



REFERENCES

- A. J. Nijjenhuis, D. W. Grijpma and A. J. Pennings (1996). Polymer, 37, 2783.
- A. Mousa, U.S. Ishiaku and Z.A. Mohd Ishak (2000). Rheological properties of dynamically vulcanized poly(vinyl chloride)/epoxidized natural rubber thermoplastic elastomers: effect of processing variables. Polymer Testing, 19, 193–204.
- Abdou - Sabet, S., Puydak, R.C., and Rader, C.P. (1996). Dynamically vulcanized thermoplastic elastomers. Rubber Chemistry and Technology, 69, 476-494.
- Ahmad Mousa, U. S. Ishiaku and Z. A. Mohd Ishak (2005). Oil Resistance of Dynamically Vulcanized Poly(Vinyl Chloride)/Nitrile Butadiene Rubber Thermoplastic Elastomers. Polymer Bulletin, 53, 203–212.
- Akiba, M. and Hashim, A.S. (1997). Vulcanization and crosslinking in elastomers. Progress in polymer science, 22(3), 475-521.
- Asha Elizabeth Mathai and Sabu Thomas (2005). Morphology, mechanical and viscoelastic properties of nitrile rubber/epoxidized natural rubber blends. Journal of applied polymer science, 97, 1561 – 1573.
- C. Nakason, P. Wannavilai and A. Kaesaman (2006). Effect of vulcanization system on properties of thermoplastic vulcanizates based on epoxidized natural rubber/polypropylene blends. Polymer Testing, 25, 34–41.
- C. S. L. Baker, I. R. Gelling and R. Newell (1985). Epoxidized Natural Rubber. Rubber Chem. Technol., 58.
- Chantara Theyy Ratnam and Mohd Saiful Ahmad (2006). Stress – Strain, morphological and rheological properties of radiation – crosslinked PVC/ENR blend. Malaysian Polymer Journal, 1, 1 – 10.
- Chen G. X. and Yoon J. S. (2005). Thermal stability of poly(L-lactide)/poly(butylene succinate)/clay nanocomposites. Polym Degrad Stab, 88(2), 206 – 212.
- Chen MH, Hsu YH, Lin CP, Chen YJ, Young TH. (2005). Interactions of acinar cells on biomaterials with various surface properties. JBiomed Mater Res A, 74(2), 254-262.

- Chen RS, Chen YJ, Chen MH and Young TH (2007). Cell-surface interactions of rat tooth germ cells on various biomaterials. Journal of Biomedical Materials Research, 83(1), 241-248.
- Das A, Costa FR, Wagenknecht U, Heinrich G (2008). Nanocomposites based on chloroprene rubber: effect of chemical nature and organic modification of nanoclay on the vulcanizate properties. Eur Polym J, 44, 3456 – 65.
- Drobny, J.G. (Ed.). (1999). Technology of Fluoropolymers. New Hampshire: CRC Press.
- F. S. Liao, A. C. Su and Tzu – Chien J. Hsu (1994). Polymer, 35, 2579.
- Frank M. McMillan and De Loss E. Winkler (1952). Softening agent for rubber and resultant rubber composition. Patented no. 2,582,264.
- G.H. Yew, W.S. Chow, Z.A. Mohd Ishak and A.M. Mohd Yusof (2009). Natural Weathering of Poly (Lactic Acid): Effects of Rice Starch and Epoxidized Natural Rubber. Journal of Elastomers and Plastics, 41(4), 369-382.
- H. Ismail and R. Ramli (2007). Organoclay filled natural rubber nanocomposites: The effects of filler loading and mixing method. Journal of reinforced plastics and composites, 27, 16 – 17.
- H. Ismail and S. Suzaimah (2000). Styrene butadiene rubber/epoxidized natural rubber blends: dynamic properties, curing characteristics and swelling studies. Polymer Testing, 19, 879–888.
- Hagen, R., Salmén, L., and Stenberg, B. (1996). Effects of the type of crosslink on viscoelastic properties of natural rubber. Journal of Polymer Science Part B: Polymer Physics, 34(12), 1997-2006.
- Hana Ismail and H.M. Hairunezam (2001). The effect of a compatibilizer on curing characteristics, mechanical properties and oil resistance of styrene butadiene rubber/epoxidized natural rubber blends. European Polymer Journal, 37, 39-44.
- Hanafî Ismail and H. Anuar (2000). Palm oil fatty acid as an activator in carbon black filled natural rubber compounds: dynamic properties, curing characteristics, reversion and fatigue studies. Polymer testing, 19, 349-359.

