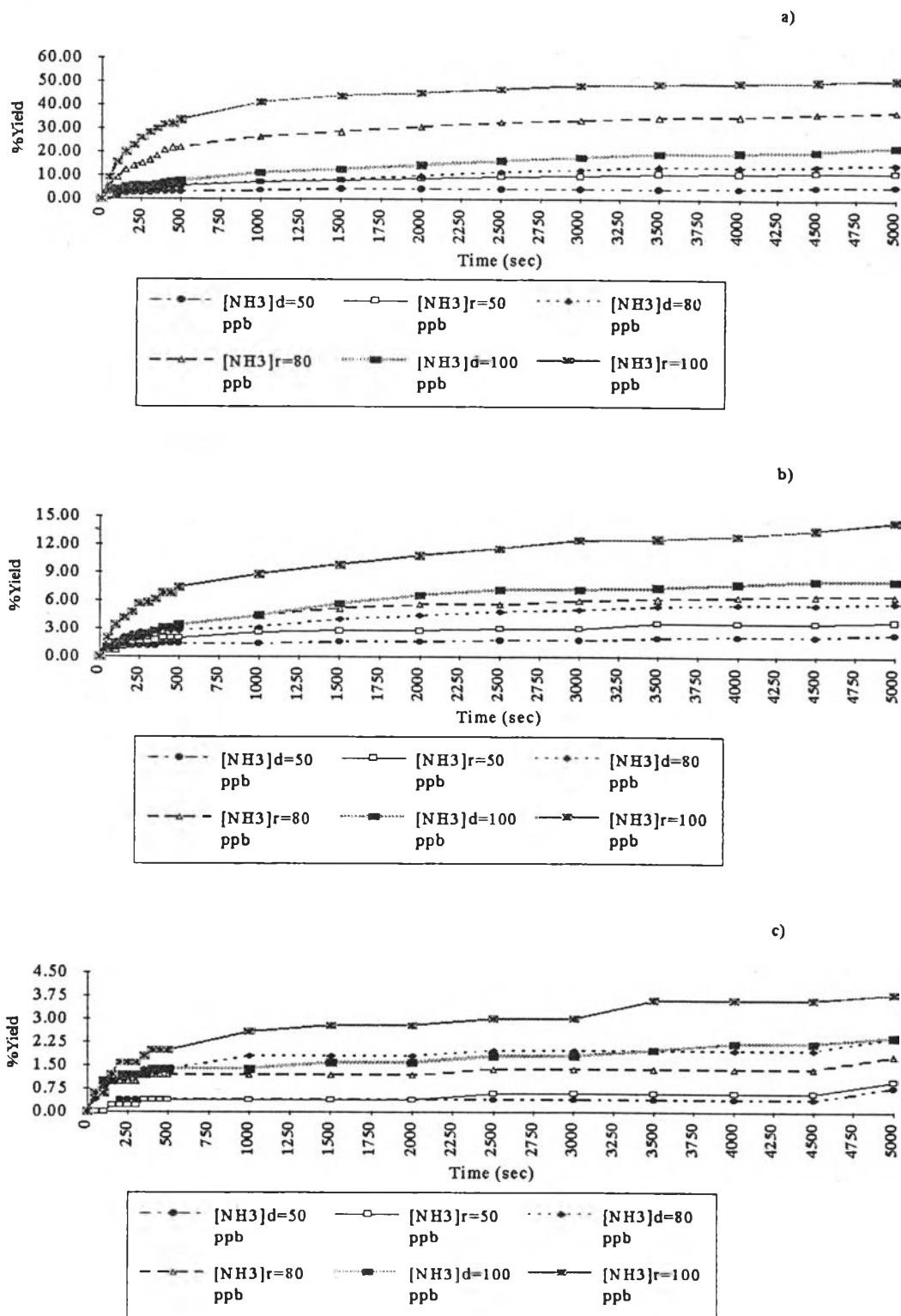


Figure 4.51 Comparison of Freiberg(1974)'s Yield between in Ammonia-Rich Environment and in Ammonia-Deficient Environment for Atmospheric Stability Class E at Relative Humidity = 99% and [Fe] = 1201 ng/m³