- Holden, G., Kricheldorf, H.R., and Quirk, R.P. (Eds.). (2004). Thermoplastic Elastomers. Munich: Hanser Verlag, 143-146.
- Hsin - Chieh Chen, Chen - Hao Tsai and Ming – Chien Yang (2011). Mechanical properties and biocompatibility of electrospun polylactide and poly(vinylidene fluoride) mats. J Polym Res, 18, 319-327.
- Ishak ZAM and Baker AA (1995). An investigation on the potentials of rice husk ash as filler for epoxidized natural rubber. Eur. Polym. J., 31(3), 259 – 269.
- Jiamjitsiripong K. and Pattamaprom C. (2011). Effects of Epoxidized Natural Rubber on Gas Barrier and mechanical properties of NR/BIIR composites. Journal of Elastomers and Plastics., Published online before print at <<http://jep.sagepub.com/content/early/2011/03/11/0095244311400502>>
- Joy K. Mishra, Young-Wook Chang and Wonho Kim (2011). The effect of peroxide crosslinking on thermal, mechanical, and rheological properties of polycaprolactone/epoxidized natural rubber blends. Polym. Bull., 66, 673-681.
- Jürdens, K. (2002). The pan-European strategic market study TPE 2000. TPE in der Prozesskette, VDI-Gesellschaft Kunststofftechnik, VDI-Verlag, Düsseldorf.
- Kaushik Pal, R. Rajasekar, Dong Jin Kang, Zhen Xiu Zhang, Jin Kuk Kim and C.K. Das (2009). Effect of epoxidized natural rubber–organoclay nanocomposites on NR/high styrene rubber blends with fillers. Materials and Design, 30, 4035–4042.
- L. E. Yahaya, K. O. Adebawale, A. R. R. Menon, S. Rugmini, B. I. Olu – Owolabi and J. Chameswary (2010). Natural rubber/organoclay nanocomposites: Effect of filler dosage on the physicomaterial properties of vulcanizates. African Journal of Pure and Applied Chemistry, 4(9), 198 – 205.
- Li. Y and Shimizu H. (2007). Macromol Biosci, 7, 921.
- Lyons, B. J. (1993). The effect of radiation on the physical and chemical properties of polyethylene containing antioxidant and /or other additives. Radiation Physical Chemistry, 42, 197 – 205.
- M.C. Senake Perera, U.S. Ishiaku and Z.A. Mohd. Ishak (2001). Characterisation of PVC/NBR and PVC/ENR50 binary blends and PVC/ENR50/NBR ternary

- blends by DMA and solid state NMR. European Polymer Journal, 37, 167-178.
- Marega C. and Marigo A. (2003). Influence of annealing and chain defects on the melting behavior of poly(vinylidene fluoride). European Polymer Journal, 39(8), 1713-1720.
- Mauro PJD, deRudder J, Etienne JP (1990). Rubber world. January.
- Nelisen L. E. (1962). Mechanical properties of polymers. Van Nostrand Reinhold Company, New York.
- Nor Azowa Ibrahim, Wan Md Zin Wan Yunus, Maizatunisa Othman, Khalina Abdan and Kamarul Arifin Hadithon (2010). Poly(Lactic Acid) (PLA) – reinforced Kenaf Bast Fiber Composites: The effect of Triacetin. Journal of reinforced plastics and composites, 29, 1099.
- P. Boonfaung, P. Wasutchanon and A. Somwangthanaroj (2011). Development of packaging film from bioplastic polylactic acid (PLA) with plasticizers. PACCON2011 (Pure and applied chemistry international conference).
- P. L. Teh, Z. A. Mohd Ishak, U. S. Ishiaku and J. Karger-Kocsis (2003). Jurnal Teknologi 39 (A) Keluaran Khas. Dis., 1 – 10.
- Peiyo L., Li W., Guojun S., Lanlan Y., Feng Q. and Liangdong S. (2008). Characterization of high performance exfoliated natural rubber/organoclay nanocomposites. J. Appl. Polym. Sci., 109, 3831.
- Phothiphon K. (2010). Functionalized natural rubber: rubber parts for gasohol resistance. A thesis submitted in partial fulfillment of the requirements for the degree of master of science, PPC, Chulalongkorn University.
- R. D. Simoes, A. E. Job, D. L. Chinaglia, V. Zucolotto, J. C. Camargo – Filho, N. Alves, J. A. Giacometti, O. N. Oliveira Jr and C. J. L. Constantino (2005). Structural characterization of blends containing both PVDF and natural rubber latex. Journal of Raman spectroscopy, 36, 1118 – 1124.
- R. Rajasekar, Kaushik Pal, Gert Heinrich, Amit Das and C.K. Das (2009). Development of nitrile butadiene rubber - nanoclay composites with epoxidized natural rubber as compatibilizer. Materials and design, 30, 3839 – 3845.

- Ray E. Drumright, Patrick R. Gruber and David E. Henton. Polylactic Acid Technology. Advanced Materials.
- Sachiko Ishida, Reiko Nagasaki, Keisuke Chino, Tunggalag Dong and Yoshio Inoue (2009). Toughening of poly(L-lactide) by melt blending with rubbers. Journal of applied polymer science, 113, 558 – 566.
- Scharnowski D. (2005). Characterisation of the influence of cooling rates on structure and properties of dynamic vulcanizates. *Dissertation: Halle (Saale)*, Martin-Luther-Universität Halle-Wittenberg, Mathematisch-Naturwissenschaftlich-Technische Fakultät, 1-134.
- Sen – lin Yang, Zhi – Hua Wu, Wei Yang and Ming – Bo Yang (2008). Thermal and mechanical properties of chemical crosslinked polylactide (PLA). Polymer testing, 27, 957 – 963.
- Snoopy George, N.R. Neelakantan, K.T. Varughese and Sabu Thomas (1997). Dynamic mechanical properties of isotactic polypropylene/nitrile rubber blends: effects of blend ratio, reactive compatibilization and dynamic vulcanization. Journal of polymer science: Part B: Polymer physics, 35, 2309 – 2327.
- Supri and H. Ismail (2006). Effects of dynamic vulcanization and glycidyl methacrylate on properties of recycled poly(vinyl chloride)/acrylonitrile butadiene rubber blends. Polymer Testing, 25, 318- 326.
- Thipmanee Ranumas, Magaraphan Rathanawan and Nampitch Tarinee (2009). ResearchGate, DOI: ISBN 978-974-660362-1.
- Unnikirishnan G. and Thomas S. (1997). Sorption and diffusion of aliphatic hydrocarbons into crosslinked natural rubber. Journal of Polymer Science Part B: Polymer Physics, 35(5), 725-734.
- Yamoun C. and Magaraphan R. Peroxide Cured Natural Rubber/Fluoroelastomer /High Density Polyethylene via Dynamic Vulcanization. Published online in Wiley Online Library (2011) DOI 10.1002/pen.22018.
- Yoksan R. (2008). Epoxidized Natural Rubber for Adhesive Applications. Kasetsart J.(Nat. Sci.), 42, 325 – 332.

Z. Mohamad, H. Ismail and R. Chantara Thevy. Tensile properties and morphology of dynamically vulcanized ENR-50/EVA blends: Different vulcanization system. Journal of chemical and natural resources engineering, 91 - 98.

Zhang G, Zhang J, Wang S and Shen D (2003). J Polym Sci Polym Phys, 41, 23.

APPENDICES

Appendix A: Calculations of Crystallinity Percentage

The calculations of crystallinity percentage were shown below including crystallinity of PVDF, PLA, PVDF in the blend of PVDF-80/PLA-20, PLA in the blend of PVDF-80/PLA-20 and PVDF in thermoplastic vulcanizates at various ENR/PVDF/PLA ratios and DBPH contents. The enthalpy of melting per gram of 100 % crystalline of PLA is 93 J/g and that of PVDF is 104.7 J/g.

$$\text{From } X_c (\%) = \frac{\Delta H^*_{\text{sample}}}{\Delta H^0_{\text{ref}}} \times 100$$

$\Delta H^*_{\text{sample}}$ is the measured enthalpy of sample. ΔH^0_{ref} is the enthalpy of melting per gram of 100 % crystalline of sample.

Percent Crystallinity of PVDF

From the measured result, ΔH^*_{PVDF} is 32.4 J/g and ΔH^0_{PVDF} is 104.7 J/g.

$$\begin{aligned} \text{Percent crystallinity of PVDF} = X_{\text{PVDF}} (\%) &= \frac{\Delta H^*_{\text{PVDF}}}{\Delta H^0_{\text{PVDF}}} \times 100 \\ &= \frac{32.4}{104.7} \times 100 \\ &= 31 \% \end{aligned}$$

Percent Crystallinity of PLA

From the measured result, ΔH^*_{PLA} is 10.9 J/g and ΔH^0_{PLA} is 93 J/g.

$$\begin{aligned} \text{Percent crystallinity of PLA} = X_{\text{PLA}} (\%) &= \frac{\Delta H^*_{\text{PLA}}}{\Delta H^0_{\text{PLA}}} \times 100 \\ &= \frac{10.9}{93} \times 100 \\ &= 12 \% \end{aligned}$$

In case of the thermoplastic blend, the crystallinity percentage of PVDF and PLA was calculated by relating to the composition of PVDF/PLA which was 80/20.

The equations were shown below:

$$X_{\text{Blend (PVDF)}} (\%) = \frac{\Delta H^*_{\text{Blend}}}{(0.8 \times \Delta H^0_{\text{PVDF}})} \times 100$$

$$\text{And } X_{\text{Blend (PLA)}} (\%) = \frac{\Delta H^*_{\text{Blend}}}{(0.2 \times \Delta H^0_{\text{PLA}})} \times 100$$

$X_{\text{Blend (PVDF)}}$ and $X_{\text{Blend (PLA)}}$ are the relative percent crystallinity of the PVDF and PLA in the PVDF/PLA blend. $\Delta H^*_{\text{Blend}}$ is the measured enthalpy. The enthalpy of melting per gram of 100 % crystalline of PLA is 93 J/g and that of PVDF is 104.7 J/g.