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

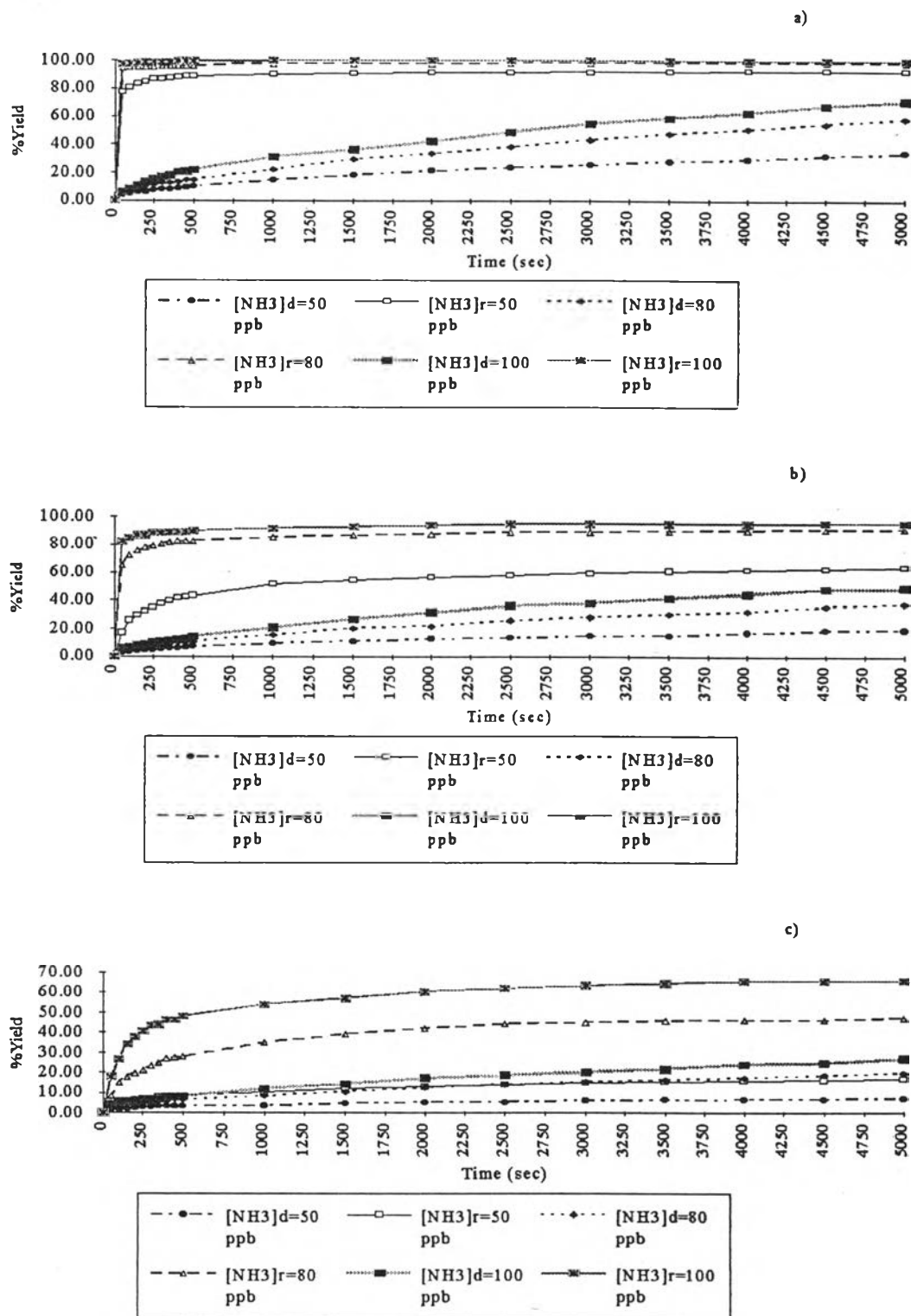


Figure 4.52 Comparison of Freiberg(1974)'s Yield between in Ammonia-Rich Environment and in Ammonia-Deficient Environment for Atmospheric Stability Class E at Relative Humidity = 99% and $[Fe] = 0.1 \text{ mg/m}^3$

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

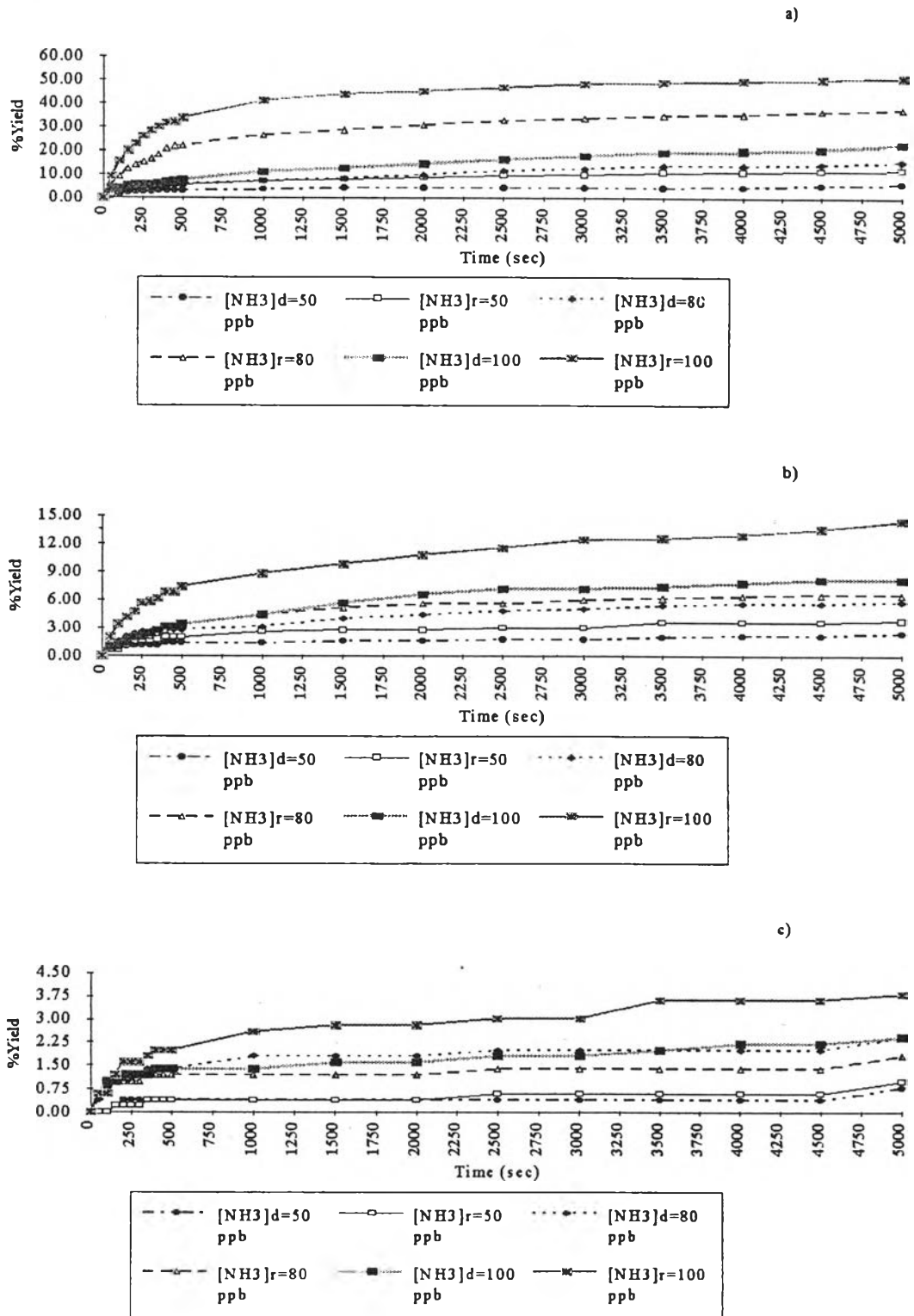


Figure 4.53 Comparison of Freiberg(1974)'s Yield between in Ammonia-Rich Environment and in Ammonia-Deficient Environment for Atmospheric Stability Class F at Relative Humidity = 99% and [Fe] = 1201 ng/m³