Relative Percent Crystallinity of PVDF in The PVDF/PLA Blend

From the measured result, $\Delta H^*_{\text{Blend (PVDF)}}$ is 39.4 J/g and ΔH^0_{PVDF} is 104.7 J/g.

$$\begin{aligned} X_{\text{Blend (PVDF)}} (\%) &= \frac{\Delta H^*_{\text{Blend}}}{(0.8 \times \Delta H^0_{\text{PVDF}})} \times 100 \\ &= \frac{39.4}{(0.8 \times 104.7)} \times 100 \\ &= 29 \% \end{aligned}$$

Relative Percent Crystallinity of PLA in The PVDF/PLA Blend

PLA *peak 1*: From the measured result, $\Delta H^*_{\text{Blend (PLA)}}$ is 0.67 J/g and ΔH^0_{PLA} is 93 J/g.

$$\begin{aligned} X_{\text{Blend (PLA)}} (\%) &= \frac{\Delta H^*_{\text{Blend}}}{(0.2 \times \Delta H^0_{\text{PLA}})} \times 100 \\ &= \frac{0.67}{(0.2 \times 93)} \times 100 \\ &= 3.6 \% \end{aligned}$$

PLA *peak 2*: From the measured result, $\Delta H^*_{\text{Blend (PLA)}}$ is 0.27 J/g and ΔH^0_{PLA} is 93 J/g.

$$X_{\text{Blend (PLA)}} (\%) = \frac{\Delta H^*_{\text{Blend}}}{(0.2 \times \Delta H^0_{\text{PLA}})} \times 100$$

$$X_{\text{Blend (PLA)}} (\%) = \frac{0.27}{(0.2 \times 93)} \times 100$$

$$= 1.4 \%$$

For thermoplastic vulcanizates (TPV), the only crystallization of PVDF was occurred. The relative percent crystallization was calculated to the composition of PVDF in TPV blend. The equations were shown below:

The ENR/PVDF/PLA ratio of 50/40/10;

$$X_{\text{TPV (PVDF)}} (\%) = \frac{\Delta H^*_{\text{TPV}}}{(0.4 \times \Delta H^0_{\text{PVDF}})} \times 100$$

The ENR/PVDF/PLA ratio of 50/50/0;

$$X_{\text{TPV (PVDF)}} (\%) = \frac{\Delta H^*_{\text{TPV}}}{(0.5 \times \Delta H^0_{\text{PVDF}})} \times 100$$

The ENR/PVDF/PLA ratio of 70/20/10;

$$X_{\text{TPV (PVDF)}} (\%) = \frac{\Delta H^*_{\text{TPV}}}{(0.2 \times \Delta H^0_{\text{PVDF}})} \times 100$$

The ENR/PVDF/PLA ratio of 70/30/0;

$$X_{\text{TPV (PVDF)}} (\%) = \frac{\Delta H^*_{\text{TPV}}}{(0.3 \times \Delta H^0_{\text{PVDF}})} \times 100$$

$X_{\text{TPV (PVDF)}}$ is the relative percent crystallinity of the PVDF in the TPV blend. ΔH^*_{TPV} is the measured enthalpy. The enthalpy of melting per gram of 100 % crystalline of PVDF is 104.7 J/g.

Relative Percent Crystallinity of PVDF in The TPV Blend

❖ The ENR/PVDF/PLA Ratio of 50/40/10 at DBPH 0 phr (Ternary blend);

From the measured result, $\Delta H^*_{\text{TPV (PVDF)}}$ is 13.4 J/g and ΔH^0_{PVDF} is 104.7 J/g.

$$X_{\text{TPV (PVDF)}} (\%) = \frac{\Delta H^*_{\text{TPV}}}{(0.4 \times \Delta H^0_{\text{PVDF}})} \times 100$$

$$= \frac{13.4}{(0.4 \times 104.7)} \times 100$$

$$= 32 \%$$

- ❖ The ENR/PVDF/PLA Ratio of 50/40/10 at DBPH 3 phr;

From the measured result, $\Delta H^*_{TPV(PVDF)}$ is 10.7 J/g and ΔH^0_{PVDF} is 104.7 J/g.

$$\begin{aligned} X_{TPV(PVDF)}(\%) &= \frac{\Delta H^*_{TPV}}{(0.4 \times \Delta H^0_{PVDF})} \times 100 \\ &= \frac{10.7}{(0.4 \times 104.7)} \times 100 \\ &= 25.5 \% \end{aligned}$$

- ❖ The ENR/PVDF/PLA Ratio of 50/40/10 at DBPH 7 phr;

From the measured result, $\Delta H^*_{TPV(PVDF)}$ is 6.0 J/g and ΔH^0_{PVDF} is 104.7 J/g.

$$\begin{aligned} X_{TPV(PVDF)}(\%) &= \frac{\Delta H^*_{TPV}}{(0.4 \times \Delta H^0_{PVDF})} \times 100 \\ &= \frac{6.0}{(0.4 \times 104.7)} \times 100 \\ &= 14.4 \% \end{aligned}$$

- ❖ The ENR/PVDF/PLA Ratio of 50/50/0 at DBPH 3 phr;

From the measured result, $\Delta H^*_{TPV(PVDF)}$ is 15.5 J/g and ΔH^0_{PVDF} is 104.7 J/g.

$$\begin{aligned} X_{TPV(PVDF)}(\%) &= \frac{\Delta H^*_{TPV}}{(0.5 \times \Delta H^0_{PVDF})} \times 100 \\ &= \frac{15.5}{(0.5 \times 104.7)} \times 100 \\ &= 29.6 \% \end{aligned}$$

- ❖ The ENR/PVDF/PLA Ratio of 50/50/0 at DBPH 7 phr;

From the measured result, $\Delta H^*_{TPV(PVDF)}$ is 12.9 J/g and ΔH^0_{PVDF} is 104.7 J/g.

$$\begin{aligned} X_{TPV(PVDF)}(\%) &= \frac{\Delta H^*_{TPV}}{(0.5 \times \Delta H^0_{PVDF})} \times 100 \\ &= \frac{12.9}{(0.5 \times 104.7)} \times 100 \\ &= 24.7 \% \end{aligned}$$

- ❖ The ENR/PVDF/PLA Ratio of 70/20/10 at DBPH 3 phr;

From the measured result, $\Delta H^*_{TPV(PVDF)}$ is 5.2 J/g and ΔH^0_{PVDF} is 104.7 J/g.

$$\begin{aligned} X_{TPV(PVDF)}(\%) &= \frac{\Delta H^*_{TPV}}{(0.2 \times \Delta H^0_{PVDF})} \times 100 \\ &= \frac{5.2}{(0.2 \times 104.7)} \times 100 \\ &= 24.6\% \end{aligned}$$

- ❖ The ENR/PVDF/PLA Ratio of 70/30/0 at DBPH 3 phr;

From the measured result, $\Delta H^*_{TPV(PVDF)}$ is 7.6 J/g and ΔH^0_{PVDF} is 104.7 J/g.

$$\begin{aligned} X_{TPV(PVDF)}(\%) &= \frac{\Delta H^*_{TPV}}{(0.3 \times \Delta H^0_{PVDF})} \times 100 \\ &= \frac{7.6}{(0.3 \times 104.7)} \times 100 \\ &= 24\% \end{aligned}$$

Appendix B: Oil Swelling Index

Data of oil swelling index before aging in chapter IV were shown in Tables B1 – B4 and data of oil swelling index after aging were shown in Tables B5 – B8. The results included immersion at room temperature and 100 °C for both of 24 hours and 7 days.

Table B1 The swelling index at room temperature for 24 hours before aging

ENR/PVDF/PLA, % wt	DBPH, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
100/0/0	3	3.02	2.86	2.75	1.29
	5	2.66	2.55	2.35	1.26
	7	2.19	2.08	2.00	1.27
50/40/10	0	2.05	2.40	1.75	1.17
	3	1.43	1.41	1.38	1.12
	5	1.33	1.37	1.34	1.11
	7	1.33	1.32	1.26	1.08
60/30/10	0	2.46	2.49	2.15	1.23
	3	1.64	1.59	1.41	1.17
70/20/10	0	4.20	4.04	2.90	1.30
	3	1.70	1.81	1.84	1.22
50/50/0	3	1.35	1.35	1.49	1.10
	5	1.27	1.28	1.25	1.07
	7	1.28	1.28	1.28	1.06
60/40/0	3	1.49	1.48	1.45	1.13
70/30/0	3	1.65	1.67	1.63	1.19

Table B2 The swelling index at room temperature for 7 days before aging

ENR/PVDF/PLA, % wt	DBPH, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
100/0/0	3	3.10	3.03	2.93	1.32
	5	2.72	2.64	2.50	1.29
	7	2.29	2.26	2.21	1.28
50/40/10	0	2.52	2.56	2.41	1.19
	3	1.48	1.47	1.44	1.12
	5	1.43	1.41	1.36	1.12
	7	1.35	1.36	1.33	1.13
60/30/10	0	2.60	2.66	2.43	1.22
	3	1.66	1.70	1.59	1.18
70/20/10	0	5.28	4.60	4.26	1.31
	3	1.93	1.90	1.80	1.21
50/50/0	3	1.40	1.39	1.37	1.13
	5	1.31	1.30	1.29	1.12
	7	1.29	1.32	1.30	1.12
60/40/0	3	1.51	1.53	1.47	1.18
70/30/0	3	1.68	1.68	1.66	1.21