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

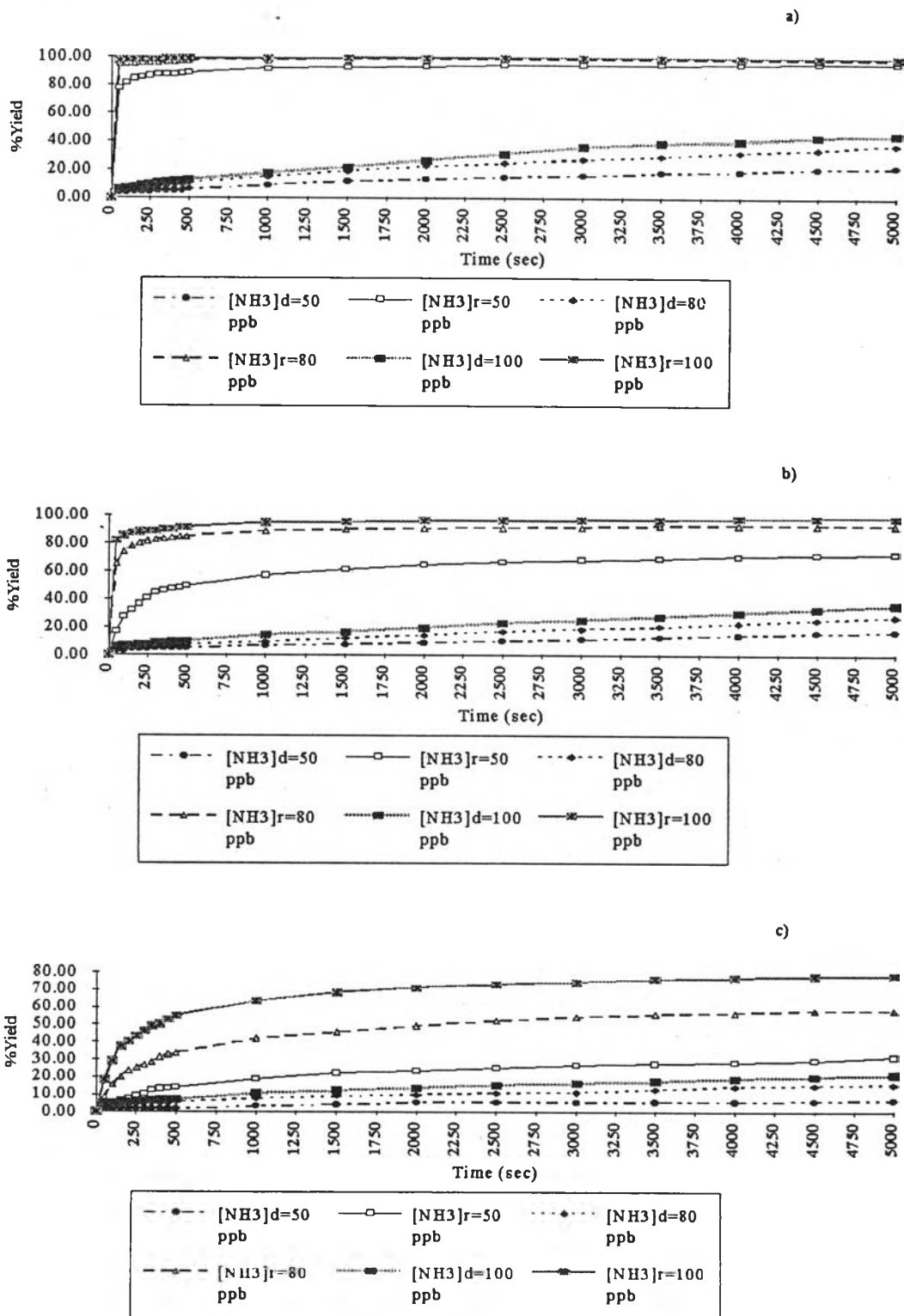


Figure 4.54 Comparison of Freiberg(1974)'s Yield between in Ammonia-Rich Environment and in Ammonia-Deficient Environment for Atmospheric Stability Class F at Relative Humidity = 99% and [Fe] = 0.1 mg/m³