Table B3 The swelling index at room temperature for 24 hours after aging

ENR/PVDF/PLA, % wt	DBPH, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
100/0/0	3	3.16	3.02	2.71	1.28
	5	2.68	2.56	2.35	1.29
	7	2.31	2.23	2.11	1.27
50/40/10	0	2.70	2.43	2.16	1.22
	3	1.46	1.48	1.42	1.10
	5	1.44	1.41	1.39	1.10
	7	1.34	1.34	1.33	1.08
60/30/10	0	2.89	2.44	2.16	1.28
	3	1.68	1.67	1.61	1.18
70/20/10	0	4.97	4.73	3.20	1.32
	3	1.92	1.96	1.83	1.21
50/50/0	3	1.37	1.38	1.35	1.07
	5	1.26	1.24	1.22	1.04
	7	1.27	1.28	1.27	1.06
60/40/0	3	1.48	1.49	1.45	1.13
70/30/0	3	1.68	1.68	1.63	1.19

Table B4 The swelling index at room temperature for 7 days after aging

ENR/PVDF/PLA, % wt	DBPH, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
100/0/0	3	3.20	3.16	3.00	1.33
	5	2.69	2.60	2.51	1.30
	7	2.36	2.31	2.24	1.30
50/40/10	0	2.91	2.81	2.44	1.25
	3	1.46	1.47	1.42	1.13
	5	1.45	1.42	1.40	1.14
	7	1.34	1.34	1.33	1.08
60/30/10	0	3.08	2.92	2.47	1.31
	3	1.68	1.67	1.61	1.18
70/20/10	0	5.41	5.20	4.53	1.33
	3	1.92	1.96	1.83	1.21
50/50/0	3	1.38	1.39	1.36	1.12
	5	1.31	1.29	1.27	1.10
	7	1.30	1.30	1.29	1.12
60/40/0	3	1.51	1.52	1.45	1.17
70/30/0	3	1.71	1.71	1.65	1.21

Table B5 The swelling index at 100 °C for 24 hours before aging

ENR/PVDF/PLA, % wt	DBPH, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
100/0/0	3	2.84	2.78	2.82	1.55
	5	2.57	2.50	2.44	1.50
	7	2.12	2.02	1.92	1.40
50/40/10	0	-	-	-	-
	3	1.60	1.50	1.69	1.23
	5	1.50	1.47	1.47	1.21
	7	1.43	1.41	1.39	1.21
60/30/10	0	-	-	-	-
	3	1.90	1.84	1.81	1.25
70/20/10	0	-	-	-	-
	3	2.22	2.18	2.46	1.26
50/50/0	3	1.53	1.47	1.64	1.30
	5	1.44	1.37	1.38	1.24
	7	1.41	1.38	1.34	1.24
60/40/0	3	1.71	1.68	1.65	1.35
70/30/0	3	1.98	1.86	1.96	1.33

Table B6 The swelling index at 100 °C for 7 days before aging

ENR/PVDF/PLA, % wt	DBPH, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
100/0/0	3	2.84	2.89	2.86	1.55
	5	2.58	2.55	2.50	1.51
	7	2.12	2.08	1.96	1.39
50/40/10	0	-	-	-	-
	3	1.63	1.55	1.73	1.37
	5	1.53	1.52	1.53	1.21
	7	1.42	1.39	1.39	1.19
60/30/10	0	-	-	-	-
	3	1.90	1.92	1.87	1.26
70/20/10	0	-	-	-	-
	3	2.43	2.31	2.48	1.33
50/50/0	3	1.57	1.54	1.68	1.46
	5	1.47	1.41	1.42	1.24
	7	1.41	1.39	1.34	1.23
60/40/0	3	1.73	1.87	1.68	1.34
70/30/0	3	2.13	1.97	1.97	1.41

Table B7 The swelling index at 100 °C for 24 hours after aging

ENR/PVDF/PLA, % wt	DBPH, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
100/0/0	3	2.90	2.81	2.87	1.56
	5	2.56	2.48	2.53	1.53
	7	2.10	2.03	2.01	1.45
50/40/10	0	-	-	-	-
	3	1.54	1.58	1.71	1.30
	5	1.45	1.52	1.42	1.19
	7	1.41	1.42	1.38	1.16
60/30/10	0	-	-	-	-
	3	1.97	1.89	1.85	1.25
70/20/10	0	-	-	-	-
	3	2.48	2.40	2.47	1.32
50/50/0	3	1.50	1.47	1.65	1.29
	5	1.41	1.39	1.34	1.20
	7	1.37	1.40	1.36	1.18
60/40/0	3	1.73	1.70	1.59	1.25
70/30/0	3	1.96	1.95	2.00	1.34

Table B8 The swelling index at 100 °C for 7 days after aging

ENR/PVDF/PLA, % wt	DBPH, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
100/0/0	3	2.89	2.87	2.87	1.56
	5	2.52	2.51	2.55	1.54
	7	2.11	2.08	2.02	1.45
50/40/10	0	-	-	-	-
	3	1.62	1.61	1.75	1.45
	5	1.49	1.55	1.51	1.18
	7	1.40	1.43	1.36	1.15
60/30/10	0	-	-	-	-
	3	1.95	1.95	1.87	1.23
70/20/10	0	-	-	-	-
	3	2.72	2.63	2.47	1.39
50/50/0	3	1.53	1.52	1.68	1.52
	5	1.43	1.42	1.41	1.20
	7	1.37	1.42	1.37	1.19
60/40/0	3	1.73	1.76	1.62	1.26
70/30/0	3	2.13	2.05	2.01	1.42

Data of oil swelling index before aging in chapter V were shown in Tables B9 – B12 and data of oil swelling index after aging were shown in Tables B13 – B16. The results included immersion at room temperature and 100 °C for both of 24 hours and 7 days.

Table B9 The swelling index of organoclay-filled TPV at room temperature for 24 hours before aging

ENR/PVDF/ PLA, % wt	Organoclay, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
50/40/10	0	1.33	1.37	1.34	1.11
	3	1.37	1.36	1.34	1.07
	5	1.37	1.36	1.33	1.08
	7	1.34	1.34	1.31	1.08
	10	1.35	1.33	1.31	1.07
50/50/0	0	1.27	1.28	1.25	1.07
	3	1.30	1.28	1.30	1.04
	5	1.32	1.32	1.31	1.09
	7	1.31	1.31	1.28	1.10
	10	1.35	1.34	1.31	1.11

Table B10 The swelling index of organoclay-filled TPV at room temperature for 7 days before aging

ENR/PVDF/ PLA, % wt	Organoclay, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
50/40/10	0	1.43	1.41	1.36	1.12
	3	1.37	1.36	1.35	1.10
	5	1.37	1.37	1.33	1.11
	7	1.35	1.34	1.32	1.11
	10	1.35	1.34	1.31	1.11
50/50/0	0	1.31	1.30	1.29	1.12
	3	1.31	1.30	1.31	1.08
	5	1.33	1.34	1.33	1.14
	7	1.33	1.32	1.29	1.12
	10	1.36	1.36	1.31	1.12

Table B11 The swelling index of organoclay-filled TPV at room temperature for 24 hours after aging

ENR/PVDF/ PLA, % wt	Organoclay, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
50/40/10	0	1.44	1.41	1.39	1.10
	3	1.38	1.36	1.33	1.06
	5	1.37	1.38	1.34	1.08
	7	1.38	1.37	1.33	1.07
	10	1.35	1.35	1.34	1.06
50/50/0	0	1.26	1.24	1.22	1.04
	3	1.33	1.31	1.31	1.08
	5	1.31	1.32	1.29	1.09
	7	1.31	1.33	1.28	1.08
	10	1.36	1.35	1.32	1.11

Table B12 The swelling index of organoclay-filled TPV at room temperature for 7 days after aging

ENR/PVDF/ PLA, % wt	Organoclay, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
50/40/10	0	1.45	1.42	1.40	1.14
	3	1.38	1.37	1.34	1.11
	5	1.39	1.37	1.35	1.10
	7	1.39	1.37	1.35	1.11
	10	1.37	1.36	1.35	1.10
50/50/0	0	1.31	1.29	1.27	1.10
	3	1.34	1.34	1.34	1.13
	5	1.32	1.33	1.30	1.13
	7	1.33	1.34	1.29	1.12
	10	1.37	1.36	1.33	1.13

Table B13 The swelling index of organoclay-filled TPV at 100 °C for 24 hours before aging

ENR/PVDF/ PLA, % wt	Organoclay, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
50/40/10	0	1.50	1.47	1.47	1.21
	3	1.46	1.47	1.43	1.18
	5	1.47	1.47	1.44	1.16
	7	1.50	1.47	1.39	1.18
	10	1.46	1.47	1.41	1.18
50/50/0	0	1.44	1.37	1.38	1.24
	3	1.43	1.41	1.51	1.19
	5	1.45	1.44	1.51	1.19
	7	1.42	1.43	1.49	1.19
	10	1.42	1.45	1.50	1.19

Table B14 The swelling index of organoclay-filled TPV at 100 °C for 7 days before aging

ENR/PVDF/ PLA, % wt	Organoclay, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
50/40/10	0	1.53	1.52	1.53	1.21
	3	1.47	1.51	1.52	1.17
	5	1.48	1.51	1.43	1.16
	7	1.48	1.50	1.41	1.16
	10	1.50	1.50	1.51	1.21
50/50/0	0	1.47	1.41	1.42	1.24
	3	1.44	1.42	1.54	1.20
	5	1.46	1.45	1.55	1.21
	7	1.43	1.44	1.53	1.19
	10	1.43	1.46	1.55	1.20