a) T = 20 °C

b) T = 25 °C

c) T = 30 °C

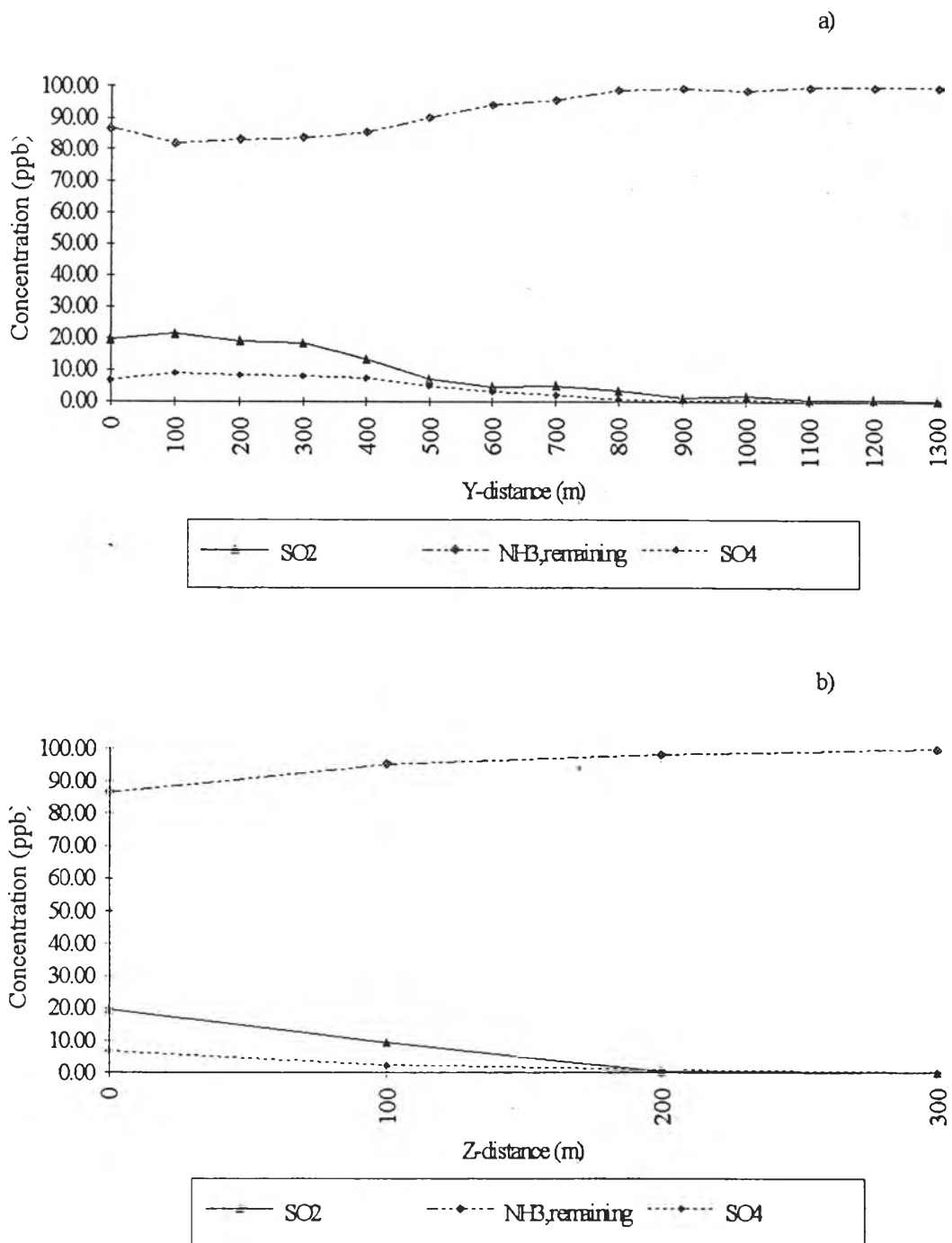


Figure 4.55 Sulfur Dioxide, Remaining Ammonia and Sulfate Concentration Profiles of Freiberg (1974)'s Reaction Rate in Ammonia-Deficient Environment for Atmospheric Stability Class D at 4 km Downwind from the Source, Relative Humidity = 99%, $T = 25\text{ }^{\circ}\text{C}$, $[\text{Fe}] = 0.1\text{ mg/m}^3$ and $[\text{NH}_3] = 100\text{ ppb}$

a) Varying Y-Distance (Fixed $z=0$)

b) Varying Z-Distance (Fixed $y=0$)