Table B15 The swelling index of organoclay-filled TPV at 100 °C for 24 hours after aging

ENR/PVDF/ PLA, % wt	Organoclay, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
50/40/10	0	1.45	1.52	1.42	1.19
	3	1.45	1.47	1.42	1.18
	5	1.48	1.47	1.42	1.17
	7	1.48	1.47	1.41	1.18
	10	1.49	1.45	1.42	1.18
50/50/0	0	1.41	1.39	1.34	1.20
	3	1.42	1.46	1.51	1.19
	5	1.44	1.43	1.52	1.20
	7	1.40	1.39	1.49	1.18
	10	1.44	1.44	1.53	1.20

Table B16 The swelling index of organoclay-filled TPV at 100 °C for 7 days after aging

ENR/PVDF/ PLA, % wt	Organoclay, phr	Swelling index			
		Gasohol 91	Gasohol 95	E20	E85
50/40/10	0	1.49	1.55	1.51	1.18
	3	1.48	1.51	1.49	1.16
	5	1.48	1.51	1.43	1.17
	7	1.51	1.51	1.44	1.16
	10	1.51	1.50	1.51	1.21
50/50/0	0	1.43	1.42	1.41	1.20
	3	1.43	1.47	1.54	1.20
	5	1.45	1.44	1.55	1.21
	7	1.40	1.41	1.53	1.19
	10	1.43	1.46	1.55	1.20

Data of oil swelling index before and after aging in chapter VI were shown in Tables B17 – B20. The results included immersion at room temperature and 100 °C for both of 24 hours and 7 days.

Table B17 The swelling index of TPV both of NBR and ENR system at room temperature for 24 hours

NBR/PVDF/ PLA, % wt	Aging		Swelling index			
	Before	After	Gasohol 91	Gasohol 95	E20	E85
50/40/10	✓		1.17	1.15	1.15	1.06
		✓	1.15	1.15	1.14	1.06
50/50/0	✓		1.20	1.17	1.15	1.06
		✓	1.19	1.16	1.15	1.05
ENR/PVDF/ PLA, % wt	Aging		Swelling index			
	Before	After	Gasohol 91	Gasohol 95	E20	E85
50/40/10	✓		1.33	1.37	1.34	1.11
		✓	1.44	1.41	1.39	1.10
50/50/0	✓		1.27	1.28	1.25	1.07
		✓	1.26	1.24	1.22	1.04

Table B18 The swelling index of TPV both of NBR and ENR system at room temperature for 7 days

NBR/PVDF/ PLA, % wt	Aging		Swelling index			
	Before	After	Gasohol 91	Gasohol 95	E20	E85
50/40/10	✓		1.23	1.22	1.22	1.11
		✓	1.22	1.21	1.20	1.12
50/50/0	✓		1.24	1.22	1.22	1.11
		✓	1.24	1.22	1.21	1.11
ENR/PVDF/ PLA, % wt	Aging		Swelling index			
	Before	After	Gasohol 91	Gasohol 95	E20	E85
50/40/10	✓		1.43	1.41	1.36	1.12
		✓	1.45	1.42	1.40	1.14
50/50/0	✓		1.31	1.30	1.29	1.12
		✓	1.31	1.29	1.27	1.10

Table B19 The swelling index of TPV both of NBR and ENR system at 100 °C for 24 hours

NBR/PVDF/ PLA, % wt	Aging		Swelling index			
	Before	After	Gasohol 91	Gasohol 95	E20	E85
50/40/10	✓		1.25	1.22	1.34	1.24
		✓	1.23	1.21	1.35	1.25
50/50/0	✓		1.28	1.27	1.28	1.24
		✓	1.27	1.25	1.28	1.22
ENR/PVDF/ PLA, % wt	Aging		Swelling index			
	Before	After	Gasohol 91	Gasohol 95	E20	E85
50/40/10	✓		1.50	1.47	1.47	1.21
		✓	1.45	1.52	1.42	1.19
50/50/0	✓		1.44	1.37	1.38	1.24
		✓	1.41	1.39	1.34	1.20

Table B20 The swelling index of TPV both of NBR and ENR system at 100 °C for 7 days

NBR/PVDF/ PLA, % wt	Aging		Swelling index			
	Before	After	Gasohol 91	Gasohol 95	E20	E85
50/40/10	✓		1.23	1.22	1.34	1.35
		✓	1.21	1.21	1.33	1.35
50/50/0	✓		1.28	1.27	1.31	1.25
		✓	1.27	1.26	1.30	1.23
ENR/PVDF/ PLA, % wt	Aging		Swelling index			
	Before	After	Gasohol 91	Gasohol 95	E20	E85
50/40/10	✓		1.53	1.52	1.53	1.21
		✓	1.49	1.55	1.51	1.18
50/50/0	✓		1.47	1.41	1.42	1.24
		✓	1.43	1.42	1.41	1.20

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