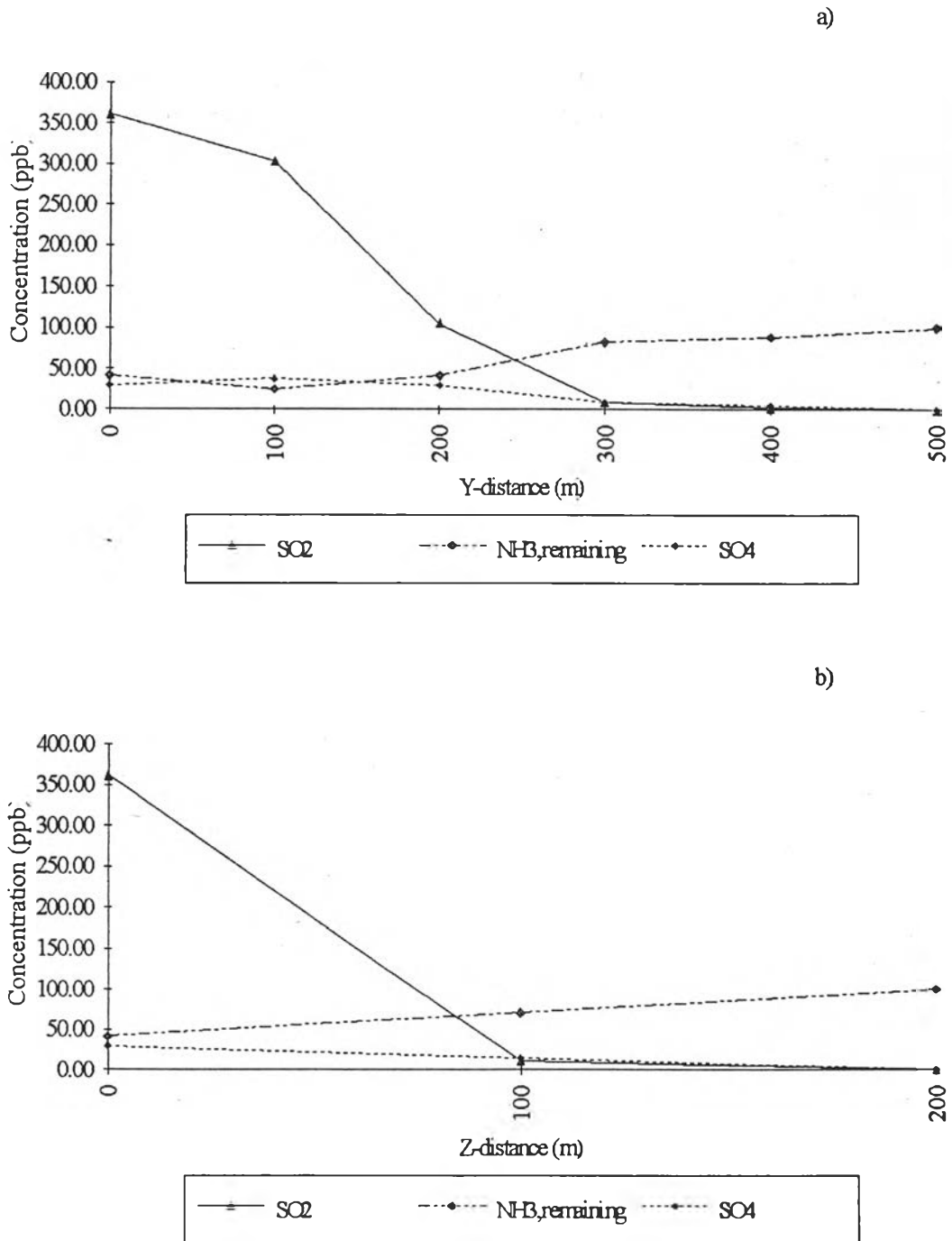


Figure 4.56 Sulfur Dioxide, Remaining Ammonia and Sulfate Concentration Profiles of Freiberg (1974)'s Reaction Rate in Ammonia-Deficient Environment for Atmospheric Stability Class F at 2 km Downwind from the Source, Relative Humidity = 99%, T = 20 °C, [Fé] = 0.1 mg/m³ and [NH₃] = 100 ppb

a) Varying Y-Distance (Fixed z=0)

b) Varing Z-Distance (Fixed y=0)

Vita

